



VIT[®]
Vellore Institute of Technology
(Deemed to be University under section 3 of UGC Act, 1956)

SCHOOL OF CHEMICAL ENGINEERING (SCHEME)



B.Tech Chemical Engineering (BCM)

Curriculum and Syllabus

[2019-2020 admitted students]



VISION STATEMENT OF VELLORE INSTITUTE OF TECHNOLOGY

Transforming life through excellence in education and research

MISSION STATEMENT OF VELLORE INSTITUTE OF TECHNOLOGY

World class Education: Excellence in education, grounded in ethics and critical thinking, for improvement of life.

Cutting edge Research: An innovation ecosystem to extend knowledge and solve critical problems.

Impactful People: Happy, accountable, caring and effective workforce and students.

Rewarding Co-creations: Active collaboration with national & international industries & universities for productivity and economic development.

Service to Society: Service to the region and world through knowledge and compassion.

VISION STATEMENT OF THE SCHOOL OF CHEMICAL ENGINEERING

- To improve the quality of life through innovations in Chemical Engineering

MISSION STATEMENT OF THE SCHOOL OF CHEMICAL ENGINEERING

- To prepare the graduates for a rewarding career by providing quality education in Chemical Engineering in tune with evolving requirements of the society.
- To impart knowledge and develop technology through quality research in frontier areas of chemical and inter-disciplinary fields.
- To produce practicing engineers with professional ethics to cater the contemporary needs of the society and environment.

B. Tech Chemical Engineering

PROGRAMME EDUCATIONAL OBJECTIVES (PEOs)

1. Graduates will be engineering practitioners and leaders, who would help solve industry's technological problems in Chemical engineering and allied disciplines.
2. Graduates will be engineering professionals, innovators or entrepreneurs engaged in technology development, technology deployment, or engineering system implementation in the industry.
3. Graduates will function in their profession with social awareness and responsibility.
4. Graduates will interact with their peers in other disciplines in industry and society and contribute to the economic growth of the country.
5. Graduates will be successful in pursuing higher studies leading to careers in engineering, management, teaching and research.

PROGRAMME OUTCOMES (POs)

1. Engineering Knowledge :Apply the knowledge of mathematics, science, engineering fundamentals, and an engineering specialisation to the solution of complex engineering problems.
2. Problem analysis :Identify, formulate, research literature, and analyse complex engineering problems reaching substantiated conclusions using first principles of mathematics, natural sciences, and engineering sciences.
3. Design/development of solutions: Design solutions for complex engineering problems and design system components or processes that meet the specified needs with appropriate consideration for the public health and safety, and the cultural, societal, and environmental considerations.
4. Conduct investigations of complex problems: Use research-based knowledge and research methods including design of experiments, analysis and interpretation of data, and synthesis of the information to provide valid conclusions.
5. Modern Tool Usage :Create, select, and apply appropriate techniques, resources, and modern engineering and IT tools including prediction and modelling to complex engineering activities with an understanding of the limitations.
6. The Engineer and Society: Apply reasoning informed by the contextual knowledge to assess societal, health, safety, legal, and cultural issues and the

consequent responsibilities relevant to the professional engineering practice.

7. **Environment and Sustainability:** Understand the impact of the professional engineering solutions in societal and environmental contexts, and demonstrate the knowledge of need for sustainable development.
8. **Ethics:** Apply ethical principles and commit to professional ethics and responsibilities and norms of the engineering practice.
9. **Individual and Team Work:** Function effectively as an individual, and as a member or leader in diverse teams, and in multidisciplinary settings.
10. **Communication:** Communicate effectively on complex engineering activities with the engineering community and with society at large, such as, being able to comprehend and write effective reports and design documentation, make effective presentations, and give and receive clear instructions.
11. **Project Management and Finance:** Demonstrate knowledge and understanding of the engineering and management principles and apply these to one's own work, as a member and leader in a team, to manage projects and in multidisciplinary environments.
12. **Life-long learning :** Recognise the need for, and have the preparation and ability to engage in independent and life-long learning in the broadest context of technological change

B. Tech Chemical Engineering

PROGRAMME SPECIFIC OUTCOMES (PSOs)

On completion of B. Tech. (Chemical Engineering) programme, graduates will be able to:

1. Analyze and solve complex engineering problems in process and allied Industries by applying core and multidisciplinary competencies.
2. Design and develop efficient chemical processes/products considering economic, safety and environmental aspects.
3. Implement the modern practices in industrial/research settings to serve as practicing engineers with professional ethics.



SCHOOL OF CHEMICAL ENGINEERING (SCHEME)

B. Tech Chemical Engineering

CREDIT STRUCTURE

Category-wise Credit distribution

S.No	Description	Credits	Maximum Credits
1	PC - Programme Core	60	60
2	PE - Programme Elective	35	35
3	UC-University Core	53	53
4	UE - University Elective	12	12
5	BC- Bridge Course	0	0
6	NC- Non Credit Course	5	5
	Total credits	165	165

B. Tech Chemical Engineering

Programme Core

Sl. No	Course Code	Course Title	L	T	P	J	C
1	CHE1001	Materials Science and Strength of Materials	3	0	0	0	3
2	CHE1002	Process Calculations	4	0	0	0	4
3	CHE1003	Process Engineering Thermodynamics	3	0	0	4	4
4	CHE1004	Chemical Technology	3	0	0	0	3
5	CHE1005	Momentum Transfer	3	0	2	0	4
6	CHE1006	Heat Transfer	2	0	2	4	4
7	CHE1022	Mechanical Operations	3	0	2	0	4
8	CHE2001	Chemical Reaction Engineering	3	0	2	0	4
9	CHE2002	Process Equipment Design and Economics	2	0	2	4	4
10	CHE3001	Computational Methods in Process Engineering	3	0	2	0	4
11	CHE3002	Process Instrumentation and Control	2	0	2	4	4
12	CHE3003	Mass Transfer	3	0	0	0	3
13	CHE4001	Equilibrium Staged Operations	2	0	2	4	4
14	MAT2002	Applications of Differential and Difference Equations	3	0	2	0	4
15	MAT3003	Complex Variables and Partial Differential Equations	3	1	0	0	4
16	MEE1001	Engineering Drawing	1	0	4	0	3
Total							60

B. Tech Chemical Engineering

Programme Electives

Sl. No.	Course Code	Course Title	L	T	P	J	C
1	CHE1007	Safety and Hazard Analysis	2	0	0	4	3
2	CHE1008	Unit Processes in Organic Synthesis	3	0	2	0	4
3	CHE1009	Biochemical Engineering	3	0	0	0	3
4	CHE1010	Process Plant Utilities	3	0	0	0	3
5	CHE1011	Optimization of Chemical Processes	3	0	0	0	3
6	CHE1013	Natural Gas Engineering	3	0	0	0	3
7	CHE1014	Petroleum Technology	3	0	0	0	3
8	CHE1015	Petrochemical Technology	3	0	0	0	3
9	CHE1016	Fermentation Technology	3	0	0	0	3
10	CHE1017	Food Process Engineering	2	0	0	4	3
11	CHE1018	Membrane Separations Technology	3	0	0	0	3
12	CHE1019	Polymer Technology	3	0	0	0	3
13	CHE1020	Fertilizer Technology	3	0	0	0	3
14	CHE1023	Production and Operations Management	3	0	0	0	3
15	CHE2003	Chemical Product Design	3	0	0	0	3
16	CHE2006	Fuels and Combustion	3	0	0	0	3
17	CHE2007	Process Intensification	3	0	0	0	3
18	CHE2008	Chemical Engineering Computational Fluid Dynamics	2	0	0	4	3
19	CHE3004	Heterogeneous Reaction Engineering	2	0	0	4	3
20	CHE3005	Chemical Process Integration	3	0	0	0	3
21	CHE3006	Process Plant Simulation	3	0	0	4	4

22	CHE3007	Multiphase Flow	3	0	0	0	3
23	CHE3008	Industrial Pollution Engineering	3	0	0	0	3
24	CHE3010	Colloids and Interfacial Sciences	3	0	0	0	3
25	CHE4002	Transport Phenomena	3	0	0	0	3
26	CHE4003	Modelling and Simulation in Process Engineering	2	0	2	0	3
27	CHE4005	Fluidization Engineering	3	0	0	0	3
28	CHE4006	Introduction to Molecular Dynamics and Simulation	3	0	0	0	3
29	CHE4007	Rheology of Complex Fluids	3	0	0	0	3
30	CHY1004	Materials and Instrumental Techniques	3	0	2	0	4
31	EEE1001	Basic Electrical and Electronics Engineering	2	0	2	0	3
32	MEE1011	Renewable Energy Sources	2	2	2	0	4
33	MEE4006	Computational Fluid Dynamics	2	2	2	0	4

B. Tech Chemical Engineering University Core

S.No	Course Code	Course Title	L	T	P	J	C
1	CHE1901	Technical Answers for Real World Problems (TARP)	1	0	0	4	2
2	CHE1902	Industrial Internship	0	0	0	0	1
3	CHE1903	Comprehensive Examination	0	0	0	0	1
4	CHE1904	Capstone Project	0	0	0	0	12
5	CHY1701	Engineering Chemistry	3	0	2	0	4
6	CSE1001	Problem Solving and Programming	0	0	6	0	3
7	CSE1002	Problem Solving and Object Oriented Programming	0	0	6	0	3
8	ENG1901/	Technical English I	0/	0/	4/	0/	2
	ENG1902/	Technical English II	0/	0/	4/	0/	
	ENG1903	Advanced Technical English	0	0	2	4	
9	FLC4097	Foreign Language Courses Basket	2	0	0	0	2
10	HUM1021	Ethics and Values	2	0	0	0	2
11	MAT1011	Calculus for Engineers	3	0	2	0	4
12	MAT2001	Statistics for Engineers	3	0	2	0	4
13	MGT1022	Lean Start-up Management	1	0	0	4	2
14	PHY1701	Engineering Physics	3	0	2	0	4
15	PHY1901	Introduction to Innovative Projects	1	0	0	0	1
16	STS 1101	Fundamentals of Aptitude	0	0	0	0	1
	STS 1201	Introduction to problem solving					
17	STS 1102	Arithmetic problem solving	0	0	0	0	1
	STS 1202	Introduction to quantitative, logical and verbal ability					
18	STS 2101	Getting started to skill enhancement	0	0	0	0	1

	STS 2201	Numerical ability and cognitive intelligence					
19	STS 2102	Enhancing problem solving skills	0	0	0	0	1
	STS 2202	Advanced aptitude and reasoning skills					
20	STS 3101	Introduction to programming skills	0	0	0	0	1
	STS 3201	Programming skills for employment					
BRIDGE COURSE – NON CREDIT COURSE							
1	CHY1002	Environmental Sciences	3	0	0	0	3
2	ENG1000/ ENG2000	Foundation English I (BC) Foundation English II (BC)	0	0	4	0	2
3	EXC4097	Extra & Co- Curricular Activities	0	0	0	0	2
	Total Credits (A)		60				
	Non Credit Course (B)		7				
	University Core Courses (A-B)		53				

B. Tech Chemical Engineering

University Elective-12 Credits

UNIVERSITYCORE

Course code	Technical Answers for Real World Problems (TARP)	L	T	P	J	C
CHE1901		1	0	0	4	2
Pre-requisite	PHY1999 and 115 Credits Earned	Syllabus version				
		v. 1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. To help students to identify the need for developing newer technologies for industrial / societal Needs 2. To train students to propose and implement relevant technology for the development of the prototypes / products 3. To make the students learn to use the methodologies available to assess the developed prototypes / products 						
Course Outcomes:						
<ol style="list-style-type: none"> 1. Identify real life problems related to society. 2. Apply appropriate technology(ies) to address the identified problems using engineering principles and arrive at innovative solutions 						
Module:1		15 hours				
<ol style="list-style-type: none"> 1. Identification of real life problems 2. Field visits can be arranged by the faculty concerned 3. 6 – 10 students can form a team (within the same/different discipline) 4. Minimum of eight hours on self-managed team activity 5. Appropriate scientific methodologies to be utilized to solve the identified issue 6. Solution should be in the form of fabrication/coding/modeling/product design/process design/relevant scientific methodology(ies) 7. Consolidated report to be submitted for assessment 8. Participation, involvement and contribution in group discussions during the contact hours will be used as the modalities for the continuous assessment of the theory component 9. Project outcome to be evaluated in terms of technical, economical, social, environmental, political and demographic feasibility 10. Contribution of each group member to be assessed 11. The project component to have three reviews with the weightage of 20:30:50 						
Mode of Evaluation: (No FAT) Continuous Assessment the project done – Mark weightage of 20:30:50 – project report to be submitted, presentation and project reviews						
Recommended by Board of Studies	04-03-2016					
Approved by Academic Council	No. 47	Date	05-10-2017			

CHE1902	Industrial Internship			L	T	P	J	C
				0	0	0	0	1
Pre-requisite	Completion of minimum of Two semesters							
Course Objectives:								
The course is designed so as to expose the students to industry environment and to take up on-site assignment as trainees or interns.								
Course Outcomes:								
<ol style="list-style-type: none"> 1. Have an exposure to industrial practices and to work in teams 2. Communicate effectively 3. Understand the impact of engineering solutions in a global, economic, environmental and societal context 4. Develop the ability to engage in research and to involve in life-long learning 5. Comprehend contemporary issues 6. Engage in establishing his/her digital footprint 								
Contents				4	Weeks			
Four weeks of work at industry site. Supervised by an expert at the industry.								
Mode of Evaluation: Internship Report, Presentation and Project Review								
Recommended by Board of Studies				28-02-2016				
Approved by Academic Council				No. 37	Date	16-06-2015		

Course code	Comprehensive Examination				
CHE1903					
Pre-requisite	Minimum of 15 Credits Earned or at the end of the 7 th semester			Syllabus version	
					v. 1.0
Course Objectives:					
<ol style="list-style-type: none"> To measure student's competency and mastery of concepts in the field of chemical engineering. To evaluate the ability of students to move into the dissertation phase of their degree. 					
Course Outcomes:					
<ol style="list-style-type: none"> Define, explain and summarize the basic principles of chemical engineering. Use the principles of science and mathematics to identify, formulate and solve advanced engineering problems. Evaluate the hypotheses, methods, results and conclusions of published scientific literature and apply conclusions to their own work. 					
Contents					
<p>Process Calculations and Thermodynamics: Steady and unsteady state mass and energy balances including multiphase, multi-component, reacting and non-reacting systems. Use of tie components; recycle, bypass and purge calculations; Gibb's phase rule and degree of freedom analysis. First and Second laws of thermodynamics. Applications of first law to closed and open systems. Second law and Entropy. Thermodynamic properties of pure substances: Equation of State and residual properties, properties of mixtures: partial molar properties, fugacity, excess properties and activity coefficients; phase equilibria: predicting VLE of systems; chemical reaction equilibrium</p> <p>Momentum Transfer: Fluid statics, Newtonian and non-Newtonian fluids, shell-balance including differential form of Bernoulli equation and energy balance, Macroscopic friction factors, dimensional analysis, flow through pipeline systems, flowmeters, pumps and compressors, flow past immersed bodies including packed and fluidized beds, Turbulent flow: fluctuating velocity, universal velocity profile and pressure drop.</p> <p>Mechanical Operations: Particle size and shape, particle size distribution, size reduction and classification of solid particles; free and hindered settling; centrifuge and cyclones; thickening and classification, flotation, filtration, agitation and mixing; conveying of solids.</p> <p>Heat Transfer: Steady and unsteady heat conduction, convection and radiation, Heat Transfer through fins, thermal boundary layer and heat transfer coefficients, boiling, condensation and evaporation; types of heat exchangers and evaporators and their process calculations. Design of double pipe, shell and tube heat exchangers, and single and multiple effect evaporators.</p>					

Mass Transfer: Fick's laws, molecular diffusion in fluids, mass transfer coefficients, film, penetration and surface renewal theories; analogies; stage-wise and continuous contacting and stage efficiencies; HTU & NTU concepts; design and operation of equipment for distillation, absorption, leaching, liquid-liquid extraction, drying, humidification, dehumidification adsorption and crystallization.

Chemical Reaction Engineering: Theories of reaction rates; kinetics of homogeneous reactions, interpretation of kinetic data, single and multiple reactions in ideal reactors, non-ideal reactors; residence time distribution, single parameter model; non-isothermal reactors; kinetics of heterogeneous catalytic reactions; diffusion effects in catalysis.

Instrumentation and Process Control: Measurement of process variables; sensors, transducers and their dynamics, process modeling and linearization, transfer functions and dynamic responses of various systems, systems with inverse response, process reaction curve, controller modes (P, PI, and PID); control valves; analysis of closed loop systems including stability, frequency response, controller tuning, cascade and feed forward control.

Chemical Technology: Inorganic chemical industries (sulfuric acid, phosphoric acid, chloro-alkali industry, cement, paint, glass industry), fertilizers (Ammonia, Urea, SSP and TSP); natural products industries (Pulp and Paper, Sugar, Oil, and Fats); petroleum refining and petrochemicals; Fermentation products: Ethanol, citric acid, antibiotics, penicillin polymerization industries (polyethylene, polypropylene, PVC and polyesters synthetic fibers).

Plant Design and Economics: Principles of process economics and cost estimation including depreciation and total annualized cost, cost indices, rate of return, payback period, discounted cash flow, optimization in process design and sizing of chemical engineering equipments such as compressors, heat exchangers, multistage contactors.

Assessment / Criteria: Computer based on line Examination

Mode of Evaluation: Computer based Evaluation

Recommended by Board of Studies	04-03-2016		
Approved by Academic Council	No. 47	Date	05.10.2017

CourseCode	CourseTitle	L	T	P	J	C
CHE1904	CapstoneProject	0	0	0	0	12
Pre-requisite	Aspertheacademicregulations	Syllabus version				
		v. 1.0				
CourseObjectives:						
To providesufficienthands-on learningexperiencerelatedtothedesign,developmentandanalysisofsuitableproduct/processsoastoe						
CourseOutcomes:						
<ol style="list-style-type: none"> 1. Formulatespecificproblemstatementsforill-definedreallifeproblemswithreasonableassumptionsandconstraints. 2. Performliteratureresearchand/orpatentsearchintheareaofinterest. 3. Conductexperiments/DesignandAnalysis/solutioniterationsanddocumenttheresults. 4. Performerroranalysis/benchmarking/costing 5. Synthesizetheresultsandarriveatscientificconclusions/products/solution 6. Documenttheresultsintheformoftechnicalreport/presentation 						
Mode ofEvaluation:Periodicreviews,Presentation,Finaloralviva,Postersubmission						
Recommended byBoardofStudies	10-06-2015					
Approved byAcademicCouncil	37 th AC	Date	16-06-2015			

Coursecode	EngineeringChemistry	L	T	P	J	C
CHY1701		3	0	2	0	4
Pre-requisite	Chemistry of 12 th standard or equivalent	Syllabus version				
		1.1				
CourseObjectives:						
1. To impart technological aspects of applied chemistry 2. To lay foundation for practical application of chemistry in engineering aspects						
CourseOutcomes(CO):						
1. Recall and analyze the issues related to impurities in water and their removal methods and apply recent methodologies in water treatment for domestic and industrial usage 2. Evaluate the causes of metallic corrosion and apply the methods for corrosion protection of metals 3. Evaluate the electrochemical energy storage systems such as lithium batteries, fuel cells and solar cells, and design for usage in electrical and electronic applications 4. Assess the quality of different fossil fuels and create an awareness to develop the alternative fuels 5. Analyze the properties of different polymers and distinguish the polymers which can be degraded and demonstrate their usefulness 6. Apply the theoretical aspects: (a) in assessing the water quality; (b) understanding the construction and working of electrochemical cells; (c) analyzing metals, alloys and soil using instrumental methods; (d) evaluating the viscosity and water absorbing properties of polymeric materials						
Module:1	Water Technology	5 hours				
Characteristics of hard water-hardness, DO, TDS in water and their determination – numerical problems in hardness determination by EDTA; Modern techniques of water analysis for industrial use- Disadvantages of hard water in industries.						
Module:2	Water Treatment	8 hours				
Water softening methods:- Lime-soda, Zeolite and ion exchange processes and their applications. Specifications of water for domestic use (ICMR and WHO); Unit processes involved in water treatment for municipal supply- Sedimentation with coagulant- Sand Filtration - chlorination; Domestic water purification – Candle filtration - activated carbon filtration; Disinfection methods- Ultrafiltration, UV treatment, Ozonolysis, Reverse Osmosis; Electrodialysis.						
Module:3	Corrosion	6 hours				
Dry and wet corrosion- detrimental effects to buildings, machines, devices & decorative art forms, emphasizing Differential aeration, Pitting, Galvanic and Stress corrosion cracking; Factors that enhance corrosion and choice of parameters to mitigate corrosion.						
Module:4	Corrosion Control	4 hours				
Corrosion protection-cathodic protection – sacrificial anodic and impressed current protection methods; Advanced protective coatings: electroplating and electroless plating, PVD and CVD. Alloying for corrosion protection – Basic concepts of Eutectic composition and Eutectic mixtures- Selected examples – Ferrous and non-ferrous alloys.						
Module:5	Electrochemical Energy Systems	6 hours				

Brief introduction to conventional primary and secondary batteries; High energy electrochemical energy systems: Lithium batteries – Primary and secondary, its Chemistry, advantages and applications. Fuel cells – Polymer membrane fuel cells, Solid-oxide fuel cells – working principles, advantages, applications. Solar cells – Types – Importance of silicon single crystal, polycrystalline and amorphous silicon solar cells, dye sensitized solar cells – working principles, characteristics and applications.		
Module:6	Fuels and Combustion	8 hours
Calorific value – Definition of LCV, HCV. Measurement of calorific value using bomb calorimeter and Boy's calorimeter including numerical problems. Controlled combustion of fuels – Air fuel ratio – minimum quantity of air by volume and by weight – Numerical problems – three way catalytic converter – selective catalytic reduction of NO _x ; Knocking in IC Engines – Octane and Cetane number – Antiknocking agents.		
Module:7	Polymers	6 hours
Difference between thermoplastics and thermosetting plastics; Engineering application of plastics – ABS, PVC, PTFE and Bakelite; Compounding of plastics: moulding of plastics for Car parts, bottle caps (Injection moulding), Pipes, Hoses (Extrusion moulding), Mobile Phone Cases, Battery Trays, (Compression moulding), Fibre reinforced polymers, Composites (Transfer moulding), PET bottles (blow moulding); Conducting polymers – Polyacetylene – Mechanism of conduction – applications (polymers in sensors, self-cleaning windows)		
Module:8	Contemporary issues:	2 hours
Lecture by Industry Experts		
Total Lecture hours:		45 hours
Text Book(s)		
<ol style="list-style-type: none"> 1. Sashi Chawla, A Textbook of Engineering Chemistry, Dhanpat Rai Publishing Co., Pvt. Ltd., Educational and Technical Publishers, New Delhi, 3rd Edition, 2015. 2. O.G. Palanna, McGraw Hill Education (India) Private Limited, 9th Reprint, 2015. 3. B. Sivasankar, Engineering Chemistry 1st Edition, McGraw Hill Education (India), 2008 4. "Photovoltaic solar energy: From fundamentals to Applications", Angèle Reinders, Pierre Verlinden, Wilfried van Sark, Alexandre Freundlich, Wiley publishers, 2017. 		
Reference Books		
<ol style="list-style-type: none"> 1. O. V. Roussak and H. D. Gesser, Applied Chemistry – A Text Book for Engineers and Technologists, Springer Science Business Media, New York, 2nd Edition, 2013. 2. S. S. Dara, A Textbook of Engineering Chemistry, S. Chand & Co Ltd., New Delhi, 20th Edition, 2013. 		
Mode of Evaluation: Internal Assessment (CAT, Quizzes, Digital Assignments) & FAT		
List of Experiments		CO:6
1.	Water Purification: Estimation of water hardness by EDTA method and its removal by ion-exchanger resin	1 h 30 min
2.	Water Quality Monitoring: Assessment of total dissolved oxygen in different water samples by Winkler's method	3 h
3.	Estimation of sulphate/ chloride in drinking water by conductivity method	
4/5	Material Analysis: Quantitative colorimetric determination of divalent metal ions of Ni/Fe/Cu using conventional and smart phone digital-imaging methods	3h
6.	Analysis of Iron in carbon steel by potentiometry	1 h 30 min

7.	ConstructionandworkingofanZn-Cuelectrochemicalcell	1 h 30 min
8.	Determination of viscosity-average molecularweight of differentnatural/syntheticpolymers	1 h 30 min
9.	Arduinomicrocontrollerbasedsensorformonitoringtemperature/con ductivityin samples.	1 h 30 min
Total Laboratory Hours		17 hours
Modeof Evaluation: Viva-voceandLab performance&FAT		
RecommendedbyBoardof Studies	31-05-2019	
ApprovedbyAcademicCouncil	55 th ACM	Date 13-06-2019

Coursecode	PROBLEM SOLVING AND PROGRAMMING	L	T	P	J	C
CSE1001		0	0	6	0	3
Pre-requisite	NIL	Syllabus version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. To develop broad understanding of computers, programming languages and their generations 2. Introduce the essential skills for a logical thinking for problem solving 3. To gain expertise in essential skills in programming for problem solving using computer 						
Course Outcome:						
<ol style="list-style-type: none"> 1. Understand the working principle of a computer and identify the purpose of a computer programming language 2. Learn various problem solving approaches and ability to identify an appropriate approach to solve the problem 3. Differentiate the programming Language constructs appropriately to solve any problem 4. Solve various engineering problems using different data structures 5. Able to modulate the given problem using structural approach of programming 6. Efficiently handle data using at less to process and store data for the given problem 						
List of Challenging Experiments (Indicative)						
1.	Steps in Problem Solving Drawing Flowchart using Edtool/Raptor Tool	4 hours				
2.	Introduction to Python, Demo on IDE, Keywords, Identifiers, I/O Statements, Simple Program to display Hello world in Python.	4 hours				
3.	Operators and Expressions in Python	4 hours				
4.	Algorithmic Approach 1: Sequential	2				
5.	Algorithmic Approach 2: Selection (if, elif, if.. else, nested if else)	2 hours				
6.	Algorithmic Approach 3: Iteration (while and for)	4 hours				
7.	Strings and its Operations	2 hours				
8.	Regular Expressions	2 hours				
9.	List and its operations.	2 hours				
10.	Dictionaries: operations	2 hours				
11.	Tuples and its operations	2 hours				
12.	Set and its operations	2 hours				
13.	Functions, Recursions	2 hours				
14.	Sorting Techniques (Bubble/Selection/Insertion)	4 hours				
15.	Searching Techniques : Sequential Search and Binary Search	3 hours				
16.	Files and its Operations	4 hours				
Total Laboratory hours						45 hours

Text Book(s)			
1.	John V. Guttag., 2016.Introduction to computation andprogrammingusingpython: withapplications to understandingdata.PHIPublisher.		
Reference Books			
1.	CharlesSeverance.2016.Pythonforeverybody: exploringdata in Python 3,CharlesSeverance.		
2	CharlesDierbach.2013.Introduction to computer scienceusingpython: a computationalproblem-solvingfocus. WileyPublishers.ModeofEvaluation: PAT/		
Modeof Evaluation: CAT / Assignment / Quiz/ FAT / Project / Seminar			
RecommendedbyBoardof Studies		04-04-2014	
ApprovedbyAcademicCouncil		38 th AC	Date 23-10-2015

Coursecode	Problem Solving AndObjectOrientedProgramming	L	T	P	J	C
CSE1002		0	0	6	0	3
Pre-requisite	NIL	Syllabusversion				
		v1.0				
CourseObjectives:						
<ol style="list-style-type: none"> 1. To emphasize thebenefits of objectorientedconcepts. 2. To enablestudents to solvethereal time applications usingobject orientedprogrammingfeatures 3. To improvetheskills of a logical thinkingand tosolve the problems usinganyprocessinglelements 						
CourseOutcomes:						
<ol style="list-style-type: none"> 1. Recall the basics of procedural programming and to represent the real world entities as programming constructs 2. Enumerate object oriented concepts and translate real-world applications into graphical representations 3. Demonstrate the usage of classes and objects of the real world entities in applications 4. Discriminate the reusability and multiple interfaces with same functionality based features to solve complex computing problems 5. Propose possible error-handling constructs for unanticipated states/inputs and to use generic programming constructs to accommodate different datatypes 6. Validate the program against file inputs towards solving the problem 						
Module:1	StructuredProgramming	12 hours				
StructuredProgramming conditionaland loopingstatements- arrays -functions- pointers - dynamic memoryallocation- structure						
Module:2	Introduction to objectorientedapproach	10 hours				
Introduction to object orientedapproach:Whyobject orientedprogramming?- Characteristics of objectorientedlanguage:classesand objects-encapsulation -data abstraction- inheritance - polymorphism-Merits and Demerits of object orientedprogramming.UML-classdiagram ofOOP -Inline functiondefaultargument function-Exceptionhandling(Standard) - reference:independentreference function returningreference passbyreference.						
Module:3	Classes and objects	14 hours				
Classesandobjects: Definition of classesaccessspecierclassversus structureconstructordestructor copyconstructorand its importance arrayofobjectsdynamic objects- friendfunction-friendclass						
Module:4	Polymorphism and Inheritance	26 hours				
Polymorphism andInheritance:Polymorphism-compile time polymorphismfunction overloadingoperatoroverloading.Inheritance- typesof inheritance - constructorsand destructors in inheritance constraints of multiple inheritance-virtual baseclass -run time polymorphism - functionoverriding.						
Module:5	Exceptionhandling and Templates	18 hours				
Exceptionhandling and TemplatesExceptionhandling(user-denedexception) - Functiontemplate, Class templateTemplatewithinheritance ,STLContainer,Algorithm,Iterator-vector,list, stack, map.						

Module:6	IO Streams and Files	10hours
IOstreams and Files IOstreams, Manipulators- overloading Inserters(<<) and Extractors(>>)Sequential and Random files – writing and reading objects into/fromfiles		
Total Lecture hours:		90hours
Text Book(s)		
1.	StanleyB Lippman, JoseeLajoie,BarbaraE, Moo, C++ primer,Fifthedition,Addison-Wesley, 2012.	
2	Ali Bahrami,Object oriented Systemsdevelopment,TataMcGraw-Hill Education, 1999.	
3	Brian W. Kernighan, Dennis M. Ritchie ,TheC programmingLanguage, 2nd edition,Prentice HallInc., 1988.	
Reference Books		
1.	Bjarne stroustrup, The C++ programmingLanguage,Addison Wesley, 4thedition, 2013.	
2	HarveyM. Deitel and Paul J.Deitel,C++ Howto Program, 7th edition,Prentice Hall, 2010.	
3	MaureenSprankleand Jim Hubbard,Problem solvingandProgrammingconcepts, 9th	
Modeof Evaluation: CAT / Assignment / Quiz/ FAT / Project / Seminar		
List ofChallengingExperiments(Indicative)		
1	PostmanProblem Apostmanneedstowalkdowneverystreetinhisareainordertodeliverthemail.Assumethathedistancesbetweenthestreetsalongtheroadsaregiven.Thepostmanstartsatthepostonceandreturnsbacktothepost_oeafterdeliveringallthemails.Implementanalgorithmtohelphthepostmantowalkminimum distance forthe purpose.	5hrs
2	BudgetAllocation for MarketingCampaign AmobilemanufacturingcompanyhasgotseveralmarketingoptionssuchasRadioadvertisementcampaign,TVnon-peakhourscampaign,Citytoppapernetwork,Viralmarketingcampaign,Webadvertising.Fromtheirpreviousexperience,theyhavegotastatisticsaboutpaybacksforeachmarketingoption.Giventhemarketingbudget(rupeesincrores)fortheurrentyearanddetailsofpaybacksforeachoption,implementanalgorithmtodeterminetheamountthatshallspentoneachmarketingoptionsothatthecompanyattainsthemaximumpro_t.	5 hrs.
3	MissionariesandCannibals Threemissionariesandthree cannibals areon oneside ofariver, alongwith aboatthatcan hold one or two people.Implementan algorithm to find awayto geteveryone to theother sideof the river,without ever leavingagroupofmissionaries in one placeoutnumberedbythe cannibals in that place.	5hrs.

4	Register Allocation Problem A register is a component of a computer processor that can hold any type of data and can be accessed faster. As registers are faster to access, it is desirable to use them to the maximum so that the code execution is faster. For each code submitted to the processor, a register interference graph (RIG) is constructed. In a RIG, a node represents a temporary variable and an edge is added between two nodes (variables) t1 and t2 if they are live simultaneously at some point in the program. During register allocation, two temporaries can be allocated to the same register if there is no edge connecting them. Given a RIG representing the dependencies between variables in a code, implement an algorithm to determine the number of registers required to store the variables and speed up the code execution.	5 hrs
5	Selective Job Scheduling Problem A server is a machine that waits for requests from other machines and responds to them. The purpose of a server is to share hardware and software resources among clients. All the clients submit the jobs to the server for execution and the server may get multiple requests at a time. In such a situation, the server schedules the jobs submitted to it based on some criteria and logic. Each job contains two values namely time and memory required for execution. Assume that there are two servers that schedule jobs based on time and memory. The servers are named as Time_Schedule_Server and memory_Schedule_Server respectively. Design a OOP model and implement the time_Schedule_Server and memory_Schedule_Server. The Time_Schedule_Server arranges jobs based on time required for execution in ascending order whereas memory_Schedule_Server arranges jobs based on memory required for execution in ascending order.	5 hrs
6	Fragment Assembly in DNA Sequencing DNA, or deoxyribonucleic acid, is the hereditary material in humans and almost all other organisms. The information in DNA is stored as a code made up of four chemical bases: adenine (A), guanine (G), cytosine (C), and thymine (T). In DNA sequencing, each DNA is sheared into millions of small fragments (reads) which assemble to form a single genomic sequence ("superstring"). Each read is a small string. In such a fragment assembly, given a set of reads, the objective is to determine the shortest superstring that contains all the reads. For example, given a set of strings, {000, 001, 010, 011, 100, 101, 110, 111} the shortest superstring is 0001110100. Given a set of reads, implement an algorithm to find the shortest superstring that contains all the given reads.	5 hrs
7	House Wiring An electrician is wiring a house which has many rooms. Each room has many power points in different locations. Given a set of power points and the distances between them, implement an algorithm to find the minimum cable required.	5 hrs
Recommended by Board of Studies		29.10.2015
Approved by Academic Council		39 th AC Date 17-12-2015

CourseCode	CourseTitle	L	T	P	J	C
ENG1901	TechnicalEnglish-I	0	0	4	0	2
Pre-requisite	FoundationEnglish-II	SyllabusVersion				
		1				
CourseObjectives:						
<ol style="list-style-type: none"> To enhance students' knowledgeofgrammarandvocabularyto readandwrite error-free language in real lifesituations. To make the students'practice the most commonareasofwrittenand spoken communications skills. To improvestudents' communicative competencythrough listeningand speakingactivities in theclassroom. 						
CourseOutcomes:						
<ol style="list-style-type: none"> Develop a betterunderstandingofadvancedgrammarrulesandwritegrammaticallycorrectsentences. Acquirewidevocabularyandlearnstrategies for error-free communication. Comprehendlanguage and improvespeakingskills in academicandsocialcontexts. Improve listeningskills so as to understand complexbusinesscommunication in a varietyofglobalEnglishaccentsthrough proper pronunciation. Interpret texts, diagramsandimproveboth readingandwritingskills which would helpthem intheir academic as well asprofessionalcareer. 						
Module:1	AdvancedGrammar					4 hours
Articles,Tenses,Voice and Prepositions Activity:Worksheets onImpersonal PassiveVoice, Exercisesfrom the prescribed text						
Module:2	VocabularyBuilding I					4 hours
IdiomsandPhrases,Homononyms,Homophones and HomographsActivity:Jigsaw Puzzles;VocabularyActivitiesthrough Web tools						
Module:3	Listeningfor SpecificPurposes					4 hours
Gist, monologues, shortconversations, announcements, briefingsanddiscussionsActivity:Gapfilling;Interpretations						
Module:4	Speakingfor Expression					6 hours
Introducingoneselfandothers,MakingRequests&responses,Inviting and Accepting/DecliningInvitationsActivity:Brief introductions; Role-Play; Skit.						
Module:5	Reading for Information					4 hours
ReadingShort Passages,NewsArticles,TechnicalPapersand Short StoriesActivity: Readingspecificnewspaperarticles;blogs						
Module:6	WritingStrategies					4 hours
Joiningthe sentences, word order, sequencingthe ideas, introductionand conclusionActivity: Short Paragraphs; Describingfamiliarevents; storywriting						
Module:7	VocabularyBuilding II					
Enrich the domain specific vocabularybydescribingObjects, Charts, Food,Sports andEmployment. Activity: Describing Objects, Charts, Food, Sports andEmployment						

Module:8	ListeningforDaily Life	4 hours
Listeningforstatistical information, Shortextracts, Radio broadcastsandTV interviewsActivity: Takingnotes and Summarizing		
Module:9	ExpressingIdeasandOpinions	6 hours
Telephonic conversations,Interpretation ofVisuals and describingproducts andprocesses.Activity: Role-Play(Telephonic); DescribingProducts and Processes		
Module:10	ComprehensiveReading	4 hours
ReadingComprehension,Makinginferences, ReadingGraphics, Note-making, andCriticalReading. Activity: Sentence Completion;Cloze Tests		
Module:11	Narration	4 hours
Writingnarrative short story,Personalmilestones,officiallettersand E-mails.Activity: Writingan E-mail;Improvingvocabularyand writingskills.		
Module:12	Pronunciation	4 hours
Speech Sounds, Word Stress,Intonation, Various accents Activity:PracticingPronunciation through web tools;Listeningto various accents of English		
Module:13	Editing	4 hours
Simple, Complex&CompoundSentences,Direct&Indirect Speech, Correction of Errors,Punctuations. Activity:PracticingGrammar		
Module:14	ShortStory Analysis	4 hours
“TheBoundary”byJhumpaLahiri Activity: Readingandanalyzingthethemeof the short story.		
Total Lecture hours		60 hours
Text Book / Workbook		
1.	Wren,P.C.;Martin,H.;PrasadaRao,N.D.V.(1973–2010). <i>HighSchoolEnglishGrammar &Composition</i> . New Delhi: SultanChandPublishers.	
2	Kumar, Sanjay,;PushpLatha.(2018) <i>EnglishLanguageandCommunicationSkillsforEngineers</i> ,I ndia:OxfordUniversityPress.	
Reference Books		
1.	GupthaS C, (2012) <i>Practical English Grammar &Composition</i> , 1 st Edition,India:Arihant Publishers	

2.	StevenBrown,(2011)Dorolyn Smith, <i>Active Listening</i> 3, 3 rd Edition, UK:Cambridge UniversityPress.
3.	LizHamp-Lyons,BenHeasley,(2010) <i>Study Writing</i> ,2 nd Edition, UK:Cambridge UniversityPres.
4.	KennethAnderson,JoanMaclean,(2013)TonyLynch, <i>StudySpeaking</i> ,2 nd Edition,UK: Cambridge,UniversityPress.
5.	EricH.Glendingning,BeverlyHolmstrom, (2012) <i>Study Reading</i> ,2 nd Edition, UK: CambridgeUniversityPress.
6.	MichaelSwan,(2017) <i>PracticalEnglishUsage</i> (PracticalEnglishUsage),4thedition,UK:Oxford UniversityPress.
7.	MichaelMcCarthy,FelicityO'Dell,(2015) <i>EnglishVocabularyinUseAdvanced</i> (SouthAsian Edition), UK:CambridgeUniversityPress.
8.	MichaelSwan,CatherineWalter,(2012) <i>OxfordEnglishGrammarCourseAdvanced</i> ,Feb,4 th Edition, UK: OxfordUniversityPress.
9.	Watkins,Peter.(2018) <i>TeachingandDevelopingReadingSkills: CambridgeHandbooksfor Language teachers</i> ,UK:Cambridge UniversityPress.
10.	(<i>The Boundary</i> byJhumpa Lahiri)URL: https://www.newyorker.com/magazine/2018/01/29/the-boundary?intcid=inline_amp
Mode ofevaluation: Quizzes,Presentation, Discussion, Role play, Assignments andFAT	
List ofChallengingExperiments(Indicative)	
1.	Self-Introduction 12 hours
2.	Sequencing Ideasand Writinga Paragraph 12 hours
3.	ReadingandAnalyzing Technical Articles 8 hours
4.	ListeningforSpecificityinInterviews(ContentSpecific) 12 hours
5.	IdentifyingErrors in a Sentence or Paragraph 8 hours
6.	WritinganE-mailbynarratinglife events 8 hours
Total Laboratory Hours	
60 hours	
Mode ofevaluation: Quizzes,Presentation, Discussion, Role play, Assignments andFAT	
Recommendedby Board ofStudies	08.06.2019
Approved by AcademicCouncil	55 Date:13-06-2019

CourseCode	CourseTitle	L	T	P	J	C	
ENG1902	TechnicalEnglish-II	0	0	4	0	2	
Pre-requisite	71%to 90% EPT score	SyllabusVersion					1
CourseObjectives:							
1. Toacquireproficiencylevels inLSRWskills on parwith the requirementsforplacementinterviews of high-endcompanies/ competitive exams. 2. Toevaluecomplexargumentsand to articulate their ownpositions on arangeof technicalandgeneraltopics. 3. Tospeak ingrammaticalandacceptable English with minimalMTI,aswellasdevelopavast andactive vocabulary.							
CourseOutcomes:							
1. Communicate proficientlyin high-endinterviewsand exam situationsandallsocialsituations 2. Comprehendacademicarticlesanddraw inferences 3. Evaluatedifferentperspectiveson atopicaltopic 4. Writeclearlyand convincinglyin academicas wellasgeneralcontexts 5. Synthesize complexconcepts and presentthem in speechandwriting							
Module:1	Listeningfor ClearPronunciation					4 hours	
Ice-breaking,Introduction to vowels,consonants,diphthongs. Listeningto formalconversations in BritishandAmerican accents (BBC and CNN)aswellas other 'native' accents Activity:Factalandinterpretive exercises; note-makingin a varietyofglobalEnglishaccents							
Module:2	IntroducingOneself					4 hours	
Speaking:Individual Presentations Activity:Self-Introductions, Extempore speech							
Module:3	EffectiveWriting					6 hours	
Writing:BusinesslettersandEmails, Minutes andMemos Structure/template ofcommon businesslettersandemails:inquiry/ complaint/ placingan order;Formats of Minutes andMemos Activity: Students writea businessletterand Minutes/ Memo							
Module:4	ComprehensiveReading					4 hours	
Reading: ReadingComprehensionPassages,Sentence Completion(Technical and GeneralInterest),Vocabularyand Word Analogy Activities: Clozetest, Logicalreasoning, Advanced grammarercises							
Module:5	Listening to Narratives					4 hours	
Listening:Listeningto audio files of shortstories,News, TVClips/ Documentaries, MotivationalSpeeches in UK/ US/globalEnglishaccents. Activity: Note-makingandInterpretiveexercises							
Module:6	Academic Writing andEditing					6 hours	
Writing:Editing/ ProofreadingsymbolsCitationFormats Structureof anAbstractandResearch Paper Activity: WritingAbstractsand research paper; Work withEditing/Proofreading exercise							
Module:7	Team Communication					4 hours	
Speaking: GroupDiscussionsandDebates oncomplex/ contemporarytopicsDiscussionevaluationparameters, usinglogic in debates Activity:Group Discussions on general topics							

Module:8	Career-oriented Writing	4 hours
Writing: Resumes and Job Application Letters, SOP Activity: Writing resumes and SOPs		
Module:9	Reading for Pleasure	4 hours
Reading: Reading short stories Activity: Classroom discussion and note-making, critical appreciation of the short story		
Module:10	Creative Writing	4 hours
Writing: Imaginative, narrative and descriptive prose Activity: Writing about personal experiences, unforgettable incidents, travelogues		
Module:11	Academic Listening	4 hours
Listening: Listening in academic contexts Activity: Listening to lectures, Academic Discussions, Debates, Review Presentations, Research Talks, Project Review Meetings		
Module:12	Reading Nature-based Narratives	4 hours
Narratives on Climate Change, Nature and Environment Activity: Classroom discussions, student presentations		
Module:13	Technical Proposals	4 hours
Writing: Technical Proposals Activities: Writing a technical proposal		
Module:14	Presentation Skills	4 hours
Persuasive and Content-Specific Presentations Activity: Technical Presentations		
Total Lecture hours:		60 hours
Text Book / Workbook		
1.	Oxenden, Clive and Christina Latham-Koenig. <i>New English File: Advanced Students Book</i> . Paperback. Oxford University Press, UK,	
2.	Rizvi, Ashraf. <i>Effective Technical Communication</i> . McGraw-Hill India, 2017.	
Reference Books		
1.	Oxenden, Clive and Christina Latham-Koenig. <i>New English File: Advanced: Teacher's Book with Test and Assessment</i> . CD-ROM: Six-level General English Course for Adults. Paperback. Oxford University Press, UK, 2013.	
2.	Balasubramanian, T. <i>English Phonetics for the Indian Students: A Workbook</i> . Laxmi Publications, 2016.	
3.	Philip Seargeant and Bill Greenwell, <i>From Language to Creative Writing</i> . Bloomsbury Academic, 2013.	
4.	Krishnaswamy, N. <i>Eco-English</i> . Bloomsbury India, 2015.	
5.	Manto, Saadat Hasan. <i>Selected Short Stories</i> . Trans. Aatish Taseer. Random House India, 2012.	
6.	Ghosh, Amitav. <i>The Hungry Tide</i> . Harper Collins, 2016.	
7.	Ghosh, Amitav. <i>The Great Derangement: Climate Change and the Unthinkable</i> . Penguin Books, 2016.	
8.	<i>The MLA Handbook for Writers of Research Papers</i> , 8th ed. 2016.	
Online Sources:		

(75 short stories) http://www.ecoction.org/dt/thinking.html (Leopold, Aldo. "Thinking like a Mountain") https://www.esl-lab.com/ ; http://www.bbc.co.uk/learningenglish/ ; https://www.bbc.com/news ; https://learningenglish.voanews.com/a/using-voa-learning-english-to-improve-listening-skills/3815547.html		
Mode of evaluation: Quizzes, Presentation, Discussion, Role play, Assignments and FAT		
List of Challenging Experiments (Indicative)		
1.	Self-Introduction using SWOT	12 hours
2.	Writing minutes of meetings	10 hours
3.	Writing an abstract	10 hours
4.	Listening to motivational speeches and interpretation	10 hours
5.	Cloze Test	6 hours
6.	Writing a proposal	12 hours
Total Laboratory Hours		60 hours
Mode of evaluation: Quizzes, Presentation, Discussion, Role play, Assignments and FAT		
Recommended by Board of Studies	08.06.2019	
Approved by Academic Council	55	Date: 13-06-2019

CourseCode	Coursetitle	L	T	P	J	C
ENG1903	AdvancedTechnicalEnglish	0	0	2	4	2
Pre-requisite	Greaterthan 90 %EPT score	SyllabusVersion				
		1				
CourseObjectives:						
<ol style="list-style-type: none"> 1. To review literaturein anyform oranytechnical article 2. To infercontent in social mediaandrespondaccordingly 3. To communicatewith people across theglobe overcomingtrans-cultural barriersandnegotiatesuccessfully 						
CourseOutcomes:						
<ol style="list-style-type: none"> 1. Analyzecriticallyand write goodreviews 2. Articulate research papers, projectproposals andreports 3. Communicate effectivelyin a trans-cultural environment 4. Negotiateand lead teamstowardsuccess 5. Presentideas in an effective manner usingweb tools 						
Module:1	Negotiationand Decision MakingSkillsthroughLiterary Analysis	5 hours				
Concepts of NegotiationandDecision MakingSkills Activity:Analysis of excerpts fromShakespeare’s“TheMerchant ofVenice” (court scene)and discussion on negotiationskills. Criticalevaluation of excerpts fromShakespeare’s“Hamlet”(Monologue byHamlet)anddiscussion on decision makingskills						
Module:2	Writingreviews and abstractsthroughmovieinterpretations	5 hours				
Review writingandabstract writingwith competency Activity: WatchingCharlesDickens“Great Expectations” andwriting amovie review WatchingWilliam F.Nolan’s“Logan’s Run” andanalyzingit in tunewith the presentscenario of depletion of resources and writinganabstract						
Module:3	TechnicalWriting	4 hours				
Stimulateeffective linguisticsforwriting:content and styleActivity:Proofreading Statement ofPurpose						
Module:4	Trans-CulturalCommunication	4 hours				
NuancesofTrans-culturalcommunicationActivity: Groupdiscussionand case studies ontrans-culturalcommunication.Debate on trans-culturalcommunication.						
Module:5	ReportWritingand ContentWriting	4 hours				
Enhancingreportage onrelevant audio-visualsActivity: Watch adocumentaryonsocialissuesand draft a reportIdentifyavideo onanysocial issueandinterpret						
Module:6	Draftingproject proposals andarticle writing	4 hours				
Dynamics of draftingproject proposals and research articles Activity: Writinga project proposal.Writinga researcharticle.						

Module:7	Technical Presentations	4 hours
Build smart presentation skills and strategies Activity: Technical presentations using PPT and Web tools		
Total Lecture hours		30 hours
Text Book/ Workbook		
1.	Raman, Meenakshi & Sangeeta Sharma. <i>Technical Communication: Principles and Practice</i> , 3 rd edition, Oxford University Press, 2015.	
Reference Books		
1	Basu B.N. <i>Technical Writing</i> , 2011 Kindle edition	
2	Arathoon, Anita. <i>Shakespeare's The Merchant of Venice</i> (Text with Paraphrase), Evergreen Publishers, 2015.	
3	Kumar, Sanjay and Pushp Lata. <i>English Language and Communication Skills for Engineers</i> , Oxford University Press, India, 2018.	
4	Frantisek, Burda. <i>On Transcultural Communication</i> , 2015, LAP Lambert Academic Publishing, UK.	
5	Geever, C. Jane. <i>The Foundation Center's Guide to Proposal Writing</i> , 5 th Edition, 2007, Reprint 2012 The Foundation Center, USA.	
6	Young, Milena. <i>Hacking Your Statement of Purpose: A Concise Guide to Writing Your SOP</i> , 2014 Kindle Edition.	
7	Ray, Ratri, <i>William Shakespeare's Hamlet</i> , The Atlantic Publishers, 2011.	
8	C Muralikrishna & Sunitha Mishra, <i>Communication Skills for Engineers</i> , 2 nd edition, NY: Pearson, 2011.	
Mode of Evaluation: Quizzes, Presentation, Discussion, Role Play, Assignments		
List of Challenging Experiments (Indicative)		
1.	Enacting a court scene- Speaking	6 hours
2.	Watching a movie and writing a review	4 hours
3.	Trans-cultural – case studies	2 hours
4.	Drafting a report on any social issue	6 hours
5.	Technical Presentation using web tools	6 hours
6.	Writing a research paper	6 hours
J- Component Sample Projects		
1.	Short Films	
2.	Field Visits and Reporting	
3.	Case studies	
4.	Writing blogs	
5.	Vlogging	
Total Hours (J-Component)		60 hours
Mode of evaluation: Quizzes, Presentation, Discussion, Role play, Assignments and FAT		
Recommended by Board of Studies	08.06.2019	
Approved by Academic Council	55	Date: 13-06-2019

CourseCode	Ethics and Values	L	T	P	J	C
HUM1021 / HUM1032		2	0	0	0	2
Pre-requisite	Nil	SyllabusVersion				
		1.1				
CourseObjectives:						
1. To understand and appreciate the ethical issues faced by an individual in profession, society and polity 2. To understand the negative health impacts of certain unhealthy behaviors 3. To appreciate the need and importance of physical, emotional health and social health						
CourseOutcomes:						
Students will be able to:						
1. Comprehend moral and ethical values. 2. Understand various social problems and learn to act ethically 3. Understand the concept of addiction and how it will affect the physical and mental health 4. Identify ethical concerns in research and intellectual contexts, including academic integrity, use and citation of sources, the objective presentation of data, and the treatment of human subjects 5. Identify the main typologies, characteristics, activities, actors and forms of cybercrime						
Module:1	Being Good and Responsible	5 hours				
Gandhian values such as truth and non-violence – Comparative analysis on leaders of past and present – Society's interests versus self-interests- Personal Social Responsibility: Helping the needy, charity and serving the society						
Module:2	Social Issues 1	4 hours				
Harassment – Types - Prevention of harassment, Violence and Terrorism						
Module:3	Social Issues 2	4 hours				
Corruption: Ethical values, causes, impact, laws, prevention – Electoral malpractices; White collar crimes - Tax evasions – Unfair trade practices						
Module:4	Addiction and Health	5 hours				
Peer pressure- Alcoholism: Ethical values, causes, impact, laws, prevention – Il-effects of smoking- Prevention of Suicides; Sexual Health: Prevention and impact of pre-marital pregnancy and Sexually Transmitted Diseases						
Module:5	Drug Abuse	3 hours				
Abuse of different types of legal and illegal drugs: Ethical values, causes, impact, laws and prevention						
Module:6	Personal and Professional Ethics	4 hours				
Dishonesty-Stealing-Malpractices in Examinations– Plagiarism						
Module:7	Abuse of Technologies	3 hours				

Hacking and other cybercrimes, Addiction to mobile phone usage, Videogames and Social networking web sites			
Module:8	Contemporary Issues:	2 hours	
Guest lectures by Industrial Experts			
		Total Lecture Hours:	30 hours
Reference Books			
1.	Dhaliwal, K. K (2016), "Gandhian Philosophy of Ethics: A Study of Relationship between his Presupposition and Precepts, Writers Choice, New Delhi, India.		
2.	Vittal, N (2012), "Ending Corruption? - How to Clean up India?", Penguin Publishers, UK.		
3.	Pagliaro, L. A. and Pagliaro, A. M (2012), "Handbook of Child and Adolescent Drug and Substance Abuse: Pharmacological, Developmental and Clinical Considerations", Wiley Publishers, U.S.A.		
4.	Pandey, P. K (2012), "Sexual Harassment and Law in India", Lambert Publishers, Germany.		
Mode of Evaluation: Quizzes, CAT, FAT, Digital assignments, poster/collage making and Seminars			
Recommended by Board of Studies		26-07-2017	
Approved by Academic Council		No. 46	Date 24-08-2017

CourseCode	Calculus for Engineers	L	T	P	J	C
MAT1011		3	0	2	0	4
Pre-requisite	10+2 Mathematics or MAT1001	Syllabus Version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> To provide the requisite and relevant background necessary to understand the other important engineering mathematics courses offered for Engineers and Scientists. To introduce important topics of applied mathematics, namely Single and Multivariable Calculus and Vector Calculus etc. To impart the knowledge of Laplace transform, an important transform technique for Engineers which requires knowledge of integration 						
Course Outcomes:						
<ol style="list-style-type: none"> Apply differentiation to solve max/min problems and compute volumes of revolution and surface areas of revolution using Integration. Apply the concepts of Laplace Transforms and solve problems with periodic functions, step functions, impulse functions and convolution. Evaluate partial derivatives, limits, total differentials, Jacobians, Taylor series and optimization problems involving several variables. Evaluate multiple integrals in Cartesian, Polar, Cylindrical and Spherical coordinates. Analyse the concepts of gradient, directional derivatives, divergence, curl and apply them to find the circulation, work done, conservative field and Greens', Stokes, Gauss divergence theorem. Develop programming tools for engineering problems and visualize solutions. 						
Module:1	Application of Single Variable Calculus	9 hours				
Differentiation- Extrema on an Interval-Rolle's Theorem and the Mean Value Theorem- Increasing and Decreasing functions and First derivative test-Second derivative test- Maxima and Minima-Concavity. Integration-Average function value - Area between curves- Volumes of solids of revolution -						
Module:2	Laplace transforms	7 hours				
Definition of Laplace transform- Properties- Laplace transform of periodic functions- Laplace transform of unit step function, Impulse function- Inverse Laplace transform- Convolution.						
Module:3	Multivariable Calculus	4 hours				
Functions of two variables- limits and continuity- partial derivatives- total differential- Jacobian and its properties.						
Module:4	Application of Multivariable Calculus	5 hours				
Taylor's expansion for two variables- maxima and minima- constrained maxima and minima- Lagrange's multiplier method.						
Module:5	Multiple integrals	8 hours				

Evaluation of double integrals – change of order of integration – change of variables between Cartesian and polar co-ordinates – Evaluation of triple integrals – change of variables between Cartesian and cylindrical and spherical co-ordinates – Beta and Gamma functions – interrelation – evaluation of multiple integrals using gamma and beta functions.			
Module:6	Vector Differentiation	5 hours	
Scalar and vector valued functions – gradient, tangent plane – directional derivative – divergence and curl – scalar and vector potentials – Statement of vector identities – Simple problems			
Module:7	Vector Integration	5 hours	
line, surface and volume integrals – Statement of Green's, Stoke's and Gauss divergence theorems – verification and evaluation of vector integrals using them.			
Module:8	Contemporary Issues:	2 hours	
Industry Expert Lecture			
Total Lecture hours:			45 hours
Text Book(s)			
[1] Thomas' Calculus, George B. Thomas, D. Weir and J. Hass, 13 th edition, Pearson, 2014. [2] Advanced Engineering Mathematics, Erwin Kreyszig, 10 th Edition, Wiley India, 2015.			
Reference Books			
1. Higher Engineering Mathematics, B.S. Grewal, 43 rd Edition, Khanna Publishers, 2015			
2. Higher Engineering Mathematics, John Bird, 6 th Edition, Elsevier Limited, 2017.			
3. Calculus: Early Transcendentals, James Stewart, 8 th edition, Cengage Learning, 2017.			
4. Engineering Mathematics, K.A. Stroud and Dexter J. Booth, 7 th Edition, Palgrave Macmillan (2013)			
Mode of Evaluation			
Digital Assignments, Quiz, Continuous Assessments, Final Assessment Test			
List of Challenging Experiments (Indicative)			
1.	Introduction to MATLAB through matrices, and general Syntax	2 hours	
2.	Plotting and visualizing curves and surfaces in MATLAB – Symbolic computations using MATLAB	2 hours	
3.	Evaluating Extremum of a single variable function	2 hours	
4.	Understanding integration as Area under the curve	2 hours	
5.	Evaluation of Volume by Integrals (Solids of Revolution)	2 hours	
6.	Evaluating maxima and minima of functions of several variables	2 hours	
7.	Applying Lagrange multiplier optimization method	2 hours	
8.	Evaluating Volume under surfaces	2 hours	
9.	Evaluating triple integrals	2 hours	
10.	Evaluating gradient, curl and divergence	2 hours	
11.	Evaluating line integrals in vectors	2 hours	
12.	Applying Green's theorem to real world problems	2 hours	
Total Laboratory Hours			24 hours
Mode of Evaluation:			
Weekly Assessment, Final Assessment Test			
Recommended by Board of Studies		03-06-2019	
Approved by Academic Council		No. 55	Date 13-06-2019

CourseCode	Statistics for Engineers	L	T	P	J	C
MAT2001		3	0	2	0	4
Prerequisites	MAT1011 – Calculus for Engineers	Syllabus Version:				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> To provide students with a framework that will help them choose the appropriate descriptive methods in various data analysis situations. To analyse distributions and relationship of real-time data. To apply estimation and testing methods to make inference and modelling techniques for decision making. 						
Course Outcomes:						
<ol style="list-style-type: none"> Analyze statistical data using measures of central tendency and dispersion. Analyze and apply the concepts of random variables and distribution functions to obtain moments and characteristic functions. Analyze the experimental data using correlation and regression analysis and interpret the results. Apply the concepts of inferential statistics and interpret the results. Apply the statistical methodology in solving reliability engineering problems. Develop programming tools for engineering problems and visualize solutions. 						
Module:1 Introduction to Statistics 6 hours						
Introduction to statistics and data analysis- Measures of central tendency – Measures of variability- [Moments-Skewness-Kurtosis (Concepts only)].						
Module:2 Random variables 8 hours						
Introduction-random variables-Probability mass Function, distribution and density functions -joint Probability distribution and joint density functions- Marginal, conditional distribution and density functions- Mathematical expectation, and its properties Covariance, moment generating function – characteristic function.						
Module:3 Correlation and regression 4 hours						
Correlation and Regression – Rank Correlation- Partial and Multiple correlation- Multiple regression.						
Module:4 Probability Distributions 7 hours						
Binomial and Poisson distributions – Normal distribution – Gamma distribution – Exponential distribution – Weibull distribution.						
Module:5 Hypothesis Testing I 4 hours						
Testing of hypothesis – Introduction- Types of errors, critical region, procedure of testing hypothesis- Large sample tests- Z test for Single Proportion, Difference of Proportion, mean and difference of means.						
Module:6 Hypothesis Testing II 9 hours						
Small sample tests- Student's t-test, F-test- chi-square test- goodness of fit- independence of attributes- Design of Experiments- Analysis of variance – one and two way classifications- CRD- RBD- LSD.						
Module:7 Reliability 5 hours						
Basic concepts- Hazard function- Reliability of series and parallel systems- System Reliability- Maintainability- Preventive and repair maintenance- Availability.						

Module:8	ContemporaryIssues	2 hours
IndustryExpertLecture		
Total Lecture hours		45 hours
Text book(s)		
1. Probability and Statistics for engineers and scientists, R.E. Walpole, R.H. Myers, S.L. Myers and K. Ye, 9 th Edition, Pearson Education (2012).		
2. Applied Statistics and Probability for Engineers, Douglas C. Montgomery, George C. Runger, 6 th Edition, John Wiley & Sons (2016).		
Reference books		
1. Reliability Engineering, E. Balagurusamy, Tata McGraw Hill, Tenth reprint 2017.		
2. Probability and Statistics, J.L. Devore, 8 th Edition, Brooks/Cole, Cengage Learning (2012).		
3. Probability and Statistics for Engineers, R.A. Johnson, Miller Freund's, 8th edition, Prentice Hall India (2011).		
4. Probability, Statistics and Reliability for Engineers and Scientists, Bilal M. Ayyub and Richard H. McCuen, 3 rd edition, CRC press (2011).		
Mode of Evaluation		
Digital Assignments (Solutions by using soft skills), Continuous Assessment Tests, Quiz, Final Assessment Test.		
List of Experiments (Indicative)		
1.	Introduction: Understanding Data types; importing/exporting data.	2 hours
2.	Computing Summary Statistics / plotting and visualizing data using Tabulation and Graphical Representations.	2 hours
3.	Applying correlation and simple linear regression model to real dataset; computing and interpreting the coefficient of determination.	2 hours
4.	Applying multiple linear regression model to real dataset; computing and interpreting the multiple coefficient of	2 hours
5.	Fitting the following probability distributions: Binomial distribution	2 hours
6.	Normal distribution, Poisson distribution	2 hours
7.	Testing of hypothesis for One sample mean and proportion from real-time problems.	2 hours
8.	Testing of hypothesis for Two sample means and proportion from real-time problems	2 hours
9.	Applying the t test for independent and dependent samples	2 hours
10.	Applying Chi-square test for goodness of fit test and Contingency test to real dataset	2 hours
11.	Performing ANOVA for real dataset for Completely randomized design, Randomized Block design, Latin square Design	2 hours
Total laboratory hours		22 hours
Mode of Evaluation: Weekly Assessment, Final Assessment Test		
Recommended by Board of Studies	03-06-2019	
Approved by Academic Council	No. 55	Date: 13-06-2019

Coursecode	LEAN START-UPMANAGEMENT	L	T	P	J	C
MGT1022		1	0	0	4	2
Pre-requisite	Nil	Syllabusversion				
		v. 2.2				
CourseObjectives:						
Theobjective of thecourseis to makeastudent to create andcommercialize the product						
CourseOutcome:						
Upon successfulcompletion ofthe coursethe students will be able to						
1. Understand developingbusiness models and growth drivers						
2. Use the businessmodelcanvas to map out keycomponents of enterprise						
3. Analyze market size, cost structure,revenue streams, and valuechain						
4. Understandbuild-measure-learnprinciples						
5. Foreseeingandquantifyingbusiness andfinancialrisks						
Module:1						
CreativityandDesignThinking (identifythe verticalforbusinessopportunity, understandyourcustomers,accuratelyassess marketopportunity)					2 hours	
Module:2						
Minimum ViableProduct(Value Proposition, CustomerSegments,Build-measure-learn process)					3 hours	
Module:3						
Business Model Development(Channels and Partners, Revenue Model andstreams,KeyResources, Activities and Costs, CustomerRelationshipsandCustomerDevelopmentProcesses,Businessmodelcanvas –the lean model-templates)					3 hours	
Module:4						
Business Plan and Access to Funding(visioningyourventure, takingthe product/ service tomarket, Market plan includingDigital&ViralMarketing, start-up finance - Costs/Profits&Losses/cash flow, Angel/VC,/BankLoansand Keyelements of raisingmoney)					3 hours	
Module:5						
Legal,Regulatory, CSR, Standards,Taxes					2 hours	
Module:6						
Contemporary discussion					2 hours	
Total Lecture hours:					15 hours	
Text Book(s)						

1.	SteveBlank,K&SRanch(2012)TheStartupOwner'sManual:TheStep-By-StepGuideforBuildinga GreatCompany, 1st edition		
2.	SteveBlank(2013)TheFour Steps to the Epiphany,K&SRanch; 2nd edition		
3.	EricRies(2011)The LeanStartup:How Today'sEntrepreneursUse ContinuousInnovation to Create RadicallySuccessfulBusinesses,CrownBusiness		
Reference Books			
1.	Steve Blank (2014) Holdinga Catbythe Tail, ,K&SRanchPublishingLLC		
2.	Karal T Ulrich, ProductDesignandDevelopment,SDEppinger,McGrawHill		
3.	PeterThiel,(2014)Zero to One:Notes on Startups, or How toBuild theFuture,CrownBusiness;		
4.	LeanAnalytics: UseDatato Build aBetter Startup Faster(LeanSeries), Alistair Croll&Benjamin Yoskovitz,O'ReillyMedia; 1 st Edition		
5.	MartyCagan, (2008)Inspired:HowToCreate ProductsCustomersLove,SVPGPress;1stedition		
RecommendedbyBoardof Studies	17-08-2017		
ApprovedbyAcademicCouncil	47	Date	05-10-2017

Course code	Course title	L	T	P	J	C
PHY1701	Engineering Physics	3	0	2	0	4
Pre-requisite	Physics of 12th standard or equivalent	Syllabus version				
		V.2.1				
Course Objectives:						
To enable the students to understand the basics of the latest advancements in Physics viz., Quantum Mechanics, Nanotechnology, Lasers, Electro Magnetic Theory and Fiber Optics.						
Expected Course Outcome: Students will be able to						
<ol style="list-style-type: none"> 1. Comprehend the dual nature of radiation and matter 2. Compute Schrodinger's equations to solve finite and infinite potential problems. 3. Analyze quantum ideas at the nanoscale. 4. Apply quantum ideas for understanding the operation and working principle of optoelectronic devices. 5. Apply the Maxwell's equations in differential and integral form. 6. Select appropriate optical fibers for different Engineering applications. 7. Explain concept of Lorentz Transformation for Engineering applications. 8. Demonstrate the quantum mechanical ideas 						
Module:1	Introduction to Modern Physics	6 hours				
Planck's concept (hypothesis), Compton Effect, Particle properties of wave: Matter Waves, Davisson Germer Experiment, Heisenberg Uncertainty Principle, Wave function, and Schrodinger equation (time dependent & independent).						
Module:2	Applications of Quantum Physics	5 hours				
Particle in a 1-D box (Eigen Value and Eigen Function), 3-D Analysis (Qualitative), Tunneling Effect (Qualitative) (AB 205), Scanning Tunneling Microscope (STM).						
Module:3	Nanophysics	5 hours				
Introduction to Nano-materials, Moore's law, Properties of Nano-materials, Quantum confinement, Quantum well, wire & dot, Carbon Nano-tubes (CNT), Applications of nanotechnology in industry.						
Module:4	Laser Principles and Engineering Application	6 hours				
Laser Characteristics, Spatial and Temporal Coherence, Einstein Coefficient & its significance, Population inversion, Two, three & four level systems, Pumping schemes, Threshold gain coefficient, Components of laser, Nd-YAG, He-Ne, CO ₂ and Dye laser and their engineering applications.						
Module:5	Electromagnetic Theory and its application	6 hours				
Physics of Divergence, Gradient and Curl, Qualitative understanding of surface and volume integral, Maxwell Equations (Qualitative), Wave Equation (Derivation), EM Waves, Phase velocity, Group velocity, Group index, Wave guide (Qualitative)						
Module:6	Propagation of EM waves in Optical fibers and Optoelectronic Devices	10 hours				
Light propagation through fibers, Acceptance angle, Numerical Aperture, Types of fibers - step index, graded index, single mode & multimode, Attenuation, Dispersion-intermodal and intramodal. Sources-LED & Laser Diode, Detectors-Photodetectors- PN & PIN - Applications of fiber optics in communication- Endoscopy.						
Module:7	Special Theory of Relativity	5 hours				
Frame of reference, Galilean relativity, Postulate of special theory of relativity, Simultaneity, length contraction and time dilation.						
Module:8	Contemporary issues:	2 hours				
Lecture by Industry Experts						
Total Lecture hours:						45 hours
Text Book(s)						

1.	Arthur Beiser et al., Concepts of Modern Physics, 2013, Sixth Edition, Tata McGraw Hill. William Silfvast,		
2.	Laser Fundamentals, 2008, Cambridge University Press.		
3.	D. J. Griffith, Introduction to Electrodynamics, 2014, 4th Edition, Pearson.		
4.	Djafar K. Mynbaev and Lowell L. Scheiner, Fiber Optic Communication Technology, 2011, Pearson		
Reference Books			
1.	Raymond A. Serway, Clement J. Mosses, Curt A. Moyer Modern Physics, 2010, 3rd Indian Edition Cengage learning.		
2.	John R. Taylor, Chris D. Zafiratos and Michael A. Dubson, Modern Physics for Scientists and Engineers, 2011, PHI Learning Private Ltd.		
3.	Kenneth Krane Modern Physics, 2010, Wiley Indian Edition.		
4.	Nityanand Choudhary and Richa Verma, Laser Systems and Applications, 2011, PHI Learning Private Ltd.		
5.	S. Nagabhushana and B. Sathyanarayana, Lasers and Optical Instrumentation, 2010, I.K. International Publishing House Pvt. Ltd.,		
6.	R. Shevgaonkar, Electromagnetic Waves, 2005, 1st Edition, Tata McGraw Hill		
7.	Principles of Electromagnetics, Matthew N.O. Sadiku, 2010, Fourth Edition, Oxford.		
8.	Ajoy Ghatak and K. Thyagarajan, Introduction to Fiber Optics, 2010, Cambridge University Press.		
Mode of Evaluation: CAT / Assignment / Quiz / FAT / Project / Seminar			
List of Experiments			
1.	Determination of Planck's constant using electroluminescence process	2 hrs	
2.	Electron diffraction	2 hrs	
3.	Determination of wavelength of laser source (He-Ne laser and diode lasers of different wavelengths) using diffraction technique	2 hrs	
4.	Determination of size of fine particle using laser diffraction	2 hrs	
5.	Determination of the track width (periodicity) in a written CD	2 hrs	
6.	Optical Fiber communication (source + optical fiber + detector)	2 hrs	
7.	Analysis of crystallite size and strain in a nano-crystalline film using X-ray diffraction	2 hrs	
8.	Numerical solutions of Schrödinger equation (e.g. particle in a box problem) (can be given as an assignment)	2 hrs	
9.	Laser coherence length measurement	2 hrs	
10.	Proof for transverse nature of E.M. waves	2 hrs	
11.	Quantum confinement and Heisenberg's uncertainty principle	2 hrs	
12.	Determination of angle of prism and refractive index for various colour – Spectrometer	2 hrs	
13.	Determination of divergence of a laser beam	2 hrs	
14.	Determination of crystalline size for nanomaterial (Computer simulation)	2 hrs	
15.	Demonstration of phase velocity and group velocity (Computer simulation)	2 hrs	
Total Laboratory Hours			30 hrs
Mode of evaluation: CAT / FAT			
Recommended by Board of Studies		04-06-2019	
Approved by Academic Council		No. 55	Date 13-06-2019

Course code	Course title	L	T	P	J	C
PHY1999	Introduction to Innovative Projects	1	0	0	0	1
Pre-requisite	Nil	Syllabus version				
						1.0
Course Objectives:						
1. To make students confident enough to handle the day to day issues. 2. To develop the “Thinking Skill” of the students, especially Creative Thinking Skills 3. To train the students to be innovative in all their activities 4. To prepare a project report on a socially relevant theme as a solution to the existing issues						
Course Outcomes:						
1. Comprehend the various types of thinking skills. 2. Explain the innovative and creative ideas. 3. Analyze a suitable solution for socially relevant issues						
Module:1 A Self Confidence						
						1 hour
Understanding self– Johari Window–SWOT Analysis– Self Esteem– Being a contributor – Case Study Project : Exploring self, understanding surrounding, thinking about how s(he) can be a contributor for the society, Creating a big picture of being an innovator– writing a 1000 words imaginary autobiography of self– Topic “Mr X– the great innovator of 2015” and upload. (4 non- contact hours)						
Module:1 B Thinking Skill						
						1 hour
Thinking and Behaviour– Types of thinking– Concrete– Abstract, Convergent, Divergent, Creative, Analytical, Sequential and Holistic thinking– Chunking Triangle– Context Grid – Examples – Case Study. Project : Meeting at least 50 people belonging to various strata of life and talk to them / make field visits to identify a min of 100 society related issues, problems for which they need solutions and categories them and upload along with details of people met and lessons learnt. (4 non- contact hours)						
Module:1 C Lateral Thinking Skill						
						1 hour
Blooms Taxonomy– HOTS– Out of the box thinking– deBono lateral thinking model – Examples Project : Last weeks -incomplete portion to be done and uploaded						
Module:2 A Creativity						
						1 hour
Creativity Models– Walla– Barrons– Koberg & Begnall– Examples Project : Selecting 5 out of 100 issues identified for future work. Criteria based approach for prioritisation, use of statistical tools & upload. (4 non- contact hours)						
Module:2 B Brainstorming						
						1 hour
25 brainstorming techniques and examples Project : Brainstorm and come out with as many solutions as possible for the top 5 issues identified & upload. (4 non- contact hours)						
Module:3 Mind Mapping						
						1 hour
Mind Mapping techniques and guidelines. Drawing a mind map Project : Using Mind Maps get another set of solutions for the next 5 issues (issue 6–10). (4 non- contact hours)						
Module:4 A System thinking						
						1 hour
Systems Thinking essentials– examples– Counter Intuitive condemnns Project : Select 1 issue / problem for which the possible solutions are available with you. Apply Systems Thinking process and pick up one solution [explanations should be given why the other possible solutions have been left out]. Go back to the customer and assess the acceptability and upload.. (4 non- contact hours)						
Module:4 B Design Thinking						
						1 hour
Design thinking process– Humanelement of design thinking– case study Project : Apply design thinking to these selected solution, apply the engineering & scientificing to it. Participate in “design week” celebrations upload the weeks learning						

out come.			
Module:5 A	Innovation	1 hour	
DifferencebetweenCreativityandInnovation–Examples ofinnovation–Being innovative. Project: Aliteraturesearchesonprototypingofyoursolution finalized.Prepare aprototypemodel or processand upload. . (4 non- contact hours)			
Module:5 B	BlocksforInnovation	1 hour	
IdentifyBlocksforcreativityandinnovation – overcomingobstacles – Case Study Project : Project presentation on problem identification,solution,innovations-expectedresults– InterimreviewwithPPTpresentation. . (4 non- contact hours)			
Module:5 C	InnovationProcess	1 hour	
StepsforInnovation–rightclimateforinnovation Project: Refiningtheproject,basedonthereviewreportanduploading the text. . (4 non- contact hours)			
Module:6 A	Innovation in India	1 hour	
Storiesof10 Indian innovations Project: Makingthe project better with add ons. . (4 non- contact hours)			
Module:6 B	JUGAAD Innovation	1 hour	
Frugal andflexible approach toinnovation -doing more with less Indian Examples Project: FinetuningtheinnovationprojectwithJUGAADprinciples and uploading (Credit for JUGAADimplementation). (4 non- contact hours)			
Module:7 A	Innovation Project Proposal Presentation	1 hour	
Projectproposal contents, economicinput, ROI– Template Project: Presentationoftheinnovativeprojectproposalandupload. (4 non- contact hours)			
Module:8 A	Contemporary issuein Innovation	1 hour	
Contemporary issuein Innovation Project: Final project Presentation , Vivavoce Exam (4 non- contact hours)			
		Total Lecture hours:	15 hours
Text Book(s)			
1.	How to have Creative Ideas, Edward deBono, Vermilionpublication, UK, 2007		
2.	The Artof Innovation, Tom Kelley &Jonathan Littman, ProfileBooks Ltd, UK, 2008		
Reference Books			
1.	Creating Confidence, MeribethBonct, Kogan Page India Ltd, New Delhi, 2000		
2.	Lateral Thinking Skills, Paul Sloane, KeoganPage India Ltd, NewDelhi, 2008		
3.	Indian Innovators, Akhat Agrawal, JaicoBooks, Mumbai, 2015		
4.	JUGAAD Innovation, Navi Radjou, Jaideep Prabhu, Simone Ahuja Random house India, Noida, 2012.		
Mode of Evaluation: CAT / Assignment / Quiz / FAT / Project / Seminar Threereviews with weightageof25 :25 : 50 alongwithreports			
Recommended by Board of Studies		15-12-2015	
Approved by Academic Council		No. 39	Date 17-12-2015

PROGRAMME CORE

Course code	CHE1001	L	T	P	J	C
Course title	MATERIAL SCIENCE AND STRENGTH OF MATERIALS	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. To understand the concept of mechanical behaviour of materials, stress - strain and their use in analysis and design of machine members and structures. 2. To learn the distributed force systems, centroid/centre of gravity and method of finding centroids of composite figures and bodies 3. To study the moment of inertia and method of finding moment of inertia of areas and bodies, bending of beams under different loading conditions 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand concept of materials and their engineering properties 2. Analyse the behaviour of structural and machine components subjected to various loading and support conditions based on principles of equilibrium 3. Solve problems using the concept of centroid/centre of gravity 4. Evaluate moment of inertia of areas and bodies 5. Apply the concept of stress and strain to analyze structural members and machine parts under various loading condition 6. Analyze the stresses developed in cylindrical and spherical shell 						
Module:1	Engineering Metallurgy	6 hours				
Properties of materials: Mechanical, Physical & Chemical properties, Industrial Engineering Materials – Ferrous & Non Ferrous metals & alloys; Introduction to various heat treatment processes & Mechanical tests.						
Module:2	Response of materials (Regular Geometry)	6 hours				
Introduction to elasticity – Stress & Strain – Types of stresses & strain – Stress strain curve and relationship – Hooke's law – Modulus of Elasticity & Modulus of Rigidity– Deformation of a body due to force acting on it – Deformation of a body due to self-weight.						
Module:3	Response of materials (Irregular Geometry)	8 hours				
Principle of Superposition – Stress & Strain analysis in bars of varying sections – Stresses in bars of uniformly tapering section.						
Module:4	Centroid	6 hours				
Introduction to Centroid & Centre of Gravity – Methods of Centroid – Centroid of plane figures by geometrical consideration. Centre of Gravity (real bodies): Centre of gravity by method of moments for symmetrical & unsymmetrical lamina – Centre of gravity for solids and cut sections.						
Module:5	Moment of Inertia	6 hours				
Concept of Moment of Inertia & Methods for Moment of Inertia – Moment of Inertia for Rectangular sections – Theory of Parallel axis – Moment of Inertia for Triangular, Circular and Semi-circular sections.						

Module:6	Transverse loading on Beams	6 hours
Introduction to Beams – Types of Loading – Shear force and Bending Moments – Sign conventions – SFD & BMD for Cantilever beams and Simply supported beams with point loads, UDL and UVL.		
Module:7	Thin and Thick Pressure vessels	5 hours
Introduction – Pressure vessels; Stresses in thin and thick cylindrical shell due to internal pressure – Circumferential and longitudinal stresses – Spherical shells subjected to internal pressure.		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	M. F. Ashby, D. R. H. Jones, Engineering Materials - An Introduction to their Properties and Applications. 2 nd ed., Butterworth Heinemann, 2011	
2.	S. Timoshenko, D.H. Young (Author), Strength of Materials: Advanced theory and problems, 4 th ed., CBS Publishers & Distributors, 2013	
Reference Books		
1.	N.M. Belayavev, Problems in Strength of Materials, Pergamon Press, 2013.	
2.	W. A. Nash, Strength of Materials, Schaum's Outline Series, Revised 4 th ed., McGraw Hill, 2010.	
3.	Beer, Tata McGraw Hill ,(in SI Units)Mechanics Of Materials ,Dewolf &Johnston 2004 ,Publications	
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by AcademicCouncil	55 th	13-06-2019

Course code	CHE1002	L	T	P	J	C
Course title	PROCESS CALCULATIONS	4	0	0	0	4
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
1. Formulate material balances to solve for compositions and flow rates of process streams						
2. Incorporate single and multiple reactions into unit operations within chemical processes						
3. Perform material and energy balance calculations in various systems						
Course Outcomes (CO):						
1. Apply mole concept and ideal gas equation to express the composition of mixtures						
2. Apply the concept of humidity and use psychrometric chart for air-water vapor systems						
3. Solve steady state material balances without chemical reactions						
4. Estimate the extent of reaction in material balances for systems involving chemical reactions						
5. Analyze the processes involving recycling and bypass involving chemical reactions.						
6. Apply simultaneous material & energy balances to industrial processes.						
Module:1	Basic Chemical Calculations					8 hours
Units and dimensions – Conversion factors – Mole concept – Concept of normality, molarity, and molality – Density and specific gravity – Methods of expressing composition of mixtures and solutions – Weight fraction – Mole fraction – Volumetric composition – Ideal gas law – Dalton's law – Amagat's law						
Module:2	Vapor pressure and Humidity calculations					6 hours
Vapor pressure and liquids – Antoine equation, Vapor pressure of immiscible liquids and ideal solutions – Raoult's law – Humidity and Saturation – Relative and percentage saturation, Wet bulb and dry bulb temperature, Dew point – Use of humidity chart for engineering calculations						
Module:3	Material Balance without Chemical Reaction					12 hours
Law of conservation of mass – Process flowsheet – Material balance calculations involving drying, dissolution, distillation, crystallization, evaporation, absorption and extraction						
Module:4	Material balance with Chemical Reaction					7 hours
Stoichiometric equation – stoichiometric ratio – limiting reactant – excess reactant – percent excess – conversion – yield						
Module:5	Recycle and Bypass Operation					7 hours
Recycle, Purge, Bypass calculations in operations such as evaporation, distillation, and drying						
Module:6	Combustion calculations					10 hours
Calorific value of fuels, Flue gas analysis, Orsat analysis, theoretical and excess air requirement for solid, liquid and gaseous fuels						
Module:7	Energy balance					8 hours

Standard heat of formation–Standard heat of combustion–Standard Heat of reaction–Hess’s law– Determination of heat of reaction at temperatures other than standard temperature using specific heat relationships– Calculation of theoretical flame temperature			
Module:8	Contemporary issues		2 hours
Total Lecture hours			60 hours
Text Books			
1.	Himmelblau D.M., Basic Principles and Calculations in Chemical Engineering, 8 th ed., Prentice Hall, India, 2012.		
2.	Bhatt B.I., Thakore S.B., Stoichiometry, 5 th ed., Tata McGraw–Hill Book Company, New Delhi, 2011.		
Reference Books			
1.	Felder R., Rousseau R., Elementary Principles of Chemical Processes, 3 rd ed., John Wiley & Sons, 2000.		
2.	Narayanan K.V., Lakshmi kutty B., Stoichiometry and Process calculations, Prentice Hall India Limited, New Delhi, 2006.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1003	L	T	P	J	C
Course title	PROCESSENGINEERING THERMODYNAMICS	3	0	0	4	4
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Enhance the basic knowledge and intuitive understanding of thermodynamics on the physical and chemical system 2. Introduce the concepts of partial molar properties, fugacity, activity, vapour-liquid equilibrium for ideal and real substances existing in more than one phase under equilibrium 3. Generalize the design thinking skills on property estimation to chemical industries 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Analyze thermodynamic equilibrium for ideal and nonideal systems 2. Evaluate the thermodynamic properties for pure substances 3. Estimate the partial molar properties of solutions 4. Construct and analyze the phase equilibrium data, P-x-y, T-x-y diagram for ideal binary miscible vapour-liquid systems. 5. Analyse VLE for non-ideal miscible systems using property models 6. Determine the equilibrium rate constant for chemical reactions. 						
Module:1	Fundamental concepts and definitions	5 hours				
Introduction-Definition and Basic Concepts-classical and statistical thermodynamics-Concept of Continuum-Thermodynamic steady state-equilibrium state process, Volumetric properties of pure fluids: PVT Relations- Ideal gas- Real gas-						
Module:2	Laws of thermodynamics	6 hours				
First law analysis- Closed non-flow system- Steady state flow systems and their analysis; Second law of thermodynamics-change in internal energy-enthalpy-entropy calculation for process-phase change; Heat effects - standard heat of reaction						
Module:3	Thermodynamic properties of pure fluids	7 hours				
Gibbs free energy-Helmholtz free energy-exact differential equation-thermodynamic property relations-Maxwell's relations and applications-fugacity-activity of pure substances-determination of fugacity of pure gases, solids and liquids						
Module:4	Thermodynamic properties of solution	7 hours				
Mixture of pure fluids-Partial molar properties-Chemical potential-fugacities in solution; Ideal solutions-Lewis-Randall rule-Raoult's law-Henry's law; Gibbs-Duhem equation; Residual properties-Property changes of mixing for ideal- non-ideal solutions- Excess properties relations and Gibbs free energy calculation						
Module:5	Phase equilibria	6 hours				
Phase rule-criteria of phase equilibrium-single component-multiple components; Vapor-Liquid Equilibria for ideal solutions-Phase diagram for binary systems using ASPEN PLUS-constant temperature equilibria- constant pressure equilibria- phase						
Module:6	Vapor liquid equilibria- non-ideal solutions	7 hours				

Nonideal solution–Azeotropic systems–minimum boiling–maximum boiling–VLE– P-x-y diagram and T-x-y diagram using ASPEN PLUS; Bubble point–Dew Point–calculation methods– Van Laar equation - Margules equation - Wilson equation. Multicomponent Systems– flash vaporization; Consistency Test for VLE Data			
Module:7		Chemical reaction equilibria	
		5 hours	
Chemical reaction equilibria–Reaction coordinates– criteria for chemical equilibrium, equilibrium constant–Gibbs Free Energy of the reaction– effect of temperature on equilibrium constant– equilibrium constant of homogeneous gas and liquid			
Module:8		Contemporary issues	
		2 hours	
Total Lecture hours			45 hours
Text Books			
1.	Narayanan K.V., A Textbook of Chemical Engineering Thermodynamics, 2 nd ed., Prentice Hall India, New Delhi, 2012		
2.	Ahuja P., Chemical Engineering Thermodynamics, 2 nd ed., PHI Learning Pvt. Ltd., New Delhi, 2012.		
Reference Books			
1.	Smith J.M., Van Ness H.C., Abbott M.M., Introduction to Chemical Engineering Thermodynamics, 8 th ed., McGraw-Hill, New York, 2018.		
2.	Rao Y.V.C., Chemical Engineering Thermodynamics, 1 st ed., University Press, New Delhi, 2005.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1004	L	T	P	J	C
Course title	CHEMICAL TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Introduce the basic information and the systematic diagrams of Unit operations involved in chemical industries. 2. Familiarize the concepts of design, operation details and schematic of industrial equipment. 3. Ascertain the right separation technology for easy separation of chemical components 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Illustrate the major unit operations and processes involved in manufacturing industries 2. Analyze the manufacturing processes in organic and inorganic chemical industries 3. Explain the different industrial gases involved in chemical industries 4. Analyze the manufacturing processes of fertilizers industries 5. Explain the process flow sheet and end uses of cellulosic material in different application 6. Analyze the manufacturing processes of petroleum refinery and petrochemical products 						
Module:1	Chloro-alkali and Cement Industries	6 hours				
Manufacture of soda ash; caustic soda – manufacture of calcium hypochlorite; manufacture of sulphur and sulphuric acid; manufacture of Portland cement; manufacture of glass						
Module:2	Industrial Gases	5 hours				
Manufacture of carbon-di-oxide; hydrogen; oxygen and nitrogen; acetylene; water gas; producer gas and manufacture of natural gas						
Module:3	Fertilizer Industries	8 hours				
Manufacture of nitric acid and urea; manufacture of phosphorus and phosphoric acid; manufacture of super phosphate and triple super phosphate; manufacture of potassium chloride						
Module:4	Cellulose, Sugar and Oil Production Industries	7 hours				
Production of pulp – manufacture of paper and manufacture of viscous rayon; manufacture of sugar and starch; refining of edible oils and fats; manufacture of soaps and detergents; bio-degradability of surfactants						
Module:5	Petroleum Industries	6 hours				
Petroleum refining processes; reforming; cracking; secondary refining processes						
Module:6	Petrochemical Industries	6 hours				
Introduction to Petrochemical processes; Manufacture of C ₂ , C ₃ , C ₄ chemical compounds						
Module:7	Polymer Industries	5 hours				
Introduction; manufacture of nylon 6; nylon 6,6; manufacture of silicones; manufacture of urea formaldehyde; manufacture of phenol formaldehyde						
Module:8	Contemporary issues	2 hours				

Total Lecture hours		45 hours	
Text Books			
1	Rao G., Sittig M., Dryden's Outlines of Chemical Technology, 3 rd ed., East West Press, India, 2010.		
2	Austin G.T., Shreve's Chemical Process Industries, 5 th ed., McGraw Hill, USA, 2012.		
Reference Books			
1	Matar S., Hatch L.F., Chemistry of Petrochemical Processes, 4 th ed., Gulf Publishing, USA, 2005.		
2	Nelson W.L., Petroleum Refinery Engineering, 4 th ed., McGraw Hill, USA, 2005.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Course code	CHE1005	L	T	P	J	C
Course title	MOMENTUMTRANSFER	3	0	2	0	4
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand the fluid properties, the fundamental principles and theorems related to momentum transfer 2. Apply the physical and mathematical models to analyse the fluid flow phenomena in engineering applications 3. Solve the steady state and un-steady state momentum transfer problems 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Explain the properties of Newtonian and Non-Newtonian fluid and basic principles of momentum transfer 2. Develop the governing equations related to momentum transfer phenomena 3. Analyze the performance of flow measuring devices and fluid moving machineries 4. Solve the problems related to the losses incurred during fluid flow 5. Predict the fluid flow behaviour using non-dimensional numbers 6. Analyze the fluid flow phenomena through packed and fluidized beds 						
Module:1	Basic Concept of Momentum Transfer					5 hours
Introduction and Significance of Momentum Transfer in Chemical Engineering. Definition of fluid- Classification of fluids- Newtonian fluid- Characteristic properties of fluids- Non-Newtonian Fluids and their classification. Fluid statics: Pascal's law and Hydrostatic law of equilibrium; Pressure and its measurement- Manometers						
Module:2	Concept of Fluid Flow Phenomena					7 hours
Kinematics of fluid flow, Dynamics of fluid flow- Basic equations governing fluid flow- types of fluid flow. Equation of Continuity and its application, Equation of motion- Derivation of Euler's equation, Bernoulli's equation and its application in fluid flow						
Module:3	Flow Measuring Devices					5 hours
Importance of metering- Classification flow measuring devices, Principle and working of Orificemeter, Venturi meter, Pitot tube, Variable area meters: Rotameter						
Module:4	Flow through Circular Pipes					8 hours
Flow of fluids in Laminar regime- Velocity Profile, Shear Stress Distribution- Hagen- Poiseuille equation- Concept of average velocity- Concept of Kinetic energy correction factor, Concept of Fluid friction- Skin friction- Form friction- Factors affecting friction- Friction factor- Application of Moody's diagram, Minor losses and						
Module:5	Dimensional Analysis					4 hours
Dimensional homogeneity- Raleigh and Buckingham π theorems- Non-dimensional numbers, model laws						
Module:6	Fluid Flow through Packed and Fluidized Bed					7 hours

Flow past immersed bodies–Significance of form friction–Concept of Drag, Drag Coefficients and Particle Reynolds number - Drag Coefficient vs. Particle Reynolds number curves for regular and irregular shaped solid particles. Flow of fluid through packed beds–Packing and types of packing–Pressure drop across packed beds–Kozeny Carman equation–Ergun’s equation–Loading and Flooding Packed Beds. Concept of Fluidization–Condition for Solid particle to be in a suspended condition in a flowing fluid– minimum fluidization velocity			
Module:7		Transportation of Fluids	
		7 hours	
Transportation Components–Pipe, Fittings and Valves, Types of Fittings, valves–Stuffing Boxes, Mechanical Seals–Estimation of head loss from fittings and valves, Concept of minor losses–types of minor losses. Fluid Moving Machinery: Pumps–Classification and working of Centrifugal Pumps and Positive Displacement Pumps Basic Principles of Centrifugal Pumps–Pump Characteristics–Concept of Specific Speed, Net Positive Suction Head–			
Module:8		Contemporary issues	
		2 hours	
Total Lecture hours			45 hours
Text Books			
1.	Fox R. W., McDonald A. T., Pirtchard P. J., Mitchell J. W., Introduction to Fluid Mechanics, 9 th ed., Wiley Publications, 2015.		
2.	Cengel Y. A., Cimbala J. M., Fluid Mechanics (SIE): Fundamentals and Applications, 3 rd ed., McGraw Hill, New York, 2014.		
Reference Books			
1.	McCabe, Smith, Harriott, Unit Operations of Chemical Engineering 7 th ed., McGraw Hill, USA, 2014.		
2.	Som S. K., Biswas G., Chakraborty S., Introduction to Fluid Mechanics and Fluid Machines, 3 rd ed., Tata McGraw Hill, India, 2011.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Laboratory Experiments			
1.	Determination of coefficient of discharge of venturimeter		2 hours
2.	Calibration of an orifice meter		2 hours
3.	Determination of friction factor for flow through circular pipe		2 hours
4.	Determination of loss of coefficient due to sudden enlargement, sudden contraction, bend and elbow		2 hours
5.	Determination of Reynolds apparatus		2 hours
6.	Verification of Bernoulli’s theorem		2 hours
7.	Performance characteristics of centrifugal pump at rated speed		2 hours
8.	Determination of pressure drop per unit length as a function of superficial velocity of fluidization medium		2 hours
9.	Verification of relationship between fluid flow and pressure drop per unit length of packing		2 hours
10.	Determination of friction factor for flow through noncircular pipe		2 hours
Total Laboratory Hours			20 hours
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1006	L	T	P	J	C
Course title	HEAT TRANSFER	2	0	2	4	4
Pre-requisite	MAT2002	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Explain the fundamental principles of heat transfer and various modes of heat transfer 2. Solve heat transfer problems using the principles of heat transfer in different modes 3. Design and estimate heat loads for heat transfer equipments such as heat exchangers and evaporators 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Classify the different modes of heat transfer with their significance 2. Analyze steady/unsteady state conductive heat transfer problems 3. Analyze the convective heat transfer mechanism in fluids with and without phase change 4. Examine radioactive heat transfer with and without radiation shields 5. Evaluate the performance of various types of heat exchangers 6. Calculate the heat transfer rate and surface area of evaporators 						
Module:1	Conduction					5 hours
Basic concepts – Conduction –Fourier’s Law of Heat conduction–Concept of Thermal Conductivity– Generalized conduction equation in cartesian, cylindrical and spherical systems; Steady State Conduction –Heat transfer composite systems–Critical thickness of insulation–Conduction with heat Generation						
Module:2	Extended Surfaces and Unsteady state conduction					3 hours
Extended surfaces– types and applications of fins– Fin efficiency and effectiveness– Fin performance; Unsteady state heat conduction– Lumped parameter system– Conduction through Semi-Infinite Solids						
Module:3	Convection (without phase change)					5 hours
Fundamentals of Convection– Thermal boundary layer & Convective heat transfer coefficients– Convection correlation through Dimensional analysis; Laminar flow over a flat plate– Turbulent flow over a flat plate– Flow over cylinders– Internal flow through pipes– annular spaces– Natural convection in vertical - inclined and horizontal surfaces.						
Module:4	Convection (with phase change)					3 hours
Condensation and Boiling– Dropwise and Film type Condensation– Film condensation on a vertical plate; Boiling– Nucleate boiling and film boiling correlations– Critical flux						
Module:5	Radiation					3 hours
Radiation heat transfer– Thermal radiation – Laws of radiation – Black body concepts– Emissive power– Radiation shape factor– Gray bodies – Radiation shields						
Module:6	Heat Exchangers					5 hours
Heat exchangers– Types and practical application– Concept of LMTD & Overall heat transfer coefficient; Effectiveness– NTU method for heat exchanger design; Fouling factor and estimation of Overall heat transfer						

Module:7	Evaporators	4 hours
Introduction–TypesofEvaporators–Capacity–Steameconomy–Boilingpointelevation(Duhrringrule);Materialandenergybalanceofsingleeffectevaporator;Theoryofmultipleeffectevaporators;Design of singleand multiple effectevaporators, Vapor		
Module:8	Contemporary issues	2 hours
Total Lecture hours		30 hours
Text Books		
1.	GhajarA.J.,CengelY.A.,Heat andMass Transfer:A Practical Approach, 5 th ed.,McGraw-Hill,USA,2014.	
2.	Holman J.P, Heat Transfer, 10 th ed.,McGraw-Hill Series, USA, 2010.	
Reference Books		
1.	FrankKreith,RajMManglik,PrinciplesofHeatTransfer,8 th ed.,CengageLearning,USA,2016.	
2.	Frank.P.Incropera,DavidP.Dewitt,FundamentalsofHeat&MassTransfer,6 th ed.,JohnWiley& Sons, USA, 2010.	
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
LaboratoryExperiments		
1.	Measurement of thermal conductivity of Metals&insulators	2 hours
2.	AnalysisofTransientHeat Conduction	2 hours
3.	Performance of NaturalConvection	2 hours
4.	AnalysisofFin efficiency&effectiveness	2 hours
5.	Emissivitymeasurement	2 hours
6.	Performance of Double Pipe Heat Exchanger	2 hours
7.	Performance of Agitated Vessel	2 hours
8.	Performance of PlatetypeHeat Exchanger	2 hours
9.	Performance of Heat Transfer inpacked bed	2 hours
10.	Performance of Cooling tower	2 hours
Total Laboratory Hours		20 hours
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
Recommended by Board of Studies		15-04-2019
Approved by AcademicCouncil		55 th Date 13-06-2019

Course code	CHE1022	L	T	P	J	C
Course title	MECHANICAL OPERATIONS	3	0	2	0	4
Pre-requisite	Nil	Syllabus version				
		1.2				
Course Objectives:						
1. Introduce the basic information and the systematic diagrams of Unit operations involved in chemical industries 2. Learn the concepts of design, operation details and schematic of industrial equipment 3. Choose the right separation technology for easy separation of chemical components						
Course Outcomes (CO):						
1. Understand the basic principles involved in different unit operations 2. Calculate the size distribution of particulate matter 3. Analyze the performance of various size reduction equipment 4. Identify the suitable separation technique based on particle characteristics 5. Estimate the filtration and sedimentation parameters involved in solid – liquid separation 6. Design an agitation vessel based on standard design criterion						
Module:1	Introduction to Particulate Solids	4 hours				
Particle Shape, Size, Mixed Particle Sizes and Size Analysis – Cumulative and Differential Analysis – Various Mean Diameters – Screen Analysis Standard Screens – Various Industrial Screens						
Module:2	Particle Separation	3 hours				
Introduction to Particle Separation – Electrostatic Precipitation and Magnetic Separation – Storage of Solids						
Module:3	Size Reduction	8 hours				
Size Reduction – Principles of Comminution – Energy and Power Requirements in Comminution – Mechanical Efficiency – Laws of Crushing – Size Reduction Equipment – Crushers – Grinders – Cutting Machines – Open and Closed Circuit Operation						
Module:4	Particulate Solids Flow	5 hours				
Motion of a Particle through a Fluid – Terminal Velocity – Free and Hindered Settling. Classification: Separations Ratio – Classification Equipment – Gravity Settling Tank – Elutriator – Cone Classifiers – Bowl Classifier – Centrifugal Classifier – Cyclone						
Module:5	Hydro-Mechanical Separations	7 hours				
Sedimentation: Gravity Sedimentation – Mechanism – Continuous Sedimentation – Thickener – Design of thickener – Classifier and Clarifier – Settling Area – Centrifugal Sedimentation – Centrifuges – Hydro clones. Flootation: Equipment – Modifiers – Collectors – Frothing Agents						
Module:6	Filtration	8 hours				
Filtration – Filter Media – Filter Aids – Principles of Cake Filtration – Constant Pressure Filtration – Constant Rate Filtration – Pressure Drop Through Filter Cake – Compressible and Incompressible Filter Cakes – Specific Cake Resistance – Filter Medium Resistance. Filtration Equipment – Filter Presses – Leaf Filter – Rotary Continuous Filters. Principles of Centrifugal Filtration – Washing of Filter Cakes						

Module:7	Agitation and Mixing	8 hours
Agitation and Mixing of Liquids – Principles of Agitation – Agitation Equipment – Impellers – Flow Pattern in Agitated Vessel – Power Consumption in Agitated vessel. Flow number – Power Correlation – Calculation of power consumption. Blending and mixing – Jet mixers – Motionless Mixers. Mixing of Solids: Mixtures for Cohesive solids – Power requirements Criteria for mixer effectiveness. Mixers for free flowing granular solids - Rate of		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	McCabe W., Smith J., Harriott P., Unit Operations of Chemical Engineering, 7 th ed., McGraw Hill Education; USA, 2014.	
Reference Books		
1.	Coulson J.M., Richardson J.F., Chemical Engineering, Volume 2 (Particle Technology & Separation Processes), 5 th ed., Butterworth – Heinemann Publishing Ltd., USA, 2001.	
2.	Narayanan C.M., Bhattacharya B.C., Mechanical Operations for Chemical Engineers, 3 rd ed., Khanna Publishers, India, 2011.	
3.	Patil K.D., Mechanical Operations (Fundamental Principles and Applications), 3 rd ed., Nirali Prakasam, India, 2012.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Laboratory Experiments		
1.	Performance of Plate and Frame filter press	2 hours
2.	Performance of Rotary Drum Filter	2 hours
3.	Performance of Leaf Filter	2 hours
4.	Analysis of Jaw crusher parameters	2 hours
5.	Analysis of Roll crusher parameters	2 hours
6.	Analysis of Ball mill parameters	2 hours
7.	Sieve analysis	2 hours
8.	Measurement of Drag	2 hours
9.	Batch sedimentation performance	2 hours
10.	Beaker decantation analysis	2 hours
Total Laboratory Hours		20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by Academic Council	55 th	Date 13-06-2019

Course code	CHE2001	L	T	P	J	C
Course title	CHEMICAL REACTION ENGINEERING	3	0	2	0	4
Pre-requisite	CHE1003	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Impart the knowledge of calculus, differential equations, thermodynamics, general chemistry, and material and energy balances to solve reactor design problems. 2. Simulate several types of reactors in order to choose the most appropriate reactor for a given need 3. Examine the problems related to multiple reactions and evaluate the selectivity, reactivity and yield 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Classify various types of chemical reactions 2. Analyze the batch reactor data using rate equations and reaction kinetics 3. Design ideal reactors for elementary chemical reactions 4. Evaluate the multiple reactor systems and multiple reactions 5. Evaluate the performance of non-isothermal reactors 6. Analyze non-ideal behavior of reactors using RTD studies 						
Module:1	Fundamental Concepts and Definitions	5 hours				
Classification of reactions- Rate and stoichiometry-rate law- rate equation-rate constant-variables affecting the rate of reaction-activation energy-reactions at equilibrium						
Module:2	Chemical Kinetics	6 hours				
Interpretation of Batch Reactor Data-Constant Volume Batch Reactor and variable volume batch reactor; Integral method-Differential method of analysis for reactions-reaction mechanism; Method of half-life; Analysis of data for Reversible and Irreversible Reactions						
Module:3	Isothermal Ideal Reactor Design of Single and Multiple reactions	7 hours				
Ideal Batch Reactor-space time-holding time and space velocity; Ideal Mixed Flow Reactor-Ideal Plug Flow Reactor for single reactions-Size comparison of single Reactors for single reactions-Semi batch reactor - Recycle reactor-Auto catalytic reactor						
Module:4	Multiple Reactors	6 hours				
Multiple Reactor Systems-equal size mixed flow reactors in series-plug flow reactors in series and or in parallel-mixed flow reactors of different sizes in series-reactors of different types in series						
Module:5	Design for Multiple Reactions	6 hours				
Reactions in parallel (simultaneous reactions) for CSTR-PFR-reactions in series (Consecutive Reactions) for CSTR-PFR-Combined series and parallel reactions						
Module:6	Non-isothermal Reactors	6 hours				
Steady state non-isothermal reactors-CSTR-PFR-Mole balance-Energy balance-Adiabatic reactors -CSTR-PFR-Batch reactor-Multiple steady states-Multiple chemical reactions						
Module:7	Non Ideal Reactors	7 hours				
Basics of non-ideal flow - Measurement of residence time distribution (RTD) - Relationship						

between C, E and F curves - Modelling of non-ideal reactors - one parameter and two parameter models - Conversion in real reactor systems			
Module:8	Contemporary issues		2 hours
Total Lecture hours			45 hours
Text Books			
1.	Levenspiel O., Chemical Reaction Engineering, 3 rd ed., Wiley Publications, USA, 2006		
2.	Fogler H.S., Elements of Chemical Reaction Engineering, 5 th ed., Prentice Hall India Pvt. Ltd., India, 2016		
Reference Books			
1.	Froment G. F, Bischoff K.B, Wilde J.D., Chemical Reactor Analysis and Design, 1 st ed., Wiley Publications, USA, 2010		
2.	Smith J.M., Chemical Engineering Kinetics, 8 th ed., McGraw-Hill, USA, 2008		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Laboratory Experiments			
1.	Analysis of Batch reactor – equimolar constant volume system		2 hours
2.	Analysis of Temperature dependency of reaction rate		2 hours
3.	Analysis of Semi batch reactor		2 hours
4.	Assessment of Adiabatic batch reactor performance		2 hours
5.	Analysis of Mixed flow reactor		2 hours
6.	Analysis of Plug flow reactor analysis		2 hours
7.	Analysis of combined reactor system		2 hours
8.	Analysis of Packed bed reactor		2 hours
9.	Analysis of RTD studies in Plug flow reactor		2 hours
10.	Analysis of RTD studies in Mixed flow reactor		2 hours
Total Laboratory Hours			20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE2002	L	T	P	J	C
Course title	PROCESS EQUIPMENT DESIGN ANDECONOMICS	2	0	2	4	4
Pre-requisite	CHE1006	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Summarize the concepts of unit operations and unit processes in chemical engineering. 2. Impart knowledge on the concepts of design of major equipment 3. Understand the economics and feasibility analysis of the process industry 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Examine the process flow diagrams used in chemical and allied industries 2. Analyze the design of fluid handling equipment 3. Evaluate the performance of Heat, Mass transfer equipment and reactors 4. Develop Heat Exchanger Network using Pinch Technology 5. Apply economic principles for cost estimation in chemical industries. 6. Evaluate the performance of process equipment using simulation tools 						
Module:1	Introduction and Pressure vessel	4 hours				
Introduction-Types of flow chart preparation; Fluid handling equipment; Mechanical design of pressure Vessel.						
Module:2	Heat transfer equipment	5 hours				
Design of Double pipe, shell and tube heat exchanger; Principles of dryer design.						
Module:3	Heat Exchanger Network	4 hours				
Introduction to Pinch Technology – pinch point– Composite and Grand Composite curves; Find Heat exchanger network for simple processes.						
Module:4	Separation process equipment	4 hours				
Design of Distillation column and absorbers– plate type and packed columns.						
Module:5	Principles and Design of Reactors	4 hours				
Concepts of ideal reactor design– adiabatic and catalytic reactors						
Module:6	Cost Estimation of Projects	4 hours				
Cost estimation of Chemical Projects; Cost estimation of individual equipment using algorithms and literature.						
Module:7	Analysis of Cost Estimation	4 hours				
Time value of money; Depreciation; Profitability analysis; Analysis of alternatives and replacements using cost diagrams.						
Module:8	Contemporary issues	1 hour				

Total Lecture hours		30 hours
Text Books		
1.	Peters M., Timmerhaus K., West R., Plant Design and Economics for Chemical Engineers, 5 th ed., McGraw Hill, USA, 2017.	
2.	Kemp I.C., Pinch Analysis and Process Integration: A User Guide on Process Integration for Efficient Use of Energy, 2 nd ed., Butterworth-Heinemann, USA, 2007.	
Reference Books		
1.	Joshi. M.V., Mahajani. V.V., Process Equipment Design, 3 rd ed., Mc-Millan India Ltd., India, 2000.	
2.	Richard A. Turton, Richard C. Bailie, Wallace B. Whiting, Joseph A. Shaeiwitz, Debangsu Bhattacharyya, Analysis, Synthesis and Design of Chemical Processes, 4 th ed., Prentice Hall, USA, 2013.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Laboratory Experiments		
1.	Basics of 3D drawing and applications	2 hours
2.	Extrusion of surfaces and geometries	2 hours
3.	Design and drawing of Pressure vessel to dimensions	2 hours
4.	Design and drawing of a Shell and Tube heat Exchanger	2 hours
5.	Design and drawing of a bubble cap tray	2 hours
6.	Design and drawing of Rotary Louvre dryer	2 hours
7.	Analysis of the performance of a Heat Exchanger (Aspen)	2 hours
8.	Design and analysis of Distillation Column (Aspen)	2 hours
9.	Cost Estimation of a Distillation Column (Aspen)	2 hours
10.	Dynamics simulation experiment on distillation column (Aspen)	2 hours
Total Laboratory Hours		20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies		15-04-2019
Approved by Academic Council		55 th Date 13-06-2019

Course code	CHE3001	L	T	P	J	C
Course title	COMPUTATIONAL METHODS IN PROCESS ENGINEERING	3	0	2	0	4
Pre-requisite	MAT3003	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Formulate problems for roots of a function, solution of simultaneous equations, optimized value of a given function, numerical integration and differentiation, ODE and PDE 2. Solve roots of a function, simultaneous equations, optimization, numerical integration, ODE and PDE 3. Develop MATLAB code for finding the roots of a function, solution of a simultaneous equations, optimization, numerical integration, ODE and PDE 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Determine the roots of a single equation using closed and open computational methods 2. Compute the solution for simultaneous equations using direct and indirect computational methods 3. Apply suitable optimization, regression and numerical integration methods for engineering applications 4. Solve ordinary differential equations (ODE) involving initial value and boundary value problems 5. Solve partial differential equation (PDE) involving elliptical and parabolic equation 6. Develop MATLAB program for the solution of single equation, simultaneous equations, optimization, regression, numerical integration, ODE and PDE 						
Module:1	Finding the Roots	6 hours				
Computers and error analysis, Mathematical models for solving engineering problems, programming and software; Finding roots of a single equation - Direct methods (bisection, Regula falsi) - Indirect methods (Newton-Raphson, Secant method)						
Module:2	Solution for Simultaneous Equations	5 hours				
Types of matrices and matrix operation rules; Solution for linear system of simultaneous equations - Direct methods (Gauss Elimination, Gauss Jordan), Iterative methods (Gauss-Jacobi and Gauss-Seidel); Overview of non-linear system of equations						
Module:3	Interpolation and Regression Analysis	7 hours				
Newton's divided-difference interpolating polynomial - Linear-polynomial-quadratic rules; Lagrange interpolating polynomial - Linear-polynomial Regression.						
Module:4	Optimization	7 hours				
One-Dimensional Unconstrained Optimization - Golden section search and Newton's Method; Overview on multidimensional unconstrained optimization - gradient and non-gradient methods; Constrained optimization - Simplex method; Optimization of Chemical Processes using Aspen Plus.						
Module:5	Integration and Differentiation	5 hours				
Newton's cotes Integration - Trapezoid method - Simpson's 1/3 rd rule - Simpson's 3/8 th rule; Numerical differentiation - Forward- Backward- Central difference methods						

Module:6	Ordinary Differential Equations	6 hours
Initial Value Problems –Euler - Predictor-corrector - Runge-Kutta methods; Boundary Value Problems –Shooting method - Central difference method		
Module:7	Partial Differential Equations	7 hours
Finite difference solutions of elliptic equations –Liebmann’s method- finite difference solutions of parabolic equations –Crank-Nicolson and implicit methods- Overview of hyperbolic equations; Case study on solving PDEs		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	Chapra S.C, Canale R.P, Numerical Methods for Engineers, 7 th ed., McGraw Hill Publications, USA, 2016.	
2.	Kamal I.M., Al-Malah, Aspen Plus: Chemical Engineering Applications, 1 st ed., John Wiley & Sons Inc., USA, 2016.	
Reference Books		
1.	Dorfman K.D., Daoutidis P, Numerical Methods with Chemical Engineering Applications, 1 st ed., Cambridge University Press, USA, 2017.	
2.	Jana A.K., Chemical Process Modelling and Computer Simulation, 2 nd ed., Prentice Hall of India, India, 2011.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Laboratory Experiments		
1.	Develop MATLAB code for bisection / Regula falsi method	2 hours
2.	Develop MATLAB code for Newton Raphson / Secant method	2 hours
3.	Develop MATLAB code for Gauss Elimination / Gauss Jordan method	2 hours
4.	Develop MATLAB code for Gauss Seidel method	2 hours
5.	Develop Aspen Plus simulation for solving simultaneous equations in distillation column	2 hours
6.	Develop MATLAB code for Numerical Integration	2 hours
7.	Develop MATLAB code for ODE – Euler / Modified Euler method	2 hours
8.	Develop MATLAB code for ODE – Runge Kutta method	2 hours
9.	Develop MATLAB code for PDE – Liebmann’s method	2 hours
10.	Develop MATLAB code to optimize a chemical process involving PDE	2 hours
Total Laboratory Hours		20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by Academic Council	55 th	Date 13-06-2019

Course code	CHE3002	L	T	P	J	C
Course title	PROCESSINSTRUMENTATIONAND CONTROL	2	0	2	4	4
Pre-requisite	MAT3003	Syllabus version				
		1.2				
CourseObjectives:						
<ol style="list-style-type: none"> 1. Understandthe basic concepts of measuring instruments usedin process industries 2. Explaintheimportanceofprocesscontrolmechanismandtheirapplicationsinchemicalprocess industries 3. Describeprinciplesofmodesofcontrollersandtheirgeneralcharacteristicsandstudythestability analysis of digitalcontrolsystem 						
CourseOutcomes (CO):						
<ol style="list-style-type: none"> 1. Explain chemical process systems and the operating principles of common instruments. 2. Analyze the performance of the obtained transfer function model of the given open loop control systems for various standard inputs. 3. Analyze the performance of closed loop system for set point and load input variations. 4. Select suitable controllers required for the given process to obtain the desired performance. 5. Analyze the stability of the given control system using time domain and/or frequency domain technique. 6. Compare the performance of various advanced control schemes used in various processes. 						
Module:1		Process Instrumentation				4 hours
Principlesandclassificationofprocesscontrolinstruments-Temperature-Pressure-FluidFlowRate-LiquidLevel-pH-Viscosity-HumidityofgasesandConcentrationbySpectroscopyandChromatographymethods.						
Module:2		Introduction to ProcessControl				6 hours
Laplacetransformation-transformofstandardfunctions-derivativesandintegrals-inversiontheorems-Openloopssystem-Transferfunctions-Forcingfunctions-step,pulse,impulseandsinusoidal-FirstorderandHigherordersystemdynamics-Firstordersystemsinseries-linearizationanditsapplicationinprocesscontrol-Continuousandbatchprocesses-Transportationlag.						
Module:3		Feedback Control Block Diagram				4 hours
Closedloopssystem-Developmentofblockdiagram-Blockdiagramreduction-ServoandRegulatorproblem-Transientresponseofclosedloopcontrolsystemsandtheirstability-OFFSETcalculation.						
Module:4		Controllers andControl Action				4 hours
Transferfunctionofcontrollersandcontrolvalve-CharacteristicsofON-OFF,Proportional,IntegralandDerivativecontrolmodes-P-PI-PD-PIDcontrolmodes-PrinciplesofPneumatic and ElectronicControllers - I/P converter - Control valve- Construction-Sizing - Characteristics.						

Module:5	Time and Frequency Domain Analysis	5 hours
Stability criteria-Routh's stability criteria-Root locus diagram-Frequency response analysis-Gain margin- Phase margin and crossover frequency - Bode plot - Polar plot and Nyquist plot.		
Module:6	Controller Tuning	2 hours
Process reaction curve - Cohen-Coon method - IMC tuning - Ziegler Nichols method.		
Module:7	Advanced Process Control	3 hours
Introduction to multivariable control- Computer applications in process control - Advanced control strategies-Cascade control-Ratio control-Feed-Forward control-Inferential control- Adaptive control - Control of Reactor-Distillation towers-Heat Exchangers.		
Module:8	Contemporary issues	2 hours
Total Lecture hours		30 hours
Text Books		
1.	Seborg D.E., Edgar, T.F., Mellichamp D.A., Process Dynamics and Control, 3 rd ed., Wiley India, New Delhi, 2013.	
2.	Stephanopoulos G., Chemical Process Control, 1 st ed., Pearson Education India, New Delhi, 2015.	
Reference Books		
1.	Coughanowr C.R., Koppel L.M., Process System Analysis and Control, 3 rd ed., McGraw Hill, New Delhi, 2013.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Laboratory Experiments		
1.	Automatic temperature control loop in a heating tank.	2 hours
2.	Automatic level control loop in a cylindrical tank.	2 hours
3.	Automatic flow control loop in a pipe line.	2 hours
4.	Automatic cascade control loop.	2 hours
5.	Dynamics of non-interacting tanks.	2 hours
6.	Dynamics of interacting tanks.	2 hours
7.	Controller tuning using an open loop method (Cohen-Coon method) in Simulink.	2 hours
8.	Controller tuning using a closed loop method (Ziegler-Nichols method) in Simulink.	2 hours
9.	Control Valve Characteristics.	2 hours
10.	Dynamics of Ratio control using ProSIM.	2 hours
Total Laboratory Hours		20 hours
Recommended by Board of Studies		15-04-2019
Approved by Academic Council		55 th Date 13-06-2019

Course code	CHE3003	L	T	P	J	C
Course title	MASSTRANSFER	3	0	0	0	3
Pre-requisite	MAT3003,CHE1005	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand the principles of diffusion in gas, liquid and solid phases 2. Interpret the relation between mass transfer coefficients and the theories of mass transfer for different separation operations 3. Demonstrate the working principles of cooling tower, dryer and crystallizer 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Determine diffusivity in fluids and solids using correlations and theories 2. Compute various mass transfer coefficients through correlations/analogies used in various chemical engineering applications 3. Apply the theories of mass transfer for individual and overall mass transfer coefficients 4. Estimate the design parameters required for humidification and dehumidification equipment 5. Estimate the psychrometric properties of gas-vapor systems using charts and/or equations 6. Explain different types of mass transfer equipment, i.e., cooling tower, drier, and crystallizer used for industrial applications 						
Module:1	Diffusion	6 hours				
Introduction to Mass transfer operation, Fick's law of diffusion, Steady state molecular diffusion in fluids under stagnant and laminar flow conditions, Diffusion coefficient measurement and prediction						
Module:2	Molecular diffusion in Fluids	6 hours				
Molecular diffusion in gas and Liquids, Multicomponent diffusion, Diffusion through variable cross-sectional area, Diffusivity in solids and its applications						
Module:3	Mass transfer coefficients	6 hours				
Introduction to mass transfer coefficient, Correlation for convective mass transfer coefficient, Correlation of mass transfer coefficients for single cylinder, Packed column, flow over a flat plate						
Module:4	Theories of mass transfer	5 hours				
Penetration theory, Surface Renewal Theory, Interphase mass transfer, two film theory, Overall mass transfer coefficients						
Module:5	Humidification	7 hours				
Basic concepts, Principles of Humidification–Definitions Wet Bulb Temperature & Adiabatic Saturation Temperatures–Air/Water System psychrometric and Psychrometric Charts–Utilization of Psychrometric Charts–Dehumidification–Cooling Towers–Mechanical Draft Towers: forced draft towers and induced draft towers; Design calculations of cooling tower						
Module:6	Drying	7 hours				

Principles of Drying- Definitions of moisture and other terms on Drying, Classification of Drying operations. Rate of Drying- Constant and Falling Rate Drying. Moisture movement in solids - Through Circulation Drying- Rate of drying for Continuous Direct heat Driers-Types of Dryers used in practice and their operation-			
Module:7		Crystallization	6 hours
Crystal Geometry-Invariant Crystals-Principles of Crystallization-Supersaturation-Nucleation- Crystal growth-Material & Energy Balance applied to Crystallizers-Types of Crystallizers used in practice			
Module:8		Contemporary issues	2 hours
Total Lecture hours			45 hours
Text Books			
1.	Dutta, B.K., Principles of Mass Transfer and Separation Processes. Prentice-Hall of India, New Delhi 2007.		
2.	Treybal, R.E., Mass-Transfer Operations, 3 rd ed, McGraw-Hill 1981.		
Reference Books			
1.	Cussler, E.L., Diffusion: Mass Transfer in Fluid Systems, Cambridge University Press, 2017		
2.	Christie J Geankoplis, Transport Processes and Unit Operations, 4 th ed, Prentice Hall India Pvt. Ltd, 2003		
3.	Anantharaman N, Meera Sheriffa Begum K. M., Mass Transfer-Theory and Practice, Prentice-Hall of India, New Delhi, 2011		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE4001	L	T	P	J	C
Course title	EQUILIBRIUM STAGED OPERATIONS	2	0	2	4	4
Pre-requisite	CHE3003	Syllabus version				
		1.2				
Course Objectives:						
<p>1. Understand the basic principles of staged and continuous contact separation equipment involved in equilibrium staged operations such as distillation, absorption, liquid-liquid extraction, leaching, adsorption and other modern separation operations</p> <p>2. Perform basic design calculations for staged and continuous contact equilibrium staged separation operations</p> <p>3. Describe various types of equipment's and modern separation methods for high purity products widely used in separation operation</p>						
Course Outcomes (CO):						
<p>1. Apply material and energy balances for various mass transfer operations.</p> <p>2. Calculate the number of equilibrium stages required for various equilibrium staged operations.</p> <p>3. Determine the number of transfer units and height requirements for continuous contact equilibrium staged operations.</p> <p>4. Identify the different column/equipment used for various separation applications.</p> <p>5. Choose modern separation techniques applied in industries for high purity products.</p> <p>6. Evaluate the performance of various equilibrium staged operations using experimental setup and/or simulation tools</p>						
Module:1	Introduction to Equilibrium Staged Operations	4 hours				
Introduction to various equilibrium staged operations-Distillation-absorption-Extraction-leaching-adsorption; Vapour-liquid equilibria; Types of distillation-Flash-azeotropic-Extractive distillations; Develop VLE data using Aspen Plus; Simple mass and energy balance in flash column using simulation software						
Module:2	Distillation	6 hours				
Distillation column-Types of contact-Tray Vs Packed Column; Derivation of operating line equation for different section and part of distillation column-rectification section-stripping section-feed tray location-condenser-reboiler-efficiency of distillation column; Determination of theoretical trays-McCabe-Thiele method-Ponchon-Savarit method; Case study of Industrial distillation column for multicomponent separation using						
Module:3	Absorption	4 hours				
Introduction to absorption-Continuous contact counter-current multi-stage absorption (Tray absorber); Design of packed tower based on overall mass transfer coefficient; Absorber column operation using Aspen Plus						
Module:4	Extraction	3 hours				
Liquid-Liquid equilibria-Determination of number theoretical stages-co-current-crosscurrent - countercurrent contact operations-Classification of extraction equipment						
Module:5	Leaching	3 hours				

General principles of leaching-Factors influencing the leaching rate-Equipment for leaching-Advanced industrial leaching processes			
Module:6	Adsorption		4 hours
Adsorption theory-Structure of adsorbents-Adsorption isotherms-Langmuir and Freundlich isotherms - Adsorption equipment			
Module:7	Modern separation techniques		4 hours
Membrane separation-microfiltration-ultrafiltration-nanofiltration-reverse osmosis; Chromatography-liquid chromatography-Advanced separation techniques - Divided wall column, melt crystallization, zone melting; Develop membrane separators using Aspen and solving for optimum purification			
Module:8	Contemporary issues		2 hours
Total Lecture hours			30 hours
Text Books			
1.	Dutta B.K., Principles of Mass transfer and Separation Processes, 1 st ed., Prentice Hall of India, India, 2007.		
2.	Seader J.D., Henley E.J., Roper D.K., Separation Process Principles, 3 rd ed., John Wiley & Sons, USA, 2010.		
Reference Books			
1.	Treybal R.E., Mass-Transfer Operations, 3 rd ed., McGraw-Hill Inc., USA. 1981.		
2.	Jana A.K., Chemical Process Modelling and Computer Simulation, 2 nd ed., Prentice Hall of India, India, 2011.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Laboratory Experiments			
1.	Diffusion in gas phase	2 hours	
2.	Diffusion in liquid phase	2 hours	
3.	Wetted wall column	2 hours	
4.	Vapor-liquid equilibria using Aspen Plus or ProSim	2 hours	
5.	Simple distillation	2 hours	
6.	Multi Component distillation using Aspen Plus or ProSim	2 hours	
7.	Liquid-liquid equilibria using Aspen Plus or ProSim	2 hours	
8.	Liquid-liquid extraction	2 hours	
9.	Continuous distillation using Aspen Plus or ProSim	2 hours	
10.	Adsorption using Aspen Plus or ProSim	2 hours	
Total Laboratory Hours			20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Coursecode	Applications of Differential and Difference Equations	L	T	P	J	C
MAT2002		3	0	2	0	4
Pre-requisite	MAT1011 -Calculus for Engineers	Syllabus Version				
		1.0				
Course Objectives (CoB):						
<p>[1] Presenting the elementary notions of Fourier series, which is vital in practical harmonic analysis</p> <p>[2] Imparting the knowledge of eigenvalues and eigen vectors of matrices and the transform techniques to solve linear systems, that arise in sciences and engineering</p> <p>[3] Enriching the skills in solving initial and boundary value problems</p> <p>[3] Impart the knowledge and application of difference equations and the Z-transform in discrete systems, that are inherent in natural and physical processes</p>						
Course Outcome (CO):						
<ol style="list-style-type: none"> 1. Apply the Fourier series tools in the fields of physics and engineering. 2. Apply the concepts of eigenvalues, eigen vectors and diagonalisation in engineering systems. 3. Analyse the concepts of differential equations and apply them to obtain the solutions of homogeneous and non-homogeneous systems. 4. Analyze the series solutions of Bessel's, Legendre and Sturm-Liouville's problem by using eigen values and eigen functions. 5. Analyse and apply the Z-transform in digital signal processing and control systems. 6. Develop programming tools for engineering problems and visualize solutions. 						
Module:1	Fourier series:	6 hours				
Fourier series -Euler's formulae- Dirichlet's conditions- Change of interval- Half range series – RMS value – Parseval's identity– Computation of harmonics						
Module:2	Matrices:	6 hours				
Eigen values and Eigen vectors- Properties of eigen values and eigen vectors– Cayley-Hamilton theorem- Similarity of transformation- Orthogonal transformation and nature of quadratic form						
Module:3	Solution of ordinary differential equations:	6 hours				
Linear second order ordinary differential equation with constant coefficients– Solution of homogeneous and non-homogeneous equations- Method of undetermined coefficients– method of variation of parameters– Solutions of Cauchy-Euler and Cauchy-Legendre differential equations						
Module:4	Solution of differential equations through Laplace transform and	8 hours				
Solution of ODE's- Nonhomogeneous terms involving Heaviside function, Impulse function- Solving nonhomogeneous system using Laplace transform– Reduction of n th order differential equation to first order system- Solving nonhomogeneous system of first order differential equations $(X' = AX + G)$ and						

Module:5	StrumLiouville's problems and power series Solutions:	6 hours
TheStrum-Liouville's Problem- Orthogonalityof Eigenfunctions-Seriessolutions ofdifferentialequations about ordinaryandregularsingularpoints- Legendredifferentialequation - Bessel'sdifferentialequation		
Module:6	Z-Transform:	6 hours
Z-transform -transforms of standardfunctions-InverseZ-transform:bypartial fractionsandconvolutionmethod		
Module:7	Difference equations:	5 hours
Differenceequation-Firstandsecondorderdifferenceequationswithconstantcoefficients -Fibonaccisequence-Solutionofdifferenceequations-Complementaryfunction- Particularintegralbythemethodofundeterminedcoefficients- Solutionofsimpledifferenceequations using Z-transform		
Module:8	ContemporaryIssues	2 hours
IndustryExpertLecture		
Total Lecture hours:		45 hours
Text Book(s)		
1.	AdvancedEngineeringMathematics,ErwinKreyszig,10 th Edition,JohnWiley India, 2015	
Reference Books		
1.	HigherEngineeringMathematics, B. S. Grewal, 43 rd Edition, Khanna Publishers, India, 2015	
2.	AdvancedEngineering MathematicsbyMichaelD. Greenberg,2 nd Edition,Pearson Education,Indian edition,2006	
Mode ofEvaluation		
DigitalAssignments(Solutions byusing soft skills), ContinuousAssessmentTests, Quiz, FinalAssessmentTest		
1.	SolvingHomogeneousdifferentialequations arisinginengineeringproblems	2 hours
2.	Solvingnon-homogeneousdifferentialequationsandCauchy,Legendre equations	2 hours
3.	Applyingthe technique ofLaplace transform to solve differentialequations	2 hours
4.	Applications of Second order differentialequations to Mass springsystem (damped, undamped,Forcedoscillations),LCRcircuitsetc.	2 hours
5.	VisualizingEigen valueand Eigen vectors	2 hours
6.	Solvingsystem of differential equationsarisinginengineeringapplications	2 hour
7.	Applyingthe Power series method to solve differentialequationsarisingin engineeringapplications	2 hours
8.	ApplyingtheFrobeniusmethod to solve differentialequationsarisingin engineeringapplications	2 hours

9.	Visualising Bessel and Legendre polynomials	2 hours
10.	Evaluating Fourier series-Harmonic series	2 hours
11.	Applying Z-Transforms to functions encountered in engineering	2 hours
12.	Solving Difference equations arising in engineering applications	2 hours
Total Laboratory Hours		24 hours
Mode of Evaluation: Weekly Assessment, Final Assessment Test		
Recommended by Board of Studies	03-06-2019	
Approved by Academic Council	No. 55	Date 13-06-2019

SCHEME 2020

Coursecode	ComplexVariables andPartialDifferentialEquation	L	T	P	J	C
MAT3003		3	2	0	0	4
Pre-requisite	MAT2002 ApplicationsofDifferential andDifference Equations	Syllabusversion				
		1.0				
CourseObjectives (CoB):						
Theaim of this course isto present a comprehensive,compactandintegratedtreatment of twomost importantbranchesof appliedmathematics for engineersand scientists namelythefunctions of complexvariable andPartialdifferentials equations in finite and infinite domains						
CourseOutcomes(CO):						
<ol style="list-style-type: none"> 1. Construct analytic functions and find complex potential of fluid flow and electric fields 2. Apply conformal mapping in geometrical transformations 3. Evaluate analytic function in Taylor's series, Laurent's series and pole of order, and apply Cauchy Residue theorem in evaluating some real integrals 4. Solve first order partial differential equations and find the solution of partial differential equations, and its applications in one dimensional heat and wave equations 5. Apply Fourier series, Fourier transform techniques to solve engineering problems 						
Module:1	AnalyticFunctions	6 hours				
Complexvariable-Analyticfunctionsand Cauchy– Riemann equations - LaplaceequationandHarmonicfunctions - Construction of Harmonicconjugateandanalyticfunctions- Applicationsof analytic functions to fluid-flow andField problems.						
Module:2	Conformal and Bilineartransformations	5 hours				
Conformalmapping -Elementarytransformations- translation,magnification,rotation,inversion.ExponentialandSquaretransformations($w = e^z, z^2$)- Bilineartransformation-Cross-ratio-Images of the regionsboundedbystraight lines under the abovetransformations.						
Module:3	Powerseries	4 hours				
FunctionsgivenbyPower Series-Taylor andLaurentseries-singularities -poles – Residues.						
Module:4	ComplexIntegration	5 hours				
Integration ofacomplexfunctionalong acontour-Cauchy-Goursat theorem-Cauchy's integralformula-Cauchy'sresiduethereorem-Evaluationofrealintegrals-Indentedcontourintegral.						
Module:5	PartialDifferentials equationsoffirstorder	6 hours				
Formationand solution of partialdifferential equation- General,Particular,CompleteandSingularintegrals- PartialDifferentials equationsof firstorder of theforms: $F(p,q)=0, F(z,p,q)=0, F(x,p)=G(y,q)$ andClairaut's form- Lagrange's equation: $Pp+Qq =R$.						
Module:6	ApplicationsofPartialDifferential	10 hours				

	Equations		
Linear partial differential equations of higher order with constant coefficients. Solution of a partial differential equation by separation of variables-Boundary Value Problems- one dimensional wave and heat equations- Fourier series solution.			
Module:7	Fourier transforms	7 hours	
Complex Fourier transform and properties-Relation between Fourier and Laplace transforms- Fourier sine and cosine transforms-Convolution Theorem and Parseval's identity.			
Module:8	Contemporary issues:	2 hours	
Industry Expert Lecture			
		Total Lecture hours:	45 hours
Tutorial	<ul style="list-style-type: none"> • A minimum of 10 problems to be worked out by students inventory Tutorial Class • Another 5 problems per Tutorial Class to be given as home work. 		30 hours
Text Book(s)			
1.	Advanced Engineering Mathematics, Erwin Kreyszig, 10 th Edition, John Wiley & Sons (Wiley Student Edition) (2015)		
Reference Books			
1	Higher Engineering Mathematics, B. S. Grewal, 43 rd Edition (2019), Khanna Publishers, New Delhi		
2	A first course in complex analysis with applications, G. Dennis Zill, Patrick D. Shanahan, 3 rd Edition, 2013, Jones and Bartlett Publishers Series in Mathematics:		
3	Advanced Engineering Mathematics, Michael, D. Greenberg, 2 nd Edition, Pearson Education (2006)		
4	Advanced Engineering Mathematics, Peter V. O'Neil, 7 th Edition, Cengage Learning (2012)		
5	Complex Analysis for Mathematics and Engineers, JH Mathews, R. W. Howell, 5 th Edition, Narosa Publishers (2013)		
Mode of Evaluation:			
Digital Assignments (Solutions by using soft skill), Quiz, Continuous Assessments, Final Assessment Test.			
Recommended by Board of Studies		03-06-2019	
Approved by Academic Council		No. 55	Date 13-06-2019

CourseCode	ENGINEERING DRAWING	L	T	P	J	C
MEE1001		1	0	4	0	3
Pre-requisite	NIL	Syllabus version				
		v. 2.2				
CourseObjectives:						
<ol style="list-style-type: none"> 1. Understand and escalate the importance of basic concepts and principles of Engineering Drawing (components, sections, views, and graphical representation). 2. Enable the students with various concepts like dimensioning, conventions and standards related to working drawings in order to become professionally efficient. 3. Develop the ability to communicate with others through the language of technical drawing and sketching. 4. Ability to read and interpret engineering drawings created by others. 5. Ability to draw orthographic projections and sections. 6. Develop an understanding for size specification procedures and use of SI and traditional units of linear measure. 						
CourseOutcomes:						
<ol style="list-style-type: none"> 1. Apply BIS and ISO Standards in Engineering Drafting. 2. Graphically construct mathematical curves in engineering applications. 3. Visualize geometrical solids in 3D space through Orthographic Projections 4. Construct isometric scale, isometric projections and views. 5. Draw sections of solids including cylinders, cones, prisms and pyramids. 6. Draw projections of lines, planes, solids, isometric projections and sections of solids including cylinders, cones, prisms and pyramids using Mini-Dafter and CAD. 7. Construct orthographic projections from pictorial views 						
Module:1	Lettering and Dimensioning					1 hours
Introduction, lettering practice, Elements of dimensioning- systems of dimensioning.						
Module:2	Geometric Constructions					2 hours
Free hand sketching, Conic sections, Special curves.						
Module:3	Projection of Points and Projection of Lines					2 hours
Projection of Points: First and Third Angle Projections; Projection of points. Projection of Lines: Projection of straight lines (First angle projection only); Projection of lines inclined to one plane and both planes, true length and true inclinations.						
Module:4	Projection of Solids and Section of Solids					3 hours
Projection of solids: Classification of solids, Projection of solids in simple position, Projection of solids inclined to one plane. Sections of Solids: Right regular solids and auxiliary views for the true shape of these sections.						

Module:5	Development of Surfaces	2 hours
Development of surfaces for various regular solids.		
Module:6	Isometric Projection and Perspective Projection	2 hours
Isometric Projection: Isometric scales, Isometric projections of simple and combination of solids; Perspective Projection: Orthographic representation of a perspective view – Plane figures and simple solids - Visual ray method.		
Module:7	Orthographic Projection	2 hours
Conversion of pictorial view into orthographic Projection.		
Module:8	Contemporary issues	1 hours
Total Lecture hours:		15 hours
Text Book(s)		
1.	Venugopal K and Prabhu Raja V, "Engineering Graphics", New AGE International Publishers, 2015.	
Reference Books		
1.	N. D. Bhatt, Engineering Drawing, Charotar Publishing House, 2012.	
2.	Natarajan, K. V., A Text book of Engineering Graphics, Dhanalakshmi Publishers, 2012.	
Mode of Evaluation: CAT / Assignment / Quiz / FAT / Project / Seminar		
List of Challenging Experiments (Indicative)		
1.	Identifying the incorrect dimensioning and correcting it as per BIS standards for Engineering Components.	4 hours
2.	Tutorials on free hand sketching of the plan view of stadium, garden, etc.,	4 hours
3.	Tutorials on geometric constructions like conics and special curves for projection of cricket ball, missile projection, etc.,	4 hours
4.	Representation of orthographic projection of points	4 hours
5.	Representation of orthographic projection of lines (First angle projection only) inclined to one plane and projection of lines inclined to both the planes - solving problems like electrical bulb hanging from the roof, finding the shortest distance between fan to electrical switch board, etc.,	8 hours
6.	Sketching orthographic projection of solids in simple position and projection of solids inclined to one plane for household accessories and objects.	8 hours
7.	Drawing the auxiliary views, orthographic views and true shape of sectioned regular solids	4 hours
8.	Development of lateral surfaces of the regular shapes and sectioned shapes for water cans, refrigerator, cylinder container, funnel, etc.,	4 hours
9.	Conversion of orthographic views to isometric views for engineering components.	8 hours
10.	Tutorial problems on perspective projection of plane figures and simple solids for train with track, landscape, etc.,	4 hours

11.	Conversion of pictorial drawing into orthographic projection for engineering components, architectural structures, etc.,	8 hours
Total Laboratory Hours		60 hours
Mode of assessment:		
Recommended by Board of Studies	17-08-2017	
Approved by Academic Council	47	Date 05-10-2017

SCHEME 2020

PROGRAMME ELECTIVE

Course code	CHE1007	L	T	P	J	C
Course title	SAFETY AND HAZARD ANALYSIS	2	0	0	4	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Critically understand the importance of safety in process industries 2. Assess and identify the potential hazards in process industries 3. Identify and evaluate the causes of accident in a chemical industry 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Identify the typical sources of risk in a process plant by hazard identification and examination of case studies 2. Analyze the severity of the consequences of incidents 3. Develop a Hazard and Operability Study (HAZOP) 4. Understand the legal frameworks controlling process plant safety in industries 5. Demonstrate how the root cause of incidents can be investigated and analysed and the various human and technical aspects of such causes 6. Identify hazard and conduct safety audit 						
Module:1	Introduction to Safety in Chemical Process Industries	5 Hours				
Need for Development of Safety Consciousness in Chemical Industries- Hazard, Risk, Danger, Accident; Promotion of industrial safety, extreme operating conditions, toxic chemicals- safe handling; Psychological attitude towards safety.						
Module:2	Safety Programs in Industries	5 Hours				
Importance of Safety Programs in industries; Elements of Safety Program; Effective Realization; Economic and Social Benefits from Safety Program; Effective Communication Training at various levels of Production and Operation. Accidents identification and prevention.						
Module:3	Potential Hazards in Chemical Process Industries	4 Hours				
Chemical and Physical job Safety Analysis; High pressure and Temperature Operation; Dangerous and Toxic Chemicals; Routes of entry, Effects of toxicants and its elimination. Toxic release and dispersion models. Radio Active materials; Safe Handling and Operation of materials and Machinery; periodic inspection and replacement.						
Module:4	Risk assessment	4 hours				
Quantitative risk assessment- rapid and comprehensive risk analysis; Risk due to Radiation, explosion due to overpressure, plant layout Personnel Safety and Protective Equipment; Occupational health and safety.						
Module:5	Hazard Identification	4 hours				
Introduction to Hazard identification- Overall risk and hazard analysis- Emergency planning- Onsite & offsite emergency planning- Risk management- ISO 14000- Safety audits- Checklist- What if analysis - Vulnerability models - Event tree analysis - Fault tree analysis.						

Module:6	HAZOP			4 hours
HAZOP study-case studies-pumpingsystem-reactor-mass transfersystem.Hazard Identification and Assessment; Involvement of Human factors and Errors- Hazard Quantifications- disaster management; Occupational and Industrial Health Hazards; Safety Systems.				
Module:7	Case studies			2 hours
Dominoseffect, Worstcasescenario, Fire, Accidents, Chemical release, Explosion, Petroleum, Commercial, Natural disasters, EMS models case studies				
Module:8	Contemporary Issues			2 hours
Total Lecture hours				30 hours
Text Books				
1.	Ericson C.A., Hazard Analysis Techniques for System Safety, 2 nd ed., Wiley, USA, 2015.			
2.	Gupta A., Industrial Safety and Environment, 2 nd ed., Laxmi Publications, India, 2015			
Reference Books				
1.	Hyatt, N., Guidelines for process hazards analysis, hazards identification & risk analysis, 1 st ed., CRC Press, USA, 2003.			
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test				
Recommended by Board of Studies		15-04-2019		
Approved by Academic Council		55 th	Date	13-06-2019

Course code	CHE1008	T	P	J	C	
Course title	UNIT PROCESSES IN ORGANIC SYNTHESIS	3	0	2	0	4
Pre-requisite	NIL	Syllabus version				
		2.1				
Course Objectives:						
<ol style="list-style-type: none"> 1. Impart knowledge on the industrial reactions used in converting organic raw materials into usable products by various processes 2. Develop student understanding towards kinetics and mechanism of various reactions involved in industries 3. Comprehend various instrumental techniques applied in contemporary industries to analyze the organic compounds 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Review the physical concepts behind the organic reactions 2. Understand the importance of heterocycles, oxidizing and reducing agents 3. Interpret kinetics and mechanism of nitration, and halogenation reactions 4. Explain the kinetics and mechanism of sulphonation reactions 5. Explain separation and purification of organic compounds through classical separation methods 6. Select suitable chromatographic technique for separation and purification of organic compounds 						
Module:1	Basic concepts					6 hours
Kinetic theory of gases - Vander Waal's equation - Critical constants - Liquifaction of gases, Raoult's law - Ideal solutions - Partially miscible liquids - Phenol water system - Henry's law - Colligative properties - Lowering of vapor pressure - Elevation of boiling point - Depression of freezing point						
Module:2	Heterocyclic compounds					8 hours
Aromatics: Structure of benzene and theories of aromaticity - Heterocyclic compounds: Classification - Aromaticity and Basicity of heterocyclic compounds - Preparation and properties of Furan - Thiophene - Pyrrole. Oxidation - Oxidising agents (SeO_2 , OsO_4 , KMnO_4) - Reduction, Reducing agents (Lithium aluminium hydride, metal/acid and sodium metal). Grignard reagents: Synthesis and applications.						
Module:3	Nitration					6 hours
Introduction to nitration, Nitrating agents, Kinetics and Mechanism of aromatic nitration process, Equipments for nitration, typical industrial nitration processes e.g. preparation of nitrobenzene, nitroacetanilide.						
Module:4	Halogenation					6 hours
Halogenating agents, Kinetics and mechanism of halogenation reactions. Apparatus and materials for construction. Technical preparation of chloral and vinyl chloride.						
Module:5	Sulphonation					6 hours
Introduction to sulphonation, sulphonation agents and sulphanation agents, chemical and physical factors affecting sulphonation. Mechanism of sulphonation, commercial sulphon benzene and naphthalene, sulphation of lauryl alcohol and dimethyl ether.						
Module:6	Separation and purification methods					5 hours
Separation and purification methods: Classical separation methods: Theories of distillation, fractional distillation, steam distillation, sublimation and zone refining - Solvent extraction - Distribution law - Separation of mixtures, Craig method; Recrystallization of solid products.						

Module:7	Chromatography	6 hours
Chromatography-Introduction,Differenttypesofchromatographictechniques-TLC,Column,GC,LC,andHPLC-TheoryandInstrumentation(GCandHPLC),Applicationsintheseparationoforganic molecules.		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	GrogginsP.H.,UnitProcessesinOrganicSynthesis,5 th ed.,TataMc.GrawHillBookCompany, India, 2009.	
2.	PuriB.R.,SharmaL.R.,PathaniaM.S.,PrinciplesofPhysicalChemistry,43 rd ed.,VishalPublishingCo., India, 2008.	
Reference Books		
1.	Atkins,P.,Paula,J.D.Atkins,PhysicalChemistry,11 th ed.,OxfordUniversityPress,USA,2018.	
2.	March,J.,AdvancedOrganicChemistry:Reactions,MechanismsandStructures,4 th ed.,John Wiley& Sons, USA, 1992.	
3.	A. Bahl, B.S. Bahl, Advanced Organic Chemistry, 5 th ed., S. Chand& Co., Ltd., India,2012.	
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
LaboratoryExperiments		
1.	Determination ofCritical Solution Temperature of the givenPhenol-Watersystem	2 hours
2.	Determination ofrate constant of the hydrolysis of ethyl acetatecatalyzed by HCl at roomtemperature	2 hours
3.	Determination ofacidvalue of thegiven oilsample	2 hours
4.	Determination of saponificationvalue of the givenoil sample	2 hours
5.	Sulphonation of 1-Naphthol	2 hours
6.	Reductionof Benzophenone by NaBH ₄	2 hours
7.	Preparation of Benzoic acid frombenzaldehyde by oxidation and itmelting point measurement	2 hours
8.	Preparation of m-DintrobenzenefromNitrobenzene by nitration andit melting pointmeasurement	2 hours
9.	Purification oforganic compounds by Fractionaldistillation	2 hours
10.	Identification of Carbonyl groupin an organiccompound.	2 hours
11.	Identification of Carboxylic acidgroup in anorganiccompound.	2 hours
12.	Preparation ofsoap or detergent	2 hours
Total Laboratory Hours		24 hours
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
Recommended by Board of Studies		15-04-2019
Approved by AcademicCouncil		55 th Date 13-06-2019

Course code	CHE1009	L	T	P	J	C
Course title	BIOCHEMICAL ENGINEERING	3	0	0	0	3
Pre-requisite	Nil	Syllabus version				
		2.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Impart the basic knowledge and overview of biotechnology covering the principles of cell and kinetics, bioreactor design, sterilization agitation and aeration 2. Understand the physical processes involved in bio-systems 3. Apply the knowledge of chemical engineering principles to biological processes 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand significance and scope of biochemical processes 2. Classify microorganisms and cell functions for industrial biochemical processes, enzyme and kinetics for bioprocesses 3. Apply Chemical Engineering Principles to develop kinetic models for bioprocesses 4. Make use of theoretical basics of chemical engineering and unit operations in designing bioprocess equipment 5. Analyze bioreactor performance 6. Distinguish downstream processing and biological Sewage treatment in solving open ended chemical problems using biochemical route 						
Module:1	Introduction to Biochemical Engineering	3 hours				
An overview of industrial biochemical processes with typical examples comparing chemical and biochemical processes – development and scope of biochemical engineering as a discipline.						
Module:2	Basic Microbiology and Biochemistry	5 hours				
Industrially important microbial strains, their classification – structure – cellular genetics – typical examples of microbial synthesis of biologicals						
Module:3	Enzymes & Applications	8 hours				
Enzymes – in industry, medicine and food – their classification with typical examples of industrially important enzymes; Mechanism of enzymatic reactions – Michaelis-Menten Kinetics – enzymes inhibition factors affecting the reaction rates; Industrial production, purification and immobilization						
Module:4	Kinetics of Cell Growth	7 hours				
Typical growth characteristics of microbial cells – factors affecting growth – Monod model; Modelling of batch and continuous cell growth; Immobilized whole cells and their characteristics						
Module:5	Unit Operations in Biochemical Engineering	6 hours				
Newtonian and non-Newtonian behaviour of broth – agitation and mixing – power consumption; Gas/liquid transport in cells – transfer resistances – mass transfer coefficients & their role in scale up of equipment – O ₂ transfer; Heat transport in microbial systems – Heat transfer correlations; Sterilization cycles; Heat addition & removal during biological production						

Module:6	Bioreactors	8 hours	
Bioreactors- Batchandcontinoustopes,immobilizedwholecellandenzymereactors.Highperformancebioreactors; Reactors inseries withand withoutrecycle.Designof reactorsandscale up with typical examples;			
Module:7	Downstream and effluenttreatmentprocesses	6 hours	
Downstreamprocessesandeffluenttreatment:DifferentUnitoperationsindownstreamingwithspecialref erence to membraneseparations,extractivefermentation;Anaerobicandaerobictreatmentofeffluents– typicalindustrialexamplesfordownstreamprocessingandeffluentdisposal.			
Module:8	Contemporary issues	2 hours	
Total Lecture hours		45 hours	
Text Books			
1.	Bailey J.B.,OllisD.F.,BiochemicalEngineeringFundamentals,4 th ed.,McGrawHill, USA, 1986.		
Reference Books			
1.	RaoD.G., Introductionto BiochemicalEngineering,1 st ed.,Tata McGrawHill, India, 2009.		
2	DoranP.M., BioprocessEngineering Principles, 3 rd ed., Academic Press,United Kingdom, 2013.		
3	Aiba A, Humphrey A.E., Milli. N.R.,Biochemical Engineering,2 nd ed., Academic Press,USA, 2004.		
Mode of evaluation: ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by AcademicCouncil		55 th	Date 13-06-2019

Course code	CHE1010	L	T	P	J	C
Course title	PROCESS PLANT UTILITIES	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Equip the students with the basic understanding and effective utilization of utilities viz. water, steam, compressor, vacuum pumps, refrigeration and cooling units, insulator, inert gases in process industries and allied operations 2. Impart insights in relation to the different types of fuels and boilers used in process industries for the generation of steam, types of compressors and blowers for handling air and inert gases 3. Expose students to different methods of treatment of wastewater and drinking water 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Explain the importance of water and various methods for water softening and purification 2. Classify the different types of fuels and boilers used in process industries for the generation of steam 3. Identify the different types of compressors and blowers for handling air and inert gases for specific applications 4. Summarize the different types of equipment used for humidification, and dehumidification 5. Select a suitable refrigeration system for a typical application in process industries 6. Interpret the application of correct type of insulation system for control of heat losses and learn about proper utilization of inert gases on the process plants 						
Module:1	Water and Steam	7 hours				
Requisites of Industrial Water and its uses; Water treatment methods- ion exchange, demineralization, membranest technology, reverse osmosis. Water resources management. Properties of steam, Boiler types and mountings, boiler accessories, Indian Boiler Act, 1923. Steam distribution and utilization, steam economy, waste heat utilization						
Module:2	Industrial fuels	6 hours				
Solid, liquid and gaseous fuels used in chemical process industries for power generation, Typical combustion calculations						
Module:3	Compressed Air	6 hours				
Types of fans, axial, reciprocating and centrifugal compressors, rotary blowers and vacuum pumps and their performance characteristics. Methods of vacuum development, ejectors and their limitations, materials handling under vacuum, piping systems.						
Module:4	Humidification and Dehumidification	5 hours				
Properties of Air- Water Vapors and use of Humidity Chart, Equipments used for Humidification, Dehumidification and						
Module:5	Refrigeration & Ventilation	6 hours				
Principle of refrigeration, Refrigeration system like compression refrigeration, absorption refrigeration, and chilled water system; Types of refrigerants; Concept of cryogenics and cryogenics characteristics. Air blending, exhaust ventilation and flaring						
Module:6	Industrial insulation and Inert Gases	8 hours				

Importance of insulation, insulation material and their effect on various materials of equipment piping, fitting and valves, insulation for high, intermediate, low and sub zero temperatures including cryogenic insulation			
Introduction, properties of inert gases & their use, sources and methods of generation, general arrangements for inerting system; operational, maintenance and safety aspects			
Module:7 Effluent treatment			5 hours
Disposal of solid, liquid and gas wastes; pollution control measures – compliance to statutory norms; Effluent Treatment – Case studies like treatment of effluents from paper mills, Dye and Textile industries, petrochemical			
Module:8 Contemporary issues			2 hours
Total Lecture hours			45 hours
Text Books			
1.	Broughton J., Process Utility Systems, 3 rd ed., Institution of Chemical Engineers, U.K., 2004		
Reference Books			
1.	Mujawar B.A., A Textbook of Plant Utilities, 3 rd ed., Nirali Prakashan Publication, India, 2007.		
2.	Poling B.E., Prausnitz J.M., O'Connell J., The Properties of Gases and Liquid, 5 th ed., McGraw Hill, USA, 2008.		
3.	Perry, R.H., Green, D.W., Perry's Chemical Engineers Handbook, 8 th ed., McGraw Hill, USA, 2007.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE-1011	L	T	P	J	C
Course title	OPTIMIZATION OF CHEMICAL PROCESSES	3	0	0	0	3
Pre-requisite	MAT3003	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Provide an overview of state-of-the-art optimization algorithms 2. Impart the theoretical knowledge of chemical engineering principles that underpin optimization techniques. 3. Enhance the modelling skills to describe and formulate optimization problems and their use for solving several types of practically relevant optimization problems in Chemical engineering 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Demonstrate the basic principles of Chemical Engineering Optimization 2. Classify the different types of optimization problems for process engineering 3. Evaluate single and multivariable optimization chemical engineering problems 4. Execute the complex chemical engineering processes using software tools 5. Identify the different types of hypotheses for the model equations chemical system 6. Solve the Optimal Control and Dynamic optimization problems 						
Module:1	Formulation of Optimization Problems					6 hours
Nature and Organization of Optimization problem; Mathematical concepts of optimization; Developing model for optimization; Taylor expansion; Gradient and Hessian matrix; Convex functions and sets; Gaussian elimination method						
Module:2	Models for Optimization					5 hours
Selection of function; Degrees of freedom; Classification of models; factorial experimental design; constraints in model; Optimality conditions for single-variable and multi-variable functions						
Module:3	Linear and Nonlinear Least square problems					6 hours
One-dimensional search- Methods requiring derivatives (Newton, Quasi Newton, Secant method); Region elimination methods (Interval halving, Fibonacci search, Golden section); Polynomial approximations (Lagrange's, quadratic &						
Module:4	Multivariable Optimization-I					6 hours
Unconstrained multivariable optimization- Graphical visualization (contour plots, 3D plots); Gradient based methods – Steepest descent, conjugate direction, and Newton methods						
Module:5	Multivariable Optimization-II					6 hours
Linear programming (LP)- Graphical solution- Simplex Method; Test for optimality – Barrier methods - Sensitivity analysis; Concept of duality; Introduction to interior-point method – Simulation of Reactor model – ASPEN PLUS and MatLab						
Module:6	Nonlinear Programming					7 hours
Nonlinear programming (NLP) with constraints; Lagrange multipliers- Graphical illustration of						

NLP problems-KKT necessary and sufficient conditions; Quadratic programming- Successive linear and quadratic programming; Penalty function method; Integer and mixed integer programming. (IP and MIP)- Graphical solution - Branch and bound methods			
Module:7	Dynamic Programming	7 hours	
Dynamic programming-Minimum cost routing problems- Solution of separable nonlinear programming problems; Global optimization problems; Introduction to multiobjective optimization problems- Pareto optimal solutions (graphical illustration)			
Module:8	Contemporary issues	2 hours	
Total Lecture hours		45 hours	
Text Books			
1.	Edger T.F., Himmelblau D.M., Lasdon L.S., Optimization of Chemical Processes, 2 nd ed., McGraw-Hill, USA, 2015.		
Reference Books			
1.	Hillier F.S., Lieberman G.J., Introduction to Operations Research, 7 th ed., McGraw-Hill, USA, 2001.		
2.	Rao S.S., Engineering Optimization: Theory and Practice, 4 th ed., John Wiley & Sons Ltd., USA, 2009.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1013	L	T	P	J	C
Course title	NATURAL GAS ENGINEERING	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Impart design experiences essential for graduates to enter the practice of Gas Engineering and pursue lifelong professional development 2. Summarize the necessary theory, application to case studies and engineering project design 3. Implement research that generates, communicates and applies new knowledge for the betterment of society 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Emphasize fundamentals of mathematics and integrates them in application to traditional Natural Gas Engineering to improve further needs 2. Select, locate and orient systems for offshore problems 3. Develop an ability to revamp and retrofit a system, process to meet desired needs within realistic constraints such as environmental, health, safety, manufacturability and sustainability in the field of Natural Gas 4. Apply natural Gas Refining principles and practices for optimizing resource development and management 5. Recognize the purification mechanism to estimate, design equipment's for processing, storage and transport 6. Inspect project economics and resource valuation methods for design and decision making under conditions of risk and uncertainty 						
Module:1	Properties and Composition of Natural Gas	6 hours				
Natural gas origin – Composition of Natural Gas – Source of Natural Gas – Thermodynamics properties – Compressibility factor for Natural Gas – Heating value and flammability limit of Natural Gas						
Module:2	Natural Gas Offshore Drilling	5 hours				
Directional Drilling and Horizontal Drilling						
Module:3	Natural Gas Offshore Production and Handling	6 hours				
Drilling Deepwater Reservoir – Deepwater production systems – Mooring Systems – Gas Terminals						
Module:4	Natural Gas Onshore Production and Handling	6 hours				
Sucker Rod pumping – Separation, Storage, Transportation of Natural Gas						
Module:5	Natural Gas Processing	8 hours				
Dehydration – Desulphurization processes (Sour gases, Toxicity of H ₂ S, Physical and Chemical Absorption process, Carbonate process, sulphur recovery) – Low temperature processes (Joule Thompson effect, Turbo						
Module:6	Liquid Recovery	6 hours				
NGL, LPG, C ₃ , C ₂ Fraction Recovery from Natural Gas						

Module:7	Economicsof NaturalGas	6 hours	
Current status in India–Trade & Selection of port location –Economicsof Gas Processing			
Module:8	Contemporary issues	2 hours	
Total Lecture hours		45 hours	
Text Books			
1.	Arthur J. Kidnay, William R. Parrish, Fundamentals of Natural Gas Processing, 4 th ed., Taylor and Francis, CRC Press, UK, 2011.		
2.	Subrata K Chakrabarti, Handbook of offshore engineering, 1 st ed., Elsevier Publishers, Netherlands, 2005.		
Reference Books			
1.	S. Mokhatab, William A. Poe, James G. Speight, Handbook of Natural Gas Transmission and Processing, 1 st ed., Gulf Professional Publishing, USA, 2014.		
2.	G. Ghalambor, Natural Gas Engineering Handbook, Gulf Publishing Company, USA, 2005.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1014	L	T	P	J	C
Course title	PETROLEUM TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand the importance of crude oil as source of fuel and the size of refining industry 2. Interpret the challenges involved in refining from viewpoint of product specifications, economic considerations and environmental regulations 3. Design application of chemical engineering principles to petroleum refining 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Explain the composition of crude oil and its products, along with its properties and characterization methods 2. Discuss the basic separation and conversion processes used in refining crude oil 3. Implement the chemical engineering principles to the analysis of safe and efficient refinery operations 4. Identify the specifications required for good quality petroleum product 5. Exemplify the process of purification and fractionation of crude oil 6. Interpret the relationship safety and environment in Petroleum Refining Industries 						
Module:1	Petroleum					6 hours
Exploration Practices-Reservoir Rock Properties-Reservoir types-Reservoir Estimation Origin – Composition-Classification and constituents of petroleum - Dehydration of crude oil- Transportation of crude oil- Classification of petroleum						
Module:2	Distillation					6 hours
Components of crude oil distillation-various crude oil distillation systems-uses of petroleum products						
Module:3	Cracking					8 hours
Necessity of cracking-Types of cracking- advantages and disadvantages of catalytic cracking over thermal cracking-Houdry's fixed bed processes- Moving bed processes-Fluid bed catalytic cracking processes						
Module:4	Reforming					4 hours
Thermal and catalytic Reforming; Polymerization; Alkylation; Isomerization						
Module:5	Purification of petroleum products					7 hours
Sweetening processes types – Merox – HDS; Dewaxing; Deasphalt; Lube oil treatment						
Module:6	Properties of Petroleum Products					7 hours
Specific gravity- Vapor pressure – Viscosity - red wood viscometer- Flash point- Fire point - Pour point- Smoke point - Aniline point- Diesel index - Octane number- Performance number- Cetane number - Properties of greases - Drop point of grease						
Module:7	Knocking					5 hours
Reasons for knocking- Additives in petrol- Aviation gasoline- Aviation turbine fuel (ATF)- Storage and handling of liquid fuels						

Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	Gary J.H., Handwerk G.E., Kaiser M.J., Petroleum Refining Technology and Economics, 6 th ed., CRC Press, USA, 2013.	
2.	Speight J.G., Petroleum Refining Process, 1 st ed., Taylor and Francis, USA, 2015	
3	Bhaskara Rao B.K., Modern Petroleum Refining Processes, 5 th ed., Oxi bh, India, 2013	
Reference Books		
1.	Mohamed A.F., Taher A., Amal E., Fundamentals of Petroleum Refining, 1 st ed., Elsevier, USA, 2010.	
2.	Nelson, Petroleum Refinery Engineering, 4 th ed., McGraw Hill, USA, 2010.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by Academic Council	55 th	Date 13-06-2019

Course code	CHE1015	L	T	P	J	C
Course title	PETROCHEMICAL TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand the technological principles of organic synthesis and related unit processes 2. Differentiate the different unit operations and unit processes involved in conversion of monomer to polymers 3. Interpret various kinds of application oriented problems faced in chemical industries using analytical techniques 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Provide a detailed insight of all the chemicals derived from petroleum 2. Explain the different methods for the conversion of monomer to polymers 3. Distinguish different type of polymers for specific application 4. Develop familiarity with major polymerization processes on industrial scale 5. Understand the different process technologies for Elastomers and resins 6. Demonstrate the manufacture of Plastics, Fibres and their applications 						
Module:1	Petrochemical & Precursors	2 hours				
Introduction; Petrochemical & its Precursors						
Module:2	Alkanes & Alkenes	7 hours				
Introduction to Alkanes and Alkenes; Manufacture of Petrochemical Derivatives from C ₁ , C ₂ , C ₃ , C ₄ compounds						
Module:3	Aromatics	6 hours				
Introduction to Aromatics; Manufacture of Petrochemical Derivatives from – Benzene, Toluene, Xylene, Styrene						
Module:4	Alternate Route and its Derivatives	8 hours				
Manufacture of VCM by thermal cracking, DMT , PTA, maleic anhydride, cumene, diphenyl carbonate.						
Module:5	Polymers	8 hours				
Production of - poly butadiene rubber, SBR, SAN, Polyalkylene Terephthalate, Alpha Olefins (Linear), Octenes.						
Module:6	Plastics & Fibres	7 hours				
Production of – Polyacrylonitrile resins, Melamine, formaldehyde resins, SNG, explosives, dyes						
Module:7	Economics of Petrochemical Industry	5 hours				
Current status in India; Trade; Selection of Petrochemical products; Economics of Petrochemical derivatives and Industry						
Module:8	Contemporary issues	2 hours				

Total Lecture hours		45 hours	
Text Books			
1.	Mall I.D., Petrochemical Process Technology, 2 nd ed., Macmillan Petroleum Chemicals Ltd, UK, 2011.		
2.	Chaudhuri U.R., Fundamentals of Petroleum and Petrochemical Engineering, 3 rd ed., CRC Press, USA, 2011.		
Reference Books			
1.	Richard A. Dawe, Modern petroleum technology, 6 th ed., John Wiley & Sons Limited, USA, 2012.		
2.	Abdulin F., Production of Oil & Gas, 2 nd ed., Mir publishers, Russia, 2014.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Course code	CHE1016	L	T	P	J	C
Course title	FERMENTATION TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Learn the basics of the various aspects of microbiology and biosystems 2. Impart experimental design thinking capability in relation to various fermenter configurations, modes of operation, growth kinetics and product recovery 3. Extrapolate the design thinking skills to bio-related processes with chemical engineering background 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the importance of fermentation with reference to industrial microbiology 2. Summarize kinetics prevalent in microbial processes 3. Understand the process to select and manage microorganisms from natural source to fermentation 4. Interpret the acquired knowledge on fermenter configuration for different types of cells and enzymes 5. Design of fermenter and the downstream processing of fermentation products 6. Create innovative applications for fermentation technologies for novel products 						
Module:1	Introduction and history of fermentation processes	4 hours				
Development of fermentation process – range of processes under fermentation, Types of fermentation.						
Module:2	Microbial growth kinetics	6 hours				
Microbial growth- Batch, Continuous and types of fed batch culture – design and kinetics. Comparison of the modes of culture						
Module:3	Microbial Strain Management	5 hours				
Industrial microorganisms-isolation, preservation and improvement of strains; Storage methods and improvement strategies.						
Module:4	Media for industrial fermentations	5 hours				
Media formulation- energy, carbon and nitrogen sources, micronutrients; oxygen requirements; Other non-nutrient and functional components. Effects of media composition on penicillin production; Media						
Module:5	Preparation of aseptic fermentation process	8 hours				
Preparation of media and air for pure culture fermentation; Media sterilization- Batch and continuous sterilization processes; Sterilization of fibrous filters and their design; Development of inoculation processes involving yeast, bacterial, fungi; Aseptic inoculation of plant fermentations.						
Module:6	Basic functions of a fermenter	8 hours				

Basic functions of fermenter – Aeration and agitation – process requirements and mechanical design aspects; Maintenance of aseptic conditions and foam control. Types of fermenters for industrial applications – stirred & sparged tanks fermenters, Tower fermenter, Packed tower, Airlift and rotating disc			
Module:7	Process technology for bulk products		7 hours
Basic downstream processing; Process technology for bulk products; Production of alcohols, organic acids, enzymes, and antibiotics – flowsheet and process description of modern processes.			
Module:8	Contemporary issues		2 hours
Total Lecture hours			45 hours
Text Books			
1.	Stanbury P.F., Whitaker A., Steve H., Principles of Fermentation Technology, 3 rd ed., Butterworth-Heinemann, USA, 2017.		
2.	El-Manşi E., Bryce C.F.A., Arnold L.D., Allman A.R., Fermentation Microbiology and Biotechnology		
Reference Books			
1.	Ashok P., Christian L., Carlos R.S., Advances in Fermentation Technology, 1 st ed., Asiatech Publishers Inc., India, 2008.		
2.	Rhodes A and Pletcher D.L.: Principles of Industrial Microbiology, 3 rd ed., Pergamon Press, UK, 1977.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1017	L	T	P	J	C
Course title	FOOD PROCESSENGINEERING	2	0	0	4	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Emphasize on the basic concepts of unit operations and unit processes in Chemical Engineering with an application to Food technology 2. Impart necessary knowledge required for food processing technology, food quality management, food standards and packaging 3. Familiarize the various properties of the raw material used in food processing and technologies required in transforming them into quality food products and to train the students to use the material handling equipment involved in food processing operations 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Determine the various engineering properties of the raw material used in food processing useful to design the various food Processing equipments 2. Select suitable dryers based on technical and economical point of view 3. Understand the knowledge in different food processing operations involved in various food manufacturing process 4. Identify and transform different processing technology to produce quality food products 5. Apply appropriate unit operations involved in food technology 6. Understand about the packaging material and methods and the cost involved 						
Module:1	Introduction to Food	4 hours				
Macromolecules-proteins, Enzymes, Carbohydrates, Micronutrients, Water, Interactions						
Module:2	Food Microbiology	3 hours				
Deteriorative factors and Control. Food additives and preservatives. Adulteration						
Module:3	Food process calculations	3 hours				
Material and energy calculations in food processing						
Module:4	Unit operations in food processing	5 hours				
Material handling, heat transfer, mixing, size reduction, mechanical separations						
Module:5	Food Preservation Techniques	5 hours				
Drying and dehydration, Irradiation, Microwave Heating, Sterilization and Pasteurization – Cleaning/sanitation In Process (CIP and SIP), Fermentation and Pickling						
Module:6	Food Processing and Food quality	5 hours				
Processing of Cereal Grains, Pulses, Vegetables, Fruits, Spices, Fats and Oils, Bakery, Confectionary and Chocolate Products Soft and Alcoholic Beverages, Dairy Products, Meat, Poultry and Fish Products, Food quality parameters and their evaluation FSSAI and safety concepts in food processing. Quality control and Food standard organizations						
Module:7	Packaging and canning	3 hours				

Concepts, definition, Significance, classification – fresh and processed; Basic packaging materials, types of packaging, Packaging methods. Newer methods of thermal processing, batch and continuous; application of infrared, microwaves. packaging design, retort pouch packing, vacuum packaging; cost of packaging and recycling of materials and Labelling			
Module:8	Contemporary issues		2 hours
Total Lecture hours		30 hours	
Text Books			
1.	Rao C.G., Essentials of Food Processing Engineering, 1 st ed., BSPublications, India, 2005.		
2.	Subbulakshmi G, Udipi Shobha A., Food Processing and Preservation, 1 st ed., New Age International, India, 2017.		
Reference Books			
1.	Khetarpaul N., Food Processing and Preservation, 1 st ed., Daya Publications, India, 2005.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1018	L	T	P	J	C
Course title	MEMBRANE SEPARATIONS TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.21				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand basic principles of membrane separation and characterization methods available for membranes 2. Derive various transport mechanisms involved in MF, UF, NF, RO and gas separation membranes 3. Select membranes for different industrial separation and purification application 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the basic principle of membrane separation processes 2. Describe different techniques available for membrane characterization 3. Derive various transport models for membrane flux and concentration polarization for various membrane systems 4. Compute membrane flux, concentration polarization and fouling using various transport models for various membrane systems 5. Analyze a membrane process and design components to carry out a specific separation 6. Select membranes for gas and bio separation application 						
Module:1	Membrane Materials, Preparation and Characterization	5 hours				
Introduction–Historical development of membranes–types of membrane processes–types of synthetic membranes–membrane materials–membrane module; Membrane preparation–Phase inversion process–casting methods; Membrane characterization–Measurement of pore size–solute properties–visual methods–bubble point method–liquid displacement method, molecular weight cut-off (MWCO), microbial challenge test						
Module:2	Membrane Transport Theory	6 hours				
Membrane transport theory– Introduction, solution-diffusion model; Structure–permeability relationship in solution diffusion membranes; Pore-flow membranes.						
Module:3	Concentration Polarization	6 hours				
Concentration polarization– Introduction, boundary layer film model; Concentration polarization in liquid separation process; Cross-flow, co-flow and counter-flow processes.						
Module:4	Microfiltration and Ultrafiltration	6 hours				
Microfiltration: Introduction and history, applications; Recent trends and progress in MF/UF technology; Ultra filtration: Introduction and history– characterization of ultrafiltration membranes– concentration polarization and membrane fouling, membrane cleaning – membrane and modules– system design – application						
Module:5	Nanofiltration	7 hours				
Nanofiltration: Introduction– process principles – application of nanofiltration for the production of drinking water and process water – solvent resistant nanofiltration						

Module:6	Reverse Osmosis	7 hours
Reverse osmosis:Introduction– membrane categories– membraneselectivity– membranetransportconcentration polarization –membranemodules– membrane fouling control–membrane cleaningapplications		
Module:7	Recent development in MembraneProcesses	6 hours
Recent material and moduleconfigurations for Microfiltrationand ultrafiltration; Thinfilmcompositemembranes– Biofouling protection; Integrated membranessystems;Gas separation -Hydrogen separation– oxygen and oxygen enriched air;Membranedistillationand Ceramicmembranes		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	DuttaB.K.,PrinciplesofMassTransferandSeparationProcesses,1 st ed.,PrenticeHallofIndia, India,2007.	
2.	MulderM.,BasicPrinciplesofMembraneTechnology,2 nd ed.,SpringerScience,USA,1991.	
Reference Books		
1.	Kaushik K.N.,MembraneSeparation Process, 1 st ed., Prentice Hall of India, India, 2008.	
2.	Cui Z.F., MuralidharaH.S.,Membrane Technology: A Practical Guide to MembraneTechnologyand Applications in Food and Bioprocessing, 1 st ed., Elsevier,USA,	
Mode of evaluation:ContinuousAssessment Test, Quizzes,Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by AcademicCouncil	55 th	Date 13-06-2019

Course code	CHE1019	L	T	P	J	C
Course title	POLYMER TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand different types of polymers 2. Identify the various technologies and types of polymerization techniques 3. Analyze the polymer processing techniques and polymer additives 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Classify various polymer and polymerization reactions 2. Explain the different methods of polymerization 3. Identify the processing technologies for different polymer synthesis and their additives 4. Explain various polymer processing techniques 5. Distinguish different type of polymers for various applications 6. Demonstrate the novel biodegradable polymers and their applications 						
Module:1	Introduction to polymer					5 hours
Monomer; polymers and their classification: Degree of polymerization. Polymeric reaction: addition; condensation and copolymerization						
Module:2	Methods of polymerization					6 hours
Bulk, solution, emulsion and suspension polymerization						
Module:3	Structure and size of polymer					6 hours
Structure of polymers, Characterization of polymers: Molecular weight, Crystallinity, Glass transition temperature and mechanical properties: testing of polymers						
Module:4	Polymer processing additives					6 hours
Fillers, plasticizers, Anti-oxidants, colorants, stabilizers, and other related additives						
Module:5	Polymer processing techniques					6 hours
Injection and compression transfer moulding methods; calendaring, extrusion, thermoforming, powder coating						
Module:6	Polymeric materials					9 hours
Polyethylene; polypropylene; polymethyl methacrylate; polyvinyl chloride; polytetrafluoroethylene, polyacrylate, polyesters; Polymeric foams – Polyurethane, polystyrene.						
Module:7	Special polymers and bio polymers					5 hours
Polycarbonates, polysulphones; aromatic polyamides; aromatic polyester; photo conductive polymers; wool silk and cellulose derivatives, Protein based polymers and Bio-nano-composites						
Module:8	Contemporary issues					2 hours

Total Lecture hours		45 hours	
Text Books			
1.	Gowariker V.R., Viswanathan N.V., Sreedhar J., Polymer Science, 2 nd ed., New Age Publishers, India, 2015.		
2.	Ebewele R.O., Polymer Science and Technology, 1 st ed., CRC press, USA, 2000.		
Reference Books			
1.	Froed J.R., Polymer science & Technology, 1 st ed., Prentice Hall Publishers, USA, 2014.		
2.	Young R.J., Lovell P.A., Introduction to Polymers, 1 st ed., CRC Press, USA, 2011.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Course code	CHE1020	L	T	P	J	C
Course title	FERTILIZER TECHNOLOGY	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Introduce production of various NPK fertilizers and their importance 2. Impart knowledge of bio fertilizers, fluid fertilizers and controlled release fertilizers 3. Identify pollutions involved in fertilizer manufacture and their controlling strategies to maintain the pollution standards 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Analyze the role of essential elements for plant growth and the need of nitrogenous, phosphate and potash fertilizers 2. Identify reactions and unit operations involved in the manufacturing of various fertilizers 3. Outline various physical and chemical properties of fertilizers 4. Categorize the major engineering problems associated in fertilizer manufacturing process 5. Explain the importance of bio fertilizers, fluid fertilizers and controlled release fertilizer 6. Analyze the impact of pollution from fertilizer industry based on pollution standards 						
Module:1	Introduction to Fertilizers	7 hours				
Introduction to fertilizers- Importance, Feedstocks for the production of Ammonia. Processes for gasification of fossil fuel and						
Module:2	Nitrogenous Fertilizers	7 hours				
Nitrogenous fertilizers – Ammonium sulphate, Urea, Ammonium chloride, Ammonium nitrate and Calcium ammonium nitrate, Their methods of production, Characteristics and specification, Storage and handling						
Module:3	Phosphatic Fertilizers	5 hours				
Phosphatic Fertilizers: Raw materials – phosphatic rock, sulphur, pyrites etc. Processes for the production of sulphuric and phosphoric acids. Phosphatic fertilizers – ground rock phosphate, bonemeal – Single superphosphate, Triple superphosphate, thermal phosphates –						
Module:4	Potassic Fertilizers	5 hours				
Potassic fertilizers – Potassium Chloride, Potassium sulphate, Potassium magnesium sulphate, Potassium hydroxide, Potassium nitrate – Methods of production: their characteristics and specifications.						
Module:5	NPK Fertilizers	7 hours				
NPK fertilizers: Urea ammonium phosphate, ammonium phosphate sulphate, Nitro phosphates, and various grades of NPK fertilizers produced in the country						
Module:6	Other Fertilizer	7 hours				
Fertilizers and granulated mixtures; Biofertilisers, Nutrient-Secondary nutrients and micronutrients; Fluid fertilizers, Granular fertilizers, Controlled release fertilizers, Slow release fertilizers						

Module:7	Pollution control	5 hours	
Pollution from fertilizer industry, Solid, liquid and gaseous pollution control and standards			
Module:8	Contemporary issues	2 hours	
Total Lecture hours		45 hours	
Text Book			
1.	Handbook of fertilizer technology, Association of India, New Delhi, 1977		
2.	Fertilizer Manual, United Nations Industrial Development Organization, United Nations, New York, 1967.		
Reference Books			
1.	Rao G., Sittig M., Dryden's Outlines of Chemical Technology, 3 rd ed., East West Press, India, 2010.		
2.	Austin T.G., Shreve's Chemical Process Industries, 5 th ed., Tata McGraw-Hill Education Pvt. Ltd, USA, 2012.		
3.	Shukla S.D., Pandey G.N., A Text Book of Chemical Technology, 1 st ed., Vikas Publishing House Pvt. Ltd, India, 1978.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE1023	L	T	P	J	C
Course title	PRODUCTION AND OPERATIONS MANAGEMENT	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Develop the student understanding levels of product and process layout fundamentals 2. Apply the knowledge of statistics for performing quality control and Inspection and project planning 3. Make the students to analyze situations and use different models for decision making 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Explain the concepts of production and operations 2. Design the product and process layout 3. Evaluate the material inventory and manage the supply 4. Examine the quality control and Inspection using statistical tools 5. Develop Gantt chart, and conduct project evaluation and review 6. Analyze situations and use different models for decision making 						
Module:1	Introduction to Production and Operations Management	6 hours				
Production system, production management; Operating system, operations management – classifications, objectives and scope						
Module:2	Plant Location and Layout	6 hours				
Factors influencing plant location – location models; Plant layout – objectives, classifications; Design of product and process layout.						
Module:3	Supply of Resources	6 hours				
Materials Management - purchasing; ABC Analysis						
Module:4	Inventory Management/Control	6 hours				
Inventory Management – objectives, benefit, technique; Inventory models – without shortage, with shortage						
Module:5	Quality Control and Inspection	6 hours				
Statistical Quality Control Methods - p, x and R charts etc.,						
Module:6	Project Planning	7 hours				
Scheduling models – Gantt chart; Priority decision rule, Network Models, PERT, CPM						
Module:7	Decision Making	6 hours				
General Model for decision making – Bayes' Decision Rule; Decision Making under Uncertainty and Risk; Decision Tree Method						
Module:8	Contemporary issues:	2 hours				

Total Lecture hours		45 hours	
Text Books			
1.	Chary S.N., Production and Operations Management, 5 th ed., Tata McGraw-Hill Education Pvt. Ltd., India, 2012		
2.	Panneerselvam R., Production and Operations Management, 3 rd ed., PHI Learning Pvt. Ltd., India, 2012		
Reference Books			
1.	Garg, A.K., Production and Operations Management, 1 st ed., Tata McGraw-Hill Education Pvt. Ltd., India, 2012		
2.	Montgomery, D.C., Introduction to Statistical Quality Control, 6 th ed., John Wiley & Sons, Inc. USA, 2009		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Course code	CHE2003	L	T	P	J	C
Course title	CHEMICAL PRODUCT DESIGN	3	0	0	0	3
Pre-requisite	CHE1004	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Train the students in identifying the needs and converting needs to product specifications 2. Facilitate generation of innovative ideas for chemical products and select among the ideas 3. Familiarize the student with intellectual property issues and manufacture and design of specialty products 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Analyze the needs of customers 2. Apply engineering knowledge to convert needs to product specifications 3. Create innovative ideas for products 4. Evaluate and select among ideas 5. Analyze the manufacture of products 6. Design better marketable products 						
Module:1	Introduction					1 hour
Introduction to chemical product design						
Module:2	Needs of chemical product					6 hours
Customer needs- consumer products						
Module:3	Needs to specifications					6 hours
Converting needs to specifications -revising product specifications						
Module:4	Ideas					8 hours
Human sources of ideas - chemical sources of ideas- sorting the ideas - screening the ideas.						
Module:5	Selection of ideas					8 hours
Selection using thermodynamics- selection using kinetics-less objective criteria-rise in product selection						
Module:6	Product manufacture					6 hours
Intellectual property - supplying missing information- final specifications- microstructured products - device manufacture						
Module:7	Specialty chemical manufacture and Economic Concerns					8 hours
First step towards production-separations- specialty scale up-Product versus process design-process economics- economics for products						
Module:8	Contemporary issues					2 hours
Total Lecture hours						45 hours

Text Books			
1.	Cussler E.L., Moggridge G.D., Chemical Product Design, Cambridge University Press, 2 nd ed., UK, 2011.		
Reference Books			
1.	Seider W.D., Seader J.D., Lewin D.R., Product and Process Design Principles, Wiley, 4 th ed., USA, 2016.		
2.	Wei J., Product Engineering: Molecular Structure and Properties, Oxford University Press, 1 st ed., UK, 2007.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Course code	CHE2006	L	T	P	J	C
Course title	FUELS AND COMBUSTION	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Develop the understanding levels of fuels and combustion fundamentals 2. Classify and introduce different types of fuel and fuel analysis techniques that assist the students to choose most convenient fuel for a process involving combustion` 3. Engage the students in designing various control techniques for handling various environmental issues resulting from combustion of fuels 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Classify the various types of fuels and their properties 2. Interpret the results of proximate and ultimate analysis of fuels 3. Identify the suitable fuel based on various factors such as availability, storage, handling, pollution and cost 4. Evaluate the properties of flue gases 5. Execute basic engineering and science concepts for the design of various combustion equipment 6. Interpret various air pollution controlling techniques for reducing the pollution generated from combustion of various fuels 						
Module:1	Classification and Properties of Fuels	5 hours				
Fuels-Types and characteristics of fuels-Determination of properties of fuels-Fuel analysis-Proximate and ultimate analysis-Calorific value (CV)-Gross and net calorific values (GCV, NCV)-Bomb Calorimetry-empirical equations for CV estimation						
Module:2	Solid Fuels	6 hours				
Origin of coal-Ranking of coal-Washing, cleaning and storage of coal-Renewable Solid Fuels-comparative study of Solid, liquid and gaseous fuels-selection of coal for different industrial applications-carbonization of coal						
Module:3	Liquid fuels	6 hours				
Origin of crude oil-composition of crude petroleum-classification of crude petroleum-Removal of salt from crude oil-processing of crude petroleum-Fractionation distillation-ADU and VDU-Cracking-Hydrotreatment and Reforming						
Module:4	Gaseous fuels	6 hours				
Rich and lean gas-Wobbe index-Natural gas-Dry and wet natural gas-Fouls and sweet NG-LPG-LNG-CNG-Methane-Producer Gas-Water gas-Coal Gasification-Gasification Efficiency						
Module:5	Combustion	7 hours				
General principles of combustion-types of combustion processes-Combustion chemistry-Combustion equations-Kinetics of combustion-combustion of solid fuels-Combustion calculations-air fuel ratio-Excess air calculations						
Module:6	Combustion Equipment	7 hours				

Analysis of flue gases by Orsat apparatus-Combustion of solid fuels-grate firing and pulverized fuel firing system-Fluidized bed combustion-Circulating fluidized bed boiler-Burners-Factors affecting burners and combustion			
Module:7		Air Pollution	6 hours
Types of pollution-Combustion generated air pollution-Effects of air pollution-Pollution of fossil fuels and its control-Pollution from automobiles and its control			
Module:8		Contemporary issues	2 hours
Total Lecture hours			45 hours
Text Books			
1.	Kenneth K.K., Principles of Combustion, 2 nd ed., Wiley Publications, USA, 2012		
2.	Phillips H.J., Fuels-solid, liquid and gases-Their analysis and valuation, 1 st ed., Foster Press, USA, 2010		
Reference Books			
1.	Speight J.G., The Chemistry and Technology of Coal, 3 rd ed., Taylor and Francis Ltd., USA, 2016		
2.	Sarkar S., Fuels and combustion, 3 rd ed., Universities Press, India, 2009		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Coursecode	CHE2007	L	T	P	J	C
Course title	PROCESS INTENSIFICATION	3	0	0	0	3
Pre-requisite	CHE1006	Syllabus version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. Understand the concept of Process Intensification 2. Apply the techniques of intensification to a range of chemical processes 3. Infer alternative solutions keeping in view point, the environmental protection, economic viability and social acceptance 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the scientific background, techniques and applications of intensification in the process industries 2. Apply process intensification in industrial processes 3. Implement methodologies for process intensification 4. Identify scale up issues in the chemical process 5. Interpret the feasibility of the process intensification 6. Formulate and solve process challenges using intensification technologies 						
Module:1	Introduction	6 hours				
Techniques of Process Intensification (PI) Applications - The philosophy and opportunities of Process Intensification - Main benefits from process intensification - Process intensifying Equipment - Process intensification toolbox - Techniques for						
Module:2	Process Intensification Through Micro Reaction Technology	6 hours				
Effect of miniaturization on unit operations and reactions - Implementation of Micro reaction Technology from basic Properties - Technical Design Rules - Inherent Process Restrictions in Miniaturized Devices and Their Potential Solutions - Micro fabrication of Reaction and unit operation Devices - Wet and Dry Etching Processes						
Module:3	Mixing And Flow Patterns	8 hours				
Scales of mixing - Flow patterns in reactors - Mixing in stirred tanks: Scale up of mixing - Heat transfer - Mixing in intensified equipment - Chemical Processing in High gravity Fields Atomizer - Ultrasound Atomization - High intensity inline MIXERS reactors - Static mixers - Ejectors - Te mixers - Impinging jets - Rotor stator mixers - Design Principles of static Mixers and Applications of static mixers - Hige reactors						
Module:4	Combined Chemical Reactor Heat Exchangers And Reactor Separators	6 hours				
Principles of operation - Applications - Reactive absorption - Reactive distillation - Applications of RD Processes						
Module:5	Compact Heat Exchangers	8 hours				

Classification of compact heat exchangers-Plate heat exchangers-Spiral heat exchangers-Flow pattern-Heat transfer and pressure drop-Flat tube and fin heat exchangers-Microchannel heat exchangers-Phase change heat transfer-Selection of heat exchanger technology-Feed/effluent heat exchangers-Integrated heat exchangers in separation processes-Design of compact heat exchanger- examples			
Module:6		Enhanced Fields	6 hours
Energy based intensifications-Sonochemistry-Basics of cavitation-Cavitation Reactors-Flow over a rotating surface- Hydrodynamic cavitation applications-Cavitation reactor design-Nusselt flow model and mass transfer-Sonocrystallization; Reactive separations			
Module:7		Case Studies	3 hours
Reactive Extraction Case Studies- Absorption of NO _x -Coke Gas Purification			
Module:8		Contemporary issues	2 hours
Total Lecture hours			45 hours
Text Books			
1.	Segovia H., Juan G., Bonilla P., Adrián, Process Intensification in Chemical Engineering design optimization and control, 1 st ed., Springer, Mexico, 20		
2.	David R., Colin R., Adam H., Process Intensification Engineering for Efficiency, Sustainability and Flexibility, 2 nd ed., Elsevier, Netherlands, 2013.		
Reference Books			
1.	Andrzej S., Jacob A., Moulijn, Re-engineering the chemical processing plant: process intensification, 1 st ed., Marcel Dekker Inc, USA, 2004.		
2.	Reay D., Ramshaw C., Harvey A., Process Intensification, 1 st ed., Elsevier, Netherlands, 2008.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Coursecode	CHE2008	L	T	P	J	C
Course title	CHEMICAL ENGINEERING COMPUTATIONAL FLUID DYNAMICS	2	0	0	4	3
Pre-requisite	CHE1005, CHE1006	Syllabus version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. Recall the basic fluid and heat transfer governing equations 2. Utilize basic aspects of discretization for grid generation 3. Estimate fluid flow and heat transfer problems 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand and select the governing equations of fluid flow and heat transfer 2. Enable to solve one and two-dimensional ordinary and partial differential equations using traditional CFD tools 3. Make use of discretization techniques for derivatives and differential equations to solve numerically 4. Examine general transformation equations for grid generation 5. Recommend suitable explicit, implicit and semi-implicit methods of finite difference scheme for given problems 6. Solve fluid flow field and temperature field to design any process equipment using some popular CFD techniques 						
Module:1	Governing Equations	4 hours				
Navier-Stokes Equations; Complete energy equations-complete mass conservation equations; Parabolized Navier-Stokes Equations; Euler Equations						
Module:2	Conservation laws and forms of equations	5 hours				
Model of Flow – Conservation form Continuity; Momentum and Energy Equation in conservation form (differential equation only) - Characteristics of PDE's - Elliptic; parabolic and hyperbolic						
Module:3	Discretization	5 hours				
Finite Difference method - Forward; Backward and Central difference schemes - Finite volume - Finite element techniques						
Module:4	Grid generation	5 hours				
Choice of grid - grid oriented velocity components - Cartesian velocity components - Staggered and collocated arrangements						
Module:5	Convection and Diffusion	3 hours				
Steady one-dimensional convection and diffusion - Central difference; upwind, quick, exponential, hybrid and power law schemes - Fictitious diffusion, Simple algorithm						
Module:6	CFD Techniques	3 hours				
ADI Technique - Pressure correction Technique Simple algorithm						
Module:7	Case Study	3 hours				

Industrially important process equipment- Heat exchangers; Fluid flow; Mixing equipments; Cyclone separators			
Module:8	Contemporary issues		2 hours
Total Lecture hours		30 hours	
Text Books			
1.	Pletcher R.H., Tannehill J.C., Anderson D.A., Computational fluid mechanics and heat transfer, 3 rd ed., CRC Press, USA, 2012.		
2.	Aref H., Computational Fluid Dynamics, 1 st ed., Cambridge University Press, USA, 2017.		
Reference Books			
1.	Versteeg H.K., Malalasekera W., An introduction to computational fluid dynamics: The finite volume method, 2 nd ed., Prentice Hall, UK, 2007.		
2.	Hirsch C., Computation of internal and external flows: The fundamentals of computational fluid dynamics, 2 nd ed., Butterworth-Heinemann, USA, 2007.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Course code	CHE3004	L	T	P	J	C
Course title	HETEROGENEOUS REACTION ENGINEERING	2	0	0	4	3
Pre-requisite	CHE2001	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1 Introduce students about catalytic phenomena with an extension to reactor design and catalyst characterization 2. Build upon the fundamentals of heterogeneous reactions, design, and analysis of non-catalytic, catalytic fluid-solid reactors including multi-phase reactors 3. Engage students in handling most common industrial chemical and biochemical reactors to achieve production goals for processes involving homogeneous or heterogeneous reaction systems 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the heterogeneous reaction systems and design the reactors for fluid-solid systems 2. Analyze the mechanism of non-catalytic solid-fluid reactions 3. Analyze the role of catalyst in reactions and the transport mechanism in heterogeneous catalysts 4. Design and characterize catalyst surface properties for better activation of the catalyst 5. Identify critical parameters affecting the performance of heterogeneous and multi-phase reactors 6. Construct and apply a general problem solving approach to design heterogeneous and multiphase reactors 						
Module:1	Introduction to Heterogeneous Reaction Engineering	2 hours				
Introduction to heterogeneous reacting systems - Sharp interface and volume reaction models - determination of rate-controlling steps and application to design of reactors - bio reactors						
Module:2	Non-catalytic solid-fluid reactions	4 hours				
Shrinking core model – Gas film controlling – Ash layer controlling – Chemical reaction controlling – Shrinking spherical particles – Fluidized bed reactor						
Module:3	Introduction to Catalytic Reactions	4 hours				
Definition and properties - Steps involved in catalytic reactions - Rate laws mechanisms - Rate limiting step						
Module:4	Transport Mechanism in heterogeneous catalysts	5 hours				
Transport effects in heterogeneous catalysis - Internal effectiveness - External transport limitations and overall effectiveness						
Module:5	Catalysts preparation & characterization	4 hours				
Definition and types of catalysts – Industrial catalysts – Preparation and characterization of the catalysts, Surface area and pore volume determination						
Module:6	Catalyst deactivation methods	4 hours				
Types of catalyst deactivation – Determining the order of deactivation – Catalyst regeneration methods						
Module:7	Design of Reactors for Fluid-Liquid and Fluid-Solid reactions	5 hours				

Reactor design fundamentals and methodology, rate data analysis - Overall view of Fluidized, Packed and Moving bed reactors- Fluid-liquid reactions: Film and Penetration theories - Fluid-solid catalytic reactions			
Module:8	Contemporary issues		2 hours
Total Lecture hours			
			30 hours
Text Books			
1.	Levenspiel O., Chemical Reaction Engineering, 3 rd ed., Wiley Publications, USA, 2006		
2.	Fogler H.S., Elements of Chemical Reaction Engineering, 5 th ed., Prentice Hall India Pvt. Ltd., India, 2016		
Reference Books			
1.	Miller, G. T., Chemical Reaction Engineering, 1 st ed., CRS publications, USA, 2016		
2.	Vannice, M. A., Kinetics of Catalytic Reactions. 2nd ed., Springer, USA, 2010		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Coursecode	CHE3005	L	T	P	J	C
Course title	CHEMICAL PROCESS INTEGRATION	3	0	0	0	3
Pre-requisite	CHE2002	Syllabus version				
		v.2.1				
Course Objectives:						
<ul style="list-style-type: none"> To impart the strategy for effective and energy efficient way of fine tuning the process design so as to maximize profit by minimizing utilities. To provide an insight on the utilities with their standards and conservation aspects. To cater the fundamental knowledge about energy efficient chemical process design. 						
Expected Course Outcome:						
<ol style="list-style-type: none"> Understand the scientific background, techniques and applications of intensification in the process industries Apply process intensification in industrial processes Implement methodologies for process intensification Identify scale up issues in the chemical process Interpret the feasibility of the process intensification Formulate and solve process challenges using intensification technologies 						
Module:1	Distillation Sequencing	7 hours				
Need for column sequencing, distillation sequencing for simple columns, columns with more than two products, columns using thermal coupling, distillation sequencing for azeotropic distillation						
Module:2	Heat Exchanger Networks – Targets	6 hours				
Composite curve, problem table algorithm, network energy targets, heat exchange area targets, number of shell targets, capital cost and total cost targets						
Module:3	Heat Exchanger Networks – Network Design	6 hours				
Pinch Design method, stream splitting, design of multiple pinches, superstructure approach, tradeoffs, network stream data.						
Module:4	Heat Integration – Process Equipment	6 hours				
Use of composite curve for integration of reactors, heat integration in distillation columns and distillation sequences.						
Module:5	Energy consuming – Process Utilities	6 hours				
Heat integration of evaporator, heat integration of dryers, heat integration of boilers, heat integration of compressors and refrigerator						
Module:6	Water system design	6 hours				
Water use in process industries, design for maximum water reuse, design for minimum wastewater treatment flow rate, targeting and design for effluent treatment and regeneration						
Module:7	Clean process technology	6 hours				
Sources of Waste from Chemical Production. Clean Process Technology for Chemical Reactors, Separation and Recycle Systems, Process Operations and Process utilities. Life Cycle Analysis						
Module:8	Contemporary issues	2 hours				

	TotalLecturehours:	45hours
TextBook(s)		
1.	RobinSmith,“ChemicalProcess:DesignandIntegration”,JohnWileyandSons,2016	
2.	RichardTurton, RichardC.Bailie,WallaceB. Whiting,JosephA. Shaeiwitz,DebangsuBhattach aryya,“Analysis,SynthesisandDesignofChemicalProcesses”,PrenticeHall,2012.	
ReferenceBooks		
1.	AlexandreC.Dimian, CostinS.Bildea, AntonA. Kiss,“Integrated Designand Simulationof ChemicalProcesses”,ElsevierPublications,2014.	
2	Ian C. Kemp, “Pinch Analysis and Process Integration: A User Guide on ProcessIntegration”,Elsevier Publications, 2008	
ModeofEvaluation:ContinuousAssessmentTest,Quizzes,Assignments,FinalAssessmentTest		
RecommendedbyBoardofStudies	04.03.2016	
ApprovedbyAcademicCouncil	40	Date 18.03.2016

Coursecode	CHE3006	L	T	P	J	C
Course title	PROCESS PLANTS SIMULATION	3	0	0	4	4
Pre-requisite	MAT3003	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Emphasize the basic concepts of steady state process plants simulation 2. Impart the knowledge and awareness to understand the validity and physicochemical interpretation of thermodynamic models and their limitations 3. Develop the skills for plant simulation and optimization, solve 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the principles for developing a Process flow sheet and its execution 2. Examine the approaches to follow in plant simulation 3. Overcome the bottleneck existing in process plant and have maximum productivity 4. Implement the strategies for solving simple and complex plant problems 5. Utilize commercial software's for simulation of chemical processes 6. Interpret steady state process plant simulation 						
Module:1	Introduction	5 hours				
Introduction to Process Synthesis- Flowsheeting & simulation- Degrees of freedom- Process Equipment's - Process flow sheet						
Module:2	Approaches to Process Simulation	6 hours				
Sequential modular approach and Simultaneous modular approaches- Equations solving approach used in process plants simulation						
Module:3	Equation Solving Approach	8 hours				
Partitioning- Decomposition- Disjointing- PTM- SWS- Steward- Rudd Algorithms; Sparsity- Direct Methods - Pivoting- Iterative methods - BTF- BBTB Block Back Substitution- BTS - etc						
Module:4	Decomposition of Networks	7 hours				
Tearing Algorithms in decomposition of networks- digraph- signal flow graph- BMA algorithm- BTA - K&S algorithm- M&H algorithm- related problems						
Module:5	Convergence Promotion	6 hours				
Linear equation - nonlinear equation - Convergence promotion scheme Newton's method - Direct Substitution- Wegstein's method- Dominant eigenvalue method- Quasi-Newton methods; Acceleration criterion						
Module:6	Application of Flow Sheet Software	5 hours				
Flow sheeting software: Aspen Plus- Steady state simulation- Aspen Hysys dynamic simulation						
Module:7	Case Studies: (Un) Steady State Process Simulation	6 hours				
Complete plant (un) steady state simulation: Any process such as Ammonia plant- Biodiesel plant - NG liquefaction						
Module:8	Contemporary issues	2 hours				

Total Lecture hours		45 hours	
Text Books			
1.	Robin S., Chemical Process Design and Integration, 2 nd ed., Wiley, USA, 2016.		
2.	Babu B. V., Process Plant Simulation, 1 st ed., Oxford University Press, India, 2004.		
Reference Books			
1.	Westerberg A. W., Hutchison H. P., Motard R. L., Winter P., Process Flowsheeting, 1 st ed., Cambridge Press, UK, 2011.		
2.	Richard T., Analysis, Synthesis and Design of Chemical Processes, 1 st ed., Pearson Education International, USA, 2009.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Coursecode	CHE3007	L	T	P	J	C
Course title	MULTIPHASE FLOW	3	0	0	0	3
Pre-requisite	CHE1005, CHE1006	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Emphasis the concepts of multiphase systems in the processing industry 2. Formulate momentum, energy and material balance models in multiphase systems 3. Develop design thinking skills to understand multiphase flows in chemical industries 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Define the basic types of multiphase systems in process industries 2. Identify the type of flow-pattern and flow regimes for fluid-fluid (gas-liquid and liquid-liquid) and fluid-solids systems 3. Construct one dimensional Steady state models in multiphase flows 4. Interpret Drift Flux models two phase system 5. Formulate and estimate flow properties for phase change systems 6. Design and fabricate the columns to handle for multiphase system in chemical engineering operations 						
Module:1	Introduction to multiphase flow, type of flow and applications	9 hours				
Basic fluid flow concepts: Flow field description – conservation laws – viscous flow – turbulent flow – pressure drop - Review of Single Phase Flow; Scope and significance - applications						
Module:2	Flow pattern maps and Regime	11 hours				
Flow patterns for gas-liquid; gas-solid; liquid-liquid; liquid-solids system; Heated tubes – horizontal – vertical – Vertical flow; horizontal flow; co-current; countercurrent systems; Gas-liquid-solid three phase flows						
Module:3	One dimensional steady state flow	9 hours				
Definitions and common Terminologies – simple analytical model – homogenous flow model.						
Module:4	Drift flux model	4 hours				
Theory of drift flux model and its application						
Module:5	Separated flow model	4 hours				
Separated flow model for stratified and annular flow; Correction factor and analysis.						
Module:6	Two phase flow with phase change	4 hours				
Boiling flow heat transfer – regimes – bubble growth						
Module:7	Measurement techniques	2 hours				
Sampling Methods – Integral Methods – Local Measurement techniques – hold up studies – analysis						
Module:8	Contemporary issues	2 hours				

Total Lecture hours		45 hours	
Text Books			
1.	Wallis, One Dimensional Two-phase flow, McGraw Hill Book Company, 1 st ed., USA, 2000.		
2.	John G.C., John R.T., Convective Boiling and Condensation, Oxford University Press, 3 rd ed., UK, 2002.		
Reference Books			
1.	Clement K.S., Two Phase Flow – Theory and Applications, 1 st ed., Taylor and Francis, USA, 2003.		
2.	Govier, G.W., Aziz K., The Flow of Complex Mixture in Pipes, 2 nd ed., Society of Petroleum Engineers Publishers, USA, 2008.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

SCHEME 2020

Coursecode	CHE3008	L	T	P	J	C
Course title	INDUSTRIAL POLLUTION ENGINEERING	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Explain the legislation and standards related to air, water and solid wastes in Indian context 2. Identify and design treatment equipments for air and water pollution 3. Illustrate the effective methods of solids waste treatment techniques 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand basics of pollution parameters and characteristics of industrial wastes 2. Distinguish types of standards and legislations and resource optimization methods 3. Categorize sources, types, and control equipment's for industrial air pollution 4. Design various equipment for wastewater treatment 5. Differentiate various solid waste disposal techniques 6. Evaluate waste treatment flow sheets of various process industries 						
Module:1	Introduction					5 hours
Types of industries- Characteristics of industrial wastes- Fundamental definition of pollution parameters- Effects of industrial pollutants on environment- air, water and land.						
Module:2	Standards and legislation					5 hours
Environmental legislations related to prevention and control of industrial effluents- EPA ACT- EIA -EMP- ISO 14000 series- Combined treatment of industrial wastewater- Resource optimization through industrial symbiosis- waste minimization techniques.						
Module:3	Industrial air pollution control					7 hours
Air pollution meteorology (generation, transportation and dispersion of air pollutants)- Principles and design of air pollution control equipment: gravity settling chambers- air cyclones- ESPs- filters - wet scrubbers.						
Module:4	Industrial wastewater treatment					6 hours
Selection, design and performance analysis of industrial waste water treatment processes: Preliminary- Primary- Secondary treatment processes.						
Module:5	Advanced wastewater treatment					7 hours
Chemical oxidation- Ozonation- Photocatalysis- Wet Air Oxidation- Adsorption- Evaporation - Ion Exchange - Membrane Technologies.						
Module:6	Hazardous Solid waste management					5 hours
Classification of hazardous waste- waste disposal methods- Composting- Landfill- Briquetting- Gasification - Incineration.						
Module:7	Case studies					8 hours

Sources-Characteristics-WastetreatmentflowsheetsforselectedindustriessuchasTextiles-Tanneries-Pharmaceuticals-Electroplating-PulpandPaper-Refineries-Fertilizer-Thermalpowerplants - Wastewaterreclamationconcepts.			
Module:8	Contemporaryissues		2 hours
TotalLecturehours		45 hours	
Text Books			
1.	RaoC.S., EnvironmentalPollutionControl Engineering,3 rd ed.,NewAgeInternational Publishers,India, 2018.		
2.	KariaG.L.,ChristianR.A.,WastewaterTreatment:ConceptsandDesignApproach,2 nd ed., Eastern EconomyEdition,India,2013.		
Reference Books			
1.	PollutionControlLawSeries:PCLS/02/2010,CentralPollutionControlBoard, 6 th ed.,India, 2010.		
2.	TchobanoglousG.,TheisenH.,VigilS.A.,IntegratedSolid WasteManagement,1 st ed., McGrawHill Education,India, 2014.		
3.	BhatiaS.C.,EnvironmentalPollutionandControlinChemicalProcessIndustries,2 nd ed., Khanna publishers,India, 2013.		
Mode ofevaluation: Continuous AssessmentTest, Quizzes, Assignments,FinalAssessmentTest			
RecommendedbyBoardof Studies		15-04-2019	
Approved by AcademicCouncil		55 th	Date 13-06-2019

Coursecode	CHE3010	L	T	P	J	C
Course title	COLLOIDS AND INTERFACIAL SCIENCE	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. Introduce the student to the theory of Colloids and Interface Phenomena 2. Emphasize the student to learn solution thermodynamics, stability of colloids, light scattering, capillary effects 3. Expose the importance of colloidal phenomena through real time examples 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand the concept of the origin of long-range, non-covalent colloidal forces (van der Waals, electrostatic, etc.) 2. Explain the link between liquid surface tension and contact angle, and demonstrate how certain experimental techniques can be used for the assessment of liquid surface tension 3. Apply the knowledge of thermodynamics for micellization in surfactant solutions 4. Describe the thermodynamics of emulsion formation and calculate the kinetic and thermodynamic stability of such emulsions 5. Design colloidal systems or engineered surfaces of high industrial or technological interest 6. Explain the interactions between colloids and visible light, as well as the principles of static and dynamic light scattering 						
Module:1	Introduction to Colloid & Interface Science	6 hours				
Fundamentals of Colloid Science- Colloids: definition- Vander Waals interactions- The Hamaker constant- Electrostatic Interactions in Colloids- The electrical double layer (EDL)- Zeta potential- Gibbs energy of electrostatic interactions						
Module:2	Surface Tension and Contact Angle	6 hours				
Surface tension of liquids- definition- Lewis Acid- Base interactions- Surface tension & contact angle- Measuring contact angles						
Module:3	Interactions at Interfaces	5 hours				
Surfactants I: definitions and applications- Surfactants II: thermodynamics- Surface excess						
Module:4	Emulsions	6 hours				
Definitions and applications- Types of emulsions- Thermodynamics of emulsification- Emulsion stability						
Module:5	Design of Interfacial Science	7 hours				
Adsorption- Models of adsorption- Adsorption at the solid-liquid interface- Adsorption at the liquid-air interface- Adsorption at the solid-air interface						
Module:6	Principles of Light Scattering	6 hours				
Fundamentals of light scattering- Static light scattering- Dynamic light scattering						

Module:7	Application of Colloids and Interfacial phenomena	7 hours
Colloidal and interfacial phenomena in biology-Photovoltaic-Water treatment-Medicine-Tribology-Engineering		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	Wang C., Leblanc R.M., Recent Progress in Colloid and Surface Chemistry, 1 st ed., Oxford University Press Inc., UK, 2016	
2.	Birdi K.S., Handbook of Surface and Colloid Chemistry, 4 th ed., CRC Press., India, 2015	
Reference Books		
1.	Hiemenz P.C., Rajagopalan R., Principles of Colloid and Surface Chemistry, 3 rd ed., CRC Press, USA, 1997	
2.	Rhodes M., Introduction to Particle Technology, 2 nd ed., Wiley Publications, USA, 2008	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by Academic Council	55 th	Date 13-06-2019

Course code	CHE4002	L	T	P	J	C
Course title	TRANSPORT PHENOMENA	3	0	0	0	3
Pre-requisite	CHE1006, CHE3003	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Emphasize the basic concepts of transport phenomena, the similarities of the governing relations of momentum, heat, and mass transfer 2. Solve appropriate differential equations such as momentum, thermal energy, and mass species balance, accounting convective and diffusive (molecular-scale) fluxes, with sources and sinks to obtain velocity, temperature and concentration profiles 3. Develop design thinking skills to solve various kinds of application oriented problems faced in chemical industries using analytical techniques 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Understand transport properties of momentum, energy and mass transport 2. Analyze one-dimensional steady state momentum transfer, heat conduction and species diffusion problems 3. Examine the problems related to fluid, heat and mass transfer using Navier-Stoke's equation 4. Evaluate the interphase transport properties for internal flow and external flow 5. Understand simultaneous heat, mass and momentum transfer analysis 6. Formulate and solve industrial problems along with appropriate approximations and boundary conditions 						
Module:1	Transport by Molecular Motion	6 hours				
Phenomenological laws of transport properties-Newtonian and non-Newtonian fluids; Rheological models-theories of transport properties of gases and liquids-effect of pressure and temperature - Transport analogy						
Module:2	Vector and tensor analysis	2 hours				
Vector- coordinates system - time derivatives						
Module:3	1D Viscous Flow– Shell Balance	8 hours				
General method of shell balance approach to transfer problems; boundary conditions-rectilinear flow-curved flow-momentum flux and velocity distribution-Newtonian fluids–non-Newtonian fluids - pipe -annular flow						
Module:4	Equations of Change	8 hours				
Equation of Motion and Continuity-Integral Conservation Equations-Navier-Stokes and Euler Equation Constitutive relation - Dimensional analysis– Applications.						
Module:5	Turbulent Flow and Interphase momentum transfer	8 hours				
Turbulent models-RANS equation-Reynolds stresses; Internal flow-External flow-Boundary Layer Theory-Isothermal System-Flow through conduits-Empirical correlation–friction factor, drag coefficient-Ergun Equation - Flow through porous media						

Module:6	Heat Transfer by conduction and convection	6 hours
Shell Balance - Equations of energy -Heat Transfer coefficient- COMSOL Simulation		
Module:7	Mass Transfer	5 hours
Microscopic balances- General equations Boundary conditions- Mass transfer coefficient, Homogeneous reaction, Fixed bed catalytic reactor- steady state system.		
Module:8	Contemporary issues	2 hours
Total Lecture hours		45 hours
Text Books		
1.	Bird R.B., Stewart W.E., Lightfoot E.N., Transport Phenomena, 2 nd ed., John Wiley & Sons Inc., USA. 2012.	
2.	Wick C.E., Welty J., Wilson R.E., Fundamentals of Momentum, heat and Mass Transfer, 5 th ed., John Wiley & Sons Inc., USA, 2016.	
Reference Books		
1.	Thomson W.J., Introduction to Transport Phenomena, Pearson Education Asia, India, 2001.	
2.	William M. Dean, Analysis of Transport Phenomena, Oxford University Press, India, 2011.	
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test		
Recommended by Board of Studies	15-04-2019	
Approved by Academic Council	55 th	Date 13-06-2019

Coursecode	CHE4003	L	T	P	J	C
Course title	MODELLING AND SIMULATION IN PROCESS ENGINEERING	2	0	2	0	3
Pre-requisite	CHE3001	Syllabus version				
		2.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Explain the representation and simulation of physical systems using mathematical formulations 2. Develop the typical mathematical models for the chemical process industries 3. Enhance the skill of engineering software applications which illustrate a variety of modelling techniques 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Demonstrate the basic principles of chemical engineering for modeling of chemical system 2. Apply mathematical tools to solve model equations 3. Analyze the linear steady state and un-steady state lumped system of process industries 4. Develop the model equations for the Chemical Engineering system 5. Solve the model equation using simulation tools for various unit processes and unit operations 6. Execute the algorithm for different chemical engineering systems 						
Module:1	Modeling Conservative Principles and Models	4 hours				
Introduction of process modeling; definition of modelling and simulation; different types of models; application of mathematical modeling; Fundamental Laws – Continuity equation, energy equation, and equation of motion, transport equation, equation of state, phase and chemical equilibrium, chemical kinetics						
Module:2	Steady State Lumped Systems	4 hours				
Degree of freedom analysis; single and network of process units; systems yielding linear and non-linear algebraic equations; solution of linear and non-linear algebraic equations						
Module:3	Flow Sheetting and Process design	4 hours				
Steady state flow sheetting; approach to flow sheetting systems; introduction to sequential modular approach; simultaneous modular approach and equation solving approach; nested inside-out algorithms						
Module:4	Unsteady State Lumped Systems	4 hours				
Microscopic balances for Unsteady state and dynamic simulation – liquid level tank – gravity flow tank – jacketed stirred tank heater; Isothermal and Non-isothermal reactors – flash and distillation column; Solution of ODE initial value problems						
Module:5	Dynamic Simulation of Unsteady State Lumped Systems	5 hours				
Solution of ODE initial value problems; matrix differential equations; simulation of closed loop systems						
Module:6	Process Modeling of Distributed Systems	4 hours				

Analysis of compressible flow; heat exchanger; plug flow reactor; solution of ODE boundary value problems–Sedimentation–Heat conduction–Diffusion; classification and solution of partial differential equations			
Module:7	Process modeling of distributed systems-II		3 hours
Pressure vessels–Stresses in thin and thick cylindrical shell due to internal pressure–Circumferential and longitudinal stresses– Spherical shells subjected to internal pressure			
Module:8	Contemporary issues		2 hours
Total Lecture hours			30 hours
Text Books			
1.	Varma A.K., Process Modelling and Simulation in Chemical, Biochemical and Environmental Engineering, 1 st ed., CRC Press, USA, 2017.		
2.	Bequette B.W., Process Dynamics: Modeling, Analysis and Simulation, 1 st ed., Prentice Hall Inc., USA, 2010.		
Reference Books			
1.	Luyben W.L., Process Modelling Simulation and Control, 3 rd ed., McGraw-Hill, USA, 1996.		
2.	Ramirez W., Computational Methods in Process Simulation, 2 nd ed., Butterworths Publishers, USA, 2005.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Laboratory Experiments			
1.	Develop and solve the objective function for reaction system using Algebraic equations		2 hours
2.	Develop mathematical model for two interacting tanks in series		2 hours
3.	Design the jacketed stirred tank heater		2 hours
4.	Optimization of Vande-Vusse reaction kinetics using semi-batch reactor operation		2 hours
5.	Determination of kinetic rate of non-isothermal CSTRs in series		2 hours
6.	Design and Develop the objective functions for Biochemical reactor		2 hours
7.	Analyze the mixing performance of reactant in mixing tank		2 hours
8.	Simulation of unsteady state heat conduction equation using MatLab		2 hours
9.	Solve the elliptic PDE using PDE toolbox		2 hours
10.	Solve the parabolic PDE using PDE toolbox		2 hours
Total Laboratory Hours			20 hours
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

Coursecode	CHE4005	L	T	P	J	C
Course title	FLUIDIZATION ENGINEERING	3	0	0	0	3
Pre-requisite	Nil	Syllabus version				
		1.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. Illustrate the physical and chemical concepts aspects of fluidization process 2. Describe the various fluidization regimes and their models 3. Design of various units of fluidized bed widely used in industrial practice 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Describe the behavior of fluidization under various operating conditions 2. Elucidate the various industrial applications of fluidization 3. Determine minimum fluidization velocity and terminal velocity 4. Design suitable gas distributor for fluidized beds 5. Apply various models for designing the fluidized bed systems 6. Analyze the performance of various fluidized bed systems 						
Module:1	Introduction to Fluidization	7 hours				
Concept of Fluidization-Special Features of Fluidization-Comparison with other Contacting Methods-Advantages and Disadvantages of Fluidized Beds-Industrial Applications of Fluidized Beds - Historical Highlights- Physical Operation-Chemical Operations.						
Module:2	Characterization of Fluidization I	6 hours				
Gross Behavior of Fluidized Beds- Minimum and Terminal Velocities in Fluidized Beds						
Module:3	Characterization of Fluidization II	6 hours				
Geldart Classification of Particles- Mapping of Fluidization Regions- Design of Distributors- Power Consumption						
Module:4	Bubble Mechanics in Fluidized Beds	7 hours				
Bubbles in Dense Beds-Single Rising Bubble-Coalescence and Splitting of Bubbles- Bubble Formation above a Distributor. Bubbling Fluidized Beds- Experimental Findings- Estimation of Bed Properties-Bubbling Bed Model						
Module:5	Entrainment and Elutriation	6 hours				
Free Board Behavior- Entrainment from Tall and Short Vessels. Constant Approach. Flow Pattern of Gases through Fluidized Beds-Solid Movement -Mixing, Segregation and Staging						
Module:6	Heat Transfer in Fluidized Beds	5 hours				
Heat Transfer between Fluid and Solid- Determination and Interpretation of Heat Transfer. Heat Transfer between Fluidized Beds and Surface-						
Module:7	Miscellaneous systems	6 hours				
Conical fluidized bed-Inverse fluidized bed-Draft tube systems; Semi fluidized bed systems, Annular systems and typical applications						

Module:8	Contemporaryissues			2 hours
TotalLecturehours				45 hours
Text Books				
1.	Kunii D.,LevenspielO.,FluidizationEngineering,2 nd ed.,ButterworthHeinemann,UK, 2013.			
2.	YangW.C.,HandbookofFluidizationandFluid–ParticleSystem,1 st ed.,CRCPress, USA, 2003.			
Reference Books				
1.	GraceJ.R.,AvidanA.A.,KnowltonT.M.,CirculatingFluidizedBeds,1 st ed.,Springer,USA, 2011.			
2.	L.G.Gibilaro,FluidizationDynamics,1 st ed.,Butterworth Heinemann, UK, 2001.			
Mode ofevaluation: Continuous AssessmentTest, Quizzes, Assignments,FinalAssessmentTest				
RecommendedbyBoardof Studies		15-04-2019		
Approved by AcademicCouncil		55 th	Date	13-06-2019

Coursecode	CHE4006				L	T	P	J	C
Course title	Introduction to Molecular Dynamics and Simulation				3	0	0	0	3
Pre-requisite	CHE1003, CHE3001				Syllabus version				
									1.0
Course Objectives:									
1. Introduce molecular simulation techniques used in soft matter in atomic time and length scales									
2. Understand the basics of Molecular Dynamics simulation									
3. Demonstrate the predictive capabilities of these methods by considering a set of case-studies									
Course Outcomes (CO):									
1. Choose appropriate potentials for a system of interest									
2. Compare various ensembles and demonstrate importance of thermodynamic properties									
3. Identify non-bonded and bonded interaction and experiment with basic of MD concepts									
4. Summarize the MD algorithm and contrast different integration schemes									
5. Survey of Gromacs terminologies									
6. Estimate the dynamic properties of proteins, lipids and surfactants									
Module:1	Model potentials				5 hours				
Electronic, atomic, molecular soft matter examples; Interaction potentials-Reduced units									
Module:2	Statistical Mechanics				7 hours				
Statistical ensembles; Thermodynamic averages fluctuations; Structural quantities; Time correlation functions and transport coefficients									
Module:3	Basics of Molecular dynamics simulations				8 hours				
Non-Bonded Interactions; Bonded interaction; Force Fields; Periodic Box and Minimum Image Convention; Long Range Forces									
Module:4	Molecular dynamics strategy				5 hours				
Integrating algorithms for velocity and acceleration updates; Different types of thermostats and barostats									
Module:5	Steps involved in GROMACS				6 hours				
Energy Minimization; Solvent and counterions addition; Equilibration of temperature and pre									
Module:6	Overview of GROMACS Files				4 hours				
Input structure files visualization by VMD- Input files required for MD Simulation- Postprocessing of output files									
Module:7	Case Study using MD Simulation				8 hours				
MD Simulation biologically important proteins; Lipids and Surfactants									
Module:8	Contemporary issues				2 hours				

Coursecode	CHE4007	L	T	P	J	C
Coursetitle	RHEOLOGY OF COMPLEX FLUIDS	3	0	0	0	3
Pre-requisite	NIL	Syllabus version				
		1.0				
Course Objectives:						
<ol style="list-style-type: none"> 1. Make student aware about complex fluids and structure length scales in polymeric and colloidal systems 2. Provide basic knowledge of the physics behind colloidal systems 3. Impart basic knowledge of the physics behind polymeric solutions and its rheological behavior with concentration and temperature 						
Course Outcomes (CO):						
<ol style="list-style-type: none"> 1. Distinguish among viscous, elastic and viscoelastic behavior of fluids 2. Explain the basic forces that give rise to complex fluid behavior 3. Identify non-linear viscoelastic properties of materials and their corresponding behavior 4. Measure extensional behavior of complex fluids 5. Apply rheological behavior of colloidal system for various applications 6. Apply rheological behavior of polymeric system for various applications 						
Module:1	Elastic Solid and Viscous liquid	4 hours				
Stress tensor - Principal stresses - Finite deformation tensor- Neo-Hookean solid; Velocity gradient, general viscous fluid, plastic behaviour.						
Module:2	Complex fluid and forces	5 hours				
Complex fluids – examples, pertinent length scales, common features & applications; Forces – basic forces that drive the dynamics and behavior – steric, van der Waals, electrostatic etc.						
Module:3	Linear Viscoelasticity	6 hours				
Introduction, models - Kelvin, Maxwell; Linear viscoelasticity in three dimensions - differential form; Stress relaxation, creep, oscillation.						
Module:4	Nonlinear Viscoelasticity	7 hours				
Nonlinear phenomenon, normal stress, shear thinning, extensional thickening; Second order fluid - Upper-Convected Maxwell Equation, Lodge Integral Equation, Integral Constitutive Equations.						
Module:5	Extensional Viscosity	7 hours				
Introduction - Importance, theory; Experimental methods - Homogeneous stretching method, Constant stress devices; Spinning, Lubricated flows, Contraction flows,						
Module:6	Suspension Rheology	7 hours				

Introduction, viscosity of suspension of solid particles in Newtonian fluids, colloidal contribution to viscosity, viscoelastic properties of suspension.			
Module:7	Rheology of Polymeric Liquids		7 hours
Introduction, polymer chain conformation, zero shear viscosity, rheology of dilute polymer solution; Concentrated Solutions and Melts - Temperature Dependence.			
Module:8	Contemporary issues		2 hours
Total Lecture hours			45 hours
Text Books			
1.	Despande A.P., Krishnan J.M., Sunil Kumar P.B., Rheology of Complex Fluids, 1 st ed., Springer-Verlag, USA, 2010.		
2.	Macosko C.W., Rheology: Principles, Measurements and Application, 9 th ed., Wiley-VCH Publications, USA, 2015.		
Reference Books			
1.	Barnes H.A., Hutton J.F., Walters K., An Introduction to Rheology, 17 th ed., Elsevier, UK, 2011.		
2.	Larson R.G., The Structure and Rheology of Complex Fluids, 1 st ed., Oxford University Press, UK, 1999.		
Mode of evaluation: Continuous Assessment Test, Quizzes, Assignments, Final Assessment Test			
Recommended by Board of Studies		15-04-2019	
Approved by Academic Council		55 th	Date 13-06-2019

CHY1004	Materials & Instrumental Techniques	L	T	P	J	C
		3	0	2	0	4
Pre-requisite	Chemistry of 12th standard or equivalent	Syllabus version				
		v. 2.0				
Course Objectives:						
<input type="checkbox"/> To understand the chemistry of engineering materials and the correlation between structure and properties <input type="checkbox"/> To improve analytical capability of students by using instrumental analytical techniques						
Course Outcomes:						
<ol style="list-style-type: none"> 1. Interpret structure, hardening mechanisms, phase behaviour and properties of selected alloys. 2. Identify and formulate composite materials and lubricants. 3. Develop methods to synthesize nanomaterials. 4. Illustrate structures of carbon nanomaterials and apply them in devices. 5. Classify and describe semiconductor materials and solar energy conversion methods. 6. Determine metals like iron, sodium and potassium using colorimetry and flame emission photometry. 7. Analyse samples employing powder X-ray diffraction, SEM, EDX and TEM techniques. 8. Apply the theoretical aspects: (a) in assessing the water quality; (b) understanding the construction and working of electrochemical cells; (c) analysing metals, alloys and soil using instrumental techniques; (d) evaluating the properties of materials 						
Module:1	Metals and Alloys	4 hours				
Powder metallurgy-metallic structures and properties – phase behaviour of iron-carbon alloys-hardening mechanisms of steel-shape memory alloys						
Module:2	Composite Materials and Lubricants	6 hours				
Composites-types of composites-polymer matrix composites, metal matrix composites, ceramic matrix composites; applications of composites in automobiles and aerospace industries. Lubricants- classification, properties and mechanism of different types of lubricants.						
Module:3	Nanomaterials-I	6 hours				
Basics of nanomaterials-unique properties of nanomaterials and their benefits; size dependency on properties of CdSe nanocrystals and silver nanoparticles; preparation of nanomaterials: top-down and bottom-up approaches- high-energy ball milling, sol-gel method, solution phase synthesis of copper nanoparticles Fullerene-preparation by laser evaporation and arc methods, properties of fullerenes and their applications						
Module:4	Nanomaterials-II	6 hours				
Carbon nanotubes and graphene-preparation of carbon nanotubes by laser evaporation, arc discharge method and CVD, properties and applications of carbon nanotubes; Graphene-preparation, properties and applications; engineering applications of nanomaterials, nanoelectromechanical systems (NEMS)						

Module:5	Semiconductor Materials and Solar Energy Conversion	11 hours
Bandgap– Fermi level;importanceof silicon– silicon wafer preparation-metallurgical silicon, electronicgradesilicon, single crystal silicon, float zonetechnique; chemical processesinvolved in preparation of complementary metal-oxide-semiconductor (CMOS)IC-photolithography, wet etching, plasma etching,ion implantation, metalation, thin film deposition;alternatives to silicon Solar energyconversionmethods-principles anddevices – photovoltaiccells, p-n junction solar cell, liquid junction solar cell, multiple junction solar cell, dye-sensitized solar cell.		
Module:6	Spectroscopic Techniques	5 hours
Interaction betweenelectromagneticradiation andmatter– absorptionandemission spectroscopy–Beer-Lambert law; spectrometric instrumentation principle, instrumentation of UV-Vis spectroscopy; colorimetric determinationofIron in steel;atomic absorption spectroscopy-principle, instrumentation and determination oflead in an environmental sample; Flame emissions photometry-principle, instrumentation and determinationofNa andK present in water		
Module:7	Diffraction and Microscopic Techniques	5 hours
powder X-raydiffraction-principle and instrumentation;XRDpattern ofrubyElectron microscopy - TEM, SEM, SEM-EDAX - principle, instrumentation and application; characterization ofmetalnanoparticles usingelectron microscopy		
Module:8	Contemporary Issues	2 hours
Lectureby IndustryExperts		
Total Lecturehours:		45 hours
TextBook(s)		
1.	BradleyD. Fahlman, “Materials Chemistry”, 2011, 2 nd Edition, Springer Publications, New York.	
2.	GaryD. Christian, Purnendu K. Dasgupta, Kevin A. Schug,“Analytical Chemistry”, 2013, 7 th Edition,John Wiley&Sons,Inc., New York.	
ReferenceBooks		
1.	DouglasA Skoog,FJames Holler, StanleyR Crouch, 2016, 7 th Edition,“Principles of Instrumental Analysis”,CengageLearning,Boston, USA.	
2.	RayF. Egerton.,“Physical Principles of ElectronMicroscopy– An introduction to TEM, SEM and AFM”, 2016, 2 nd Edition, Springer, USA,	
Mode ofEvaluation:Internal Assessment (CAT,Quizzes, Digital Assignments) &FAT		
List of Challenging Experiments (Indicative)		
1.	Preparation ofrubybycombustion method and X-raydiffraction analysis	1 hour
2.	Preparation of semiconductor ZnO nanoparticles and UV-Vis spectroscopic analysis	2 hours
3.	Analysis of copper in brass using iodometry	2 hours
4.	Quantification of sodium and potassium in oral dehydration solution by flame photometry	2 hours

5.	Estimation of sulphate ions in water by light scattering technique	2 hours
6.	Quantification of divalent iron content in steel using calorimetry	2 hours
7.	Aromatic content of given lubricating oil by measuring its aniline point	1 hour
8.	Determination of pour point and cloud point of a lubricant oil	1 hour
9.	Quality checking of lubricant by measuring its viscosity	1 hour
10.	Photodegradation kinetics of methylene blue dye by nano ZnO photocatalyst	2 hours
11.	Preparation of Cu/CuO nanoparticles and imaging using electron microscopy	1 hour
12.	Preparation of iron nanoparticles and investigating its magnetic property	1 hour
Total Laboratory Hours		18 hours
Mode of Evaluation: Viva-voce and Lab performance & FAT		
Recommended by Board of Studies	12.08.2017	
Approved by Academic Council	No. 46	Date 24.08.2017

EEE1001	Basic Electrical and Electronics Engineering	L	T	P	J	C
		2	0	2	0	3
Pre-requisite	NIL	Syllabus version				
		v. 1.0				
Course Objectives:						
1. To understand the various laws and theorems applied to solve electric circuits and networks 2. To provide the students with an overview of the most important concepts in Electrical and Electronics Engineering which is the basic need for every engineer						
Course Outcomes:						
1. Solve basic electrical circuit problems using various laws and theorems 2. Analyze AC power circuits and networks, its measurement and safety concerns 3. Classify and compare various types of electrical machines 4. Design and implement various digital circuits 5. Analyze the characteristics of semiconductor devices and comprehend the various modulation techniques in communication engineering 6. Design and conduct experiments to analyze and interpret data						
Module:1	DC circuits	5 hours				
Basic circuit elements and sources, Ohms law, Kirchhoff's laws, series and parallel connection of circuit elements, Node voltage analysis, Mesh current analysis, Thevenin's and Maximum power transfer theorem						
Module:2	AC circuits	6 hours				
Alternating voltages and currents, AC values, Single Phase RL, RC, RLC Series circuits, Power in AC circuits-Power Factor-Three Phase Systems- Star and Delta Connection-Three Phase Power Measurement -Electrical Safety-Fuses and Earthing, Residential wiring						
Module:3	Electrical Machines	7 hours				
Construction, Working Principle and applications of DC Machines, Transformers, Single phase and Three-phase Induction motors, Special Machines-Stepper motor, Servo Motor and BLDC motor						
Module:4	Digital Systems	5 hours				
Basic logic circuit concepts, Representation of Numerical Data in Binary Form-Combinational logic circuits, Synthesis of logic circuits						
Module:5	Semiconductor devices and Circuits	7 hours				
Conduction in Semiconductor materials, PN junction diodes, Zener diodes, BJTs, MOSFETs, Rectifiers, Feedback Amplifiers using transistors. Communication Engineering: Modulation and Demodulation -Amplitude and Frequency Modulation						
		Total Lecture hours:	30 hours			
Text Book(s)						
1.	1. John Bird, 'Electrical circuit theory and technology', Newnes publications, 4th Edition, 2010.					
Reference Books						
1.	Allan R. Hambley, 'Electrical Engineering-Principles & Applications' Pearson Education, First Impression, 6/e, 2013					

2.	Simon Haykin, 'Communication Systems', John Wiley & Sons, 5 th Edition, 2009.		
3.	Charles K Alexander, Mathew N O Sadiku, 'Fundamentals of Electric Circuits', Tata McGraw Hill, 2012.		
4.	Batarseh, 'Power Electronics Circuits', Wiley, 2003		
5.	H. Hayt, J.E. Kemmerly and S. M. Durbin, 'Engineering Circuit Analysis', 6/e, Tata McGraw Hill, New Delhi, 2011.		
7.	Fitzgerald, Higgabogan, Grabel, 'Basic Electrical Engineering', 5 th edn, McGraw Hill, 2009.		
8.	S.L.Uppal, 'Electrical Wiring Estimating and Costing', Khanna publishers, New Delhi, 2008.		
Mode of Evaluation: CAT / Assignment / Quiz / FAT / Project / Seminar			
List of Challenging Experiments (Indicative)			
1.	Thevenin's and Maximum Power Transfer Theorems – Impedance matching of source and load	2 hours	
2.	Sinusoidal steady state Response of RLC circuits	2 hours	
3.	Three phase power measurement for ac loads	2 hours	
4.	Staircase wiring circuit layout for multi storey building	2 hours	
5.	Fabricate and test a PCB layout for a rectifier circuit	2 hours	
6.	Half and full ladder circuits.	2 hours	
7.	Full wave Rectifier circuits used in DC power supplies. Study the characteristics of the semiconductor device used	2 hours	
8.	Regulated power supply using zener diode. Study the characteristics of the Zener diode used	2 hours	
9.	Lamp dimmer circuit (Darlington pair circuit using transistors) used in cars. Study the characteristics of the transistor used	2 hours	
10.	Characteristics of MOSFET	2 hours	
Total Laboratory Hours			20 hours
Mode of assessment: Assignment / FAT			
Recommended by Board of Studies		29/05/2015	
Approved by Academic Council		37 th AC	Date 16/06/2015

Course code	RENEWABLE ENERGY SOURCES	L	T	P	J	C
MEE1011		2	2	2	0	4
Pre-requisite	NIL	Syllabus version				
		v. 2.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. To help students gain essential knowledge on the importance of various renewable energy sources 2. To familiarize the students with principles of energy conversion for various renewable energy sources 3. To do practical experiments for energy resource performance under different operating conditions 4. To understand the method for assessment of various input energy resources for meeting the specific requirements. 5. To know the limitations in renewable energy conversion techniques 						
Course Outcomes:						
<ol style="list-style-type: none"> 1. To help students gain essential knowledge on the importance of various renewable energy sources 2. To familiarize the students with principles of energy conversion for various renewable energy sources 3. To do practical experiments for energy resource performance under different operating conditions 4. To understand the method for assessment of various input energy resources for meeting the specific requirements 5. To know the limitations in renewable energy conversion techniques 						
Module:1	Classification of Energy	5 hours				
Energy chain and common forms of usable energy - Present energy scenario - World energy status - Energy scenario in India - Introduction to renewable energy resources - Introduction to Solar Energy - Energy from Sun - Spectral distribution of Solar radiation - Instruments for measurement of solar radiation - Solar radiation data analysis						
Module:2	Applications of Solar Energy	6 hours				
Thermal applications - Introduction to Solar thermal collectors - Types - Principle of operation of different collectors - Flat plate - Evacuated tube collectors - Compound parabolic collectors - Solar air heaters - Solar dryers - solar cookers - solar stills - Solar ponds - concentrating collectors - line type - point type - Methods of Solar power generation - Power towers						
Module:3	Introduction to Solar Photovoltaics	5 hours				

Physics of solar cells-Cell and module. Manufacturing Process-Characteristics of cells and module-Performance parameters-BoS-PV System applications- Stand alone-Grid connected systems.		
Module:4	Bio Energy Sources	4 hours
Energy through various processes-Energy through fermentation -Gasification-various types of gasifiers -Pyrolysis- Fixed bed and fast Pyrolysis-Bio energy through digestion-Types of Digesters-Factors affecting the yield of products.		
Module:5	Wind Energy	4 hours
resource assessment-types of wind turbines-selection of components-blade materials-power regulation-various methods of control-wind farms-site selection - off shore wind farms - Solar Wind Hybrid energy systems.		
Module:6	Small Hydro Power Systems	2 hours
Introduction -types-system components, discharge curve and estimation of power potential- Turbines for SHP.		
Module:7	Ocean Energy	2 hours
Power generation through OTEC systems-various types-Energy through waves and tides- Energy generation through geothermal systems- types.		
Module:8	Contemporary issues:	2 hours
Discussion on Recent developments in the area of renewable energy systems and their integration		
Total Lecture hours:		30 hours
Text Book(s)		
1.	John Andrews, Nick Jelley(2013), Energy Science: Principles, technologies and impacts, Oxford Universities press.	
Reference Books		
1.	Fang Lin You, Hongye (2012), Renewable Energy Systems, Advanced conversion technologies and applications, CRC Press	
2	John.A.Duffie, William A.Beckman(2013), Solar Engineering of Thermal processes, Wiley	
3	A.R.Jha (2010), Wind Turbine technology, CRC Press.	
4	Godfrey Boyle (2012), Renewable Energy, power for a sustainable future, Oxford University Press..	
Mode of Evaluation: CAT / Assignment / Quiz/ FAT / Project / Seminar		

List of Challenging Experiments (Indicative)		
1.	1. Estimation of Solar radiation: Pyranometer, pyr heliometer. 2. Testing the yield of a Solar still in outdoor conditions (Multiple sessions). 3. Wind Energy Experimental Set up– I. 4. Wind Energy Experimental Set up– II. 5. Testing of Solar PV system in PV training Kit. 6. Fuel Cell Experiment. 7. Performance of Biomass stove. 8. Production of Bio-diesel by Transesterification process. 9. Flash Point and Fire point comparison for conventional fuels and alternate fuels. 10. Production of Hydrogen from Electrolysis with PV system. 11. Estimation of Figures of Merit in a Solar cooker. 12. Performance characteristics of a Solar thermal collector. 13. Exergy analysis of a Solar cabinet dryer.	30 x 14
Total Laboratory Hours		17 hours
Mode of assessment:		
Recommended by Board of Studies	17-08-2017	
Approved by Academic Council	No. 47	Date 05-10-2017

Course code	COMPUTATIONAL FLUID DYNAMICS	L	T	P	J	C
MEE4006		2	1	2	0	4
Pre-requisite	MEE1004, MEE2005, MAT3005 (or) MEE1032, MEE1033/MEE2005, MAT3005	Syllabus version				
		v. 2.2				
Course Objectives:						
<ol style="list-style-type: none"> 1. To provide the students with sufficient background to understand the mathematical representation of the governing equations for fluid flow and heat transfer problems. 2. To equip the students to address complex fluid flow and heat transfer problems by approximating the governing differential equations with boundary conditions through Finite difference and finite volume discretization methods. 3. To enable students to understand different types of grid and its attributes and their suitability for different engineering applications 4. Develop the students to use appropriate turbulence model for solving engineering problems. 						
Course Outcomes:						
<ol style="list-style-type: none"> 1. Apply mathematics and engineering fundamentals to recognize the type of fluid flow and heat transfer that occur in a particular physical system and to use the appropriate model equations to investigate the problem. 2. Solve governing equations using finite difference discretization technique 3. Solve governing equations using finite volume method 4. Generate appropriate type of grids required for solving engineering problems accurately. 5. Apply suitable turbulence model for the chosen real world engineering problems. 6. Solve fluid flow and heat transfer problems using commercial CFD tools 						
Module:1	Introduction					1 hour
CFD overview -Applications of CFD.						
Module:2	Governing Equations of Fluid Dynamics and Heat Transfer:					6 hours
Models of Flow – Conservation and Non-conservation form-Continuity, Momentum and Energy Equation in conservation and non-conservation form (differential equations only)-Characteristics of PDE's -elliptic, parabolic and hyperbolic.						
Module:3	Discretization and Finite Difference method					7 hours
<p>Discretization: Basic aspects of Discretization– Comparison of finite difference, finite volume and finite element techniques.</p> <p>Finite Difference method: Forward, Backward and Central difference schemes, Transient one and two dimensional conduction-Explicit, implicit, semi-implicit and ADI methods -Stability analysis and error estimation.</p>						

Module:4	Grid Generation	3 hours
Grid Generation: Choice of grid, grid oriented velocity components, Cartesian velocity components, staggered and collocated arrangements.		
Module:5	Convection and Diffusion	7 hours
Convection and Diffusion: Steady one-dimensional convection and diffusion-Central difference, upwind, quick, exponential, hybrid and power law schemes-False diffusion, SIMPLE – Algorithm.		
Module:6	Turbulence Modeling	4 hours
Turbulence Modeling: Introduction – Types of Turbulence modeling– Reynolds Time Averaging– Reynolds Time Averaged conservation equations– Boussinesq approach– One equation k - ϵ model.		
Module:7	Contemporary issues	2 hours
Total Lecture hours:		30 hours
Text Book(s)		
1.	John D Anderson, Computational Fluid Dynamics– The Basics with Applications, 1st Edition, McGraw Hill, 2012.	
Reference Books		
1.	Chung T.J, Computational Fluid Dynamics, Cambridge University Press, 2014.	
2.	Muralidhar Kand Sundararajan T, Computational Fluid Flow and Heat Transfer, Narosa Publications, New Delhi, 2014.	
3.	Versteeg H.K and Malalasekara W, An Introduction to Computational Fluid Dynamics-The Finite Volume Method, 2nd Edition, Pearson, 2010.	
Mode of Evaluation: CAT / Assignment / Quiz/ FAT / Project/ Seminar		
List of Challenging Experiments (Indicative)		
1.	Modeling of simple and complex geometries.	3 hours
2.	Hexahedral meshing for simple geometries like square duct, circular pipe.	3 hours
3.	O-grid hexa meshing for circular pipe.	3 hours
4.	Tetrahedral meshing for simple geometries including fluid and solid domains.	3 hours
5.	Preprocessing in FLUENT– Case setup and analyzing for already mesh generated model.	3 hours
6.	Steady state temperature distribution in a rectangular plate (ANSYS Fluent and FDM).	3 hours

7.	Diffuserforahydropower turbine.	3 hours
8.	Flow overan airfoil-Laminar and turbulent flow.	3 hours
9.	Supersonic flow past a wedgein a channel.	3 hours
10.	Exercise (foreach student– different exercise) from FLUENTtutorial (casesetup, analyzing,and post-processing).	3 hours
Total Laboratory Hours		30 hours
Mode of assessment:		
Recommended byBoardofStudies	17-08-2017	
Approved byAcademicCouncil	47	Date 05-10-2017

SCHEME 2020